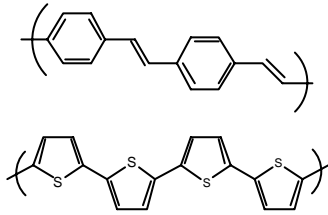


Exciton diffusion and interfacial charge separation in porphyrin/TiO₂ bilayers

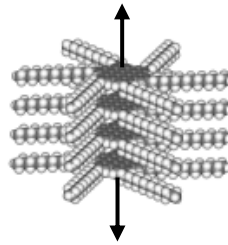
Laurens D.A. Siebbeles
Delft University of Technology
The Netherlands

October 21, 2004

Research in Delft on opto-electronic properties of materials



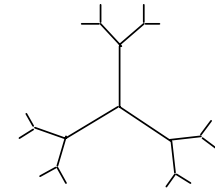
conducting polymers



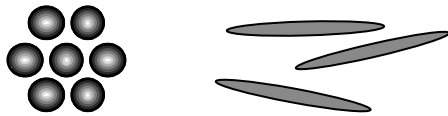
discotic liquid crystals



DNA



supra-molecular
assemblies

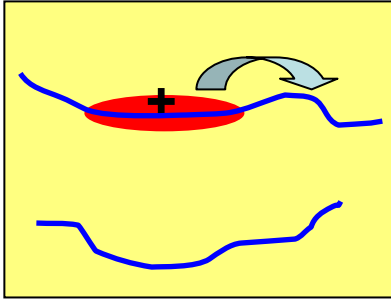


inorganic nanoparticles, nanorods

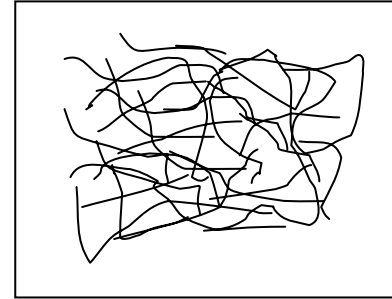
composite systems

Sample morphologies

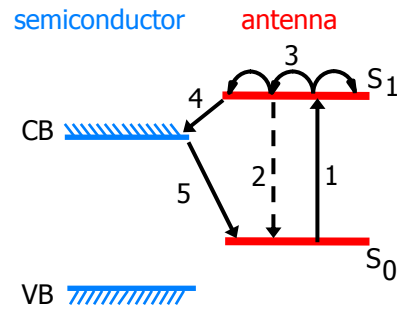
dilute solution / gel



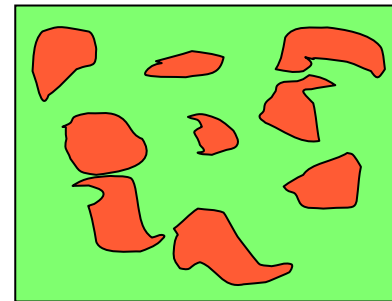
bulk solid



thin film on (active) substrate



heterogeneous blends



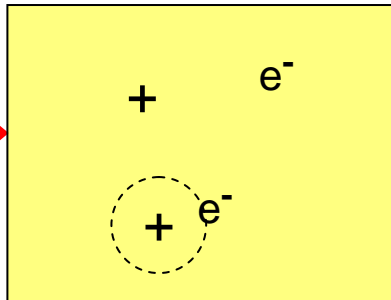
Dynamics of charge carriers and excitons

Formation (fsec - nsec)

Radiation pulse



- e⁻ accelerator
- lasers



Time-resolved
detection

Detection (microwave, THz, optical)

Charge carriers

- mobility, trapping, recombination
- optical abs. spectra

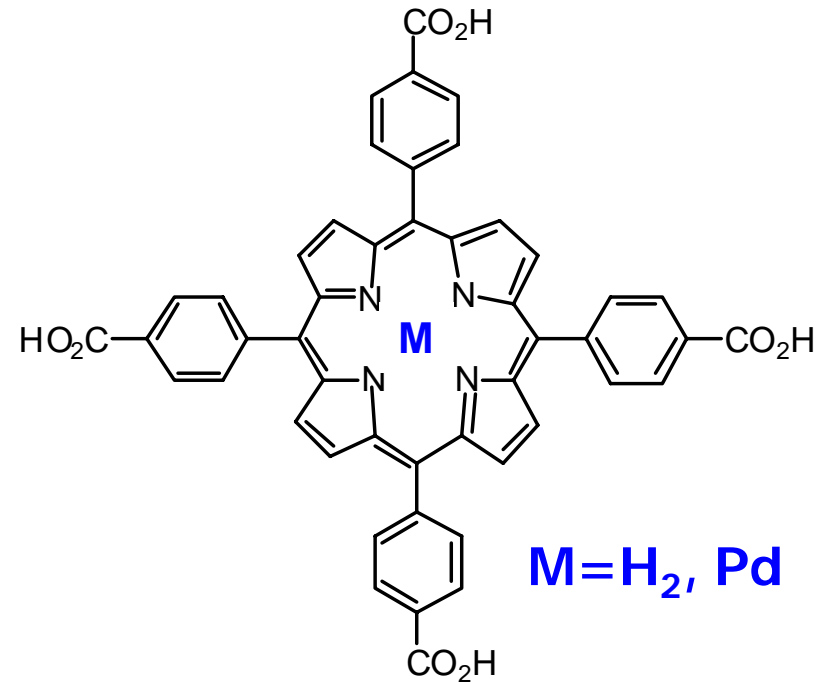
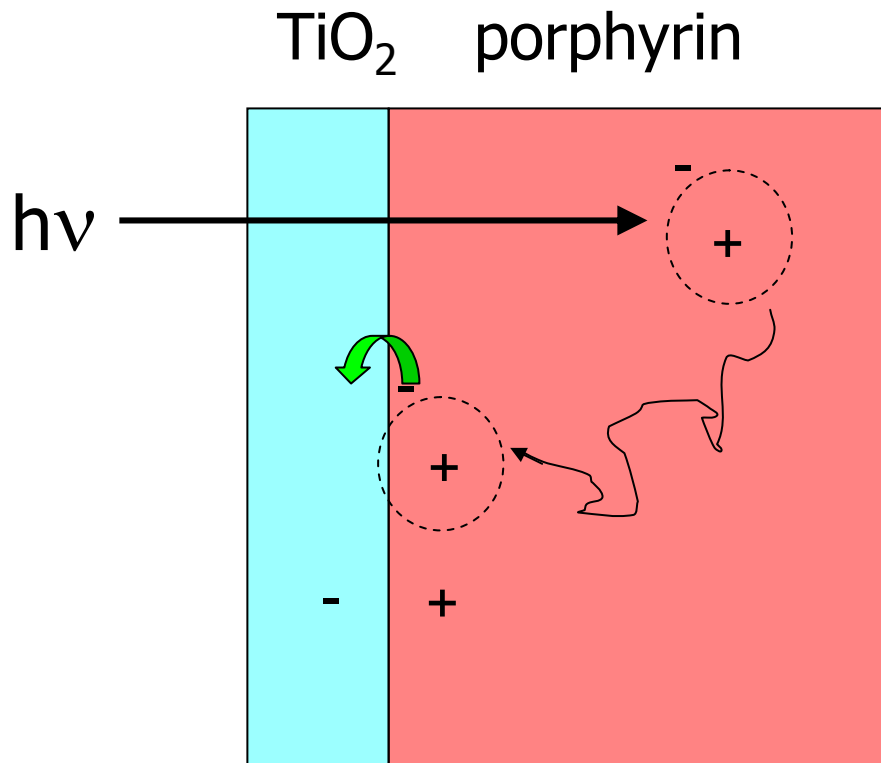
Excitons

- dissociation, diffusion and decay
- polarizability, opt. abs. spectra

Theory

- electronic structure calculations (HF, DFT etc.)
- quantum mechanical calcs. on charge and exciton motion
- Monte Carlo simulations of hopping transport

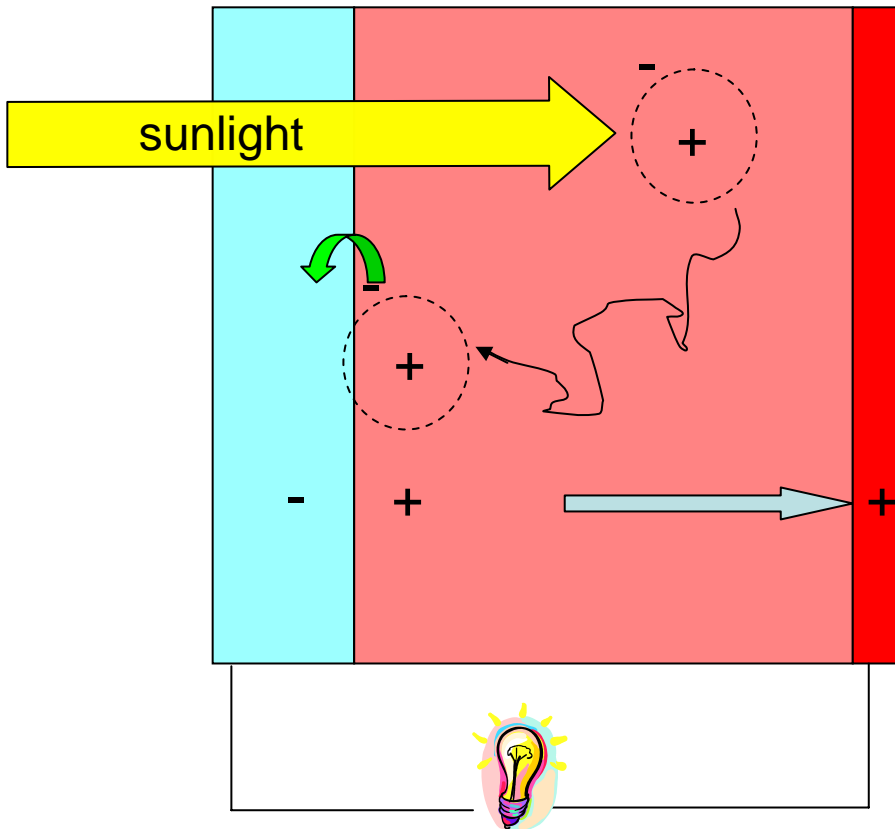
Exciton diffusion in bilayer model systems



5,10,15,20-tetrakis(4-carboxyphenyl) porphyrin (TPPC)

Simple bilayer dye-sensitized solar cell

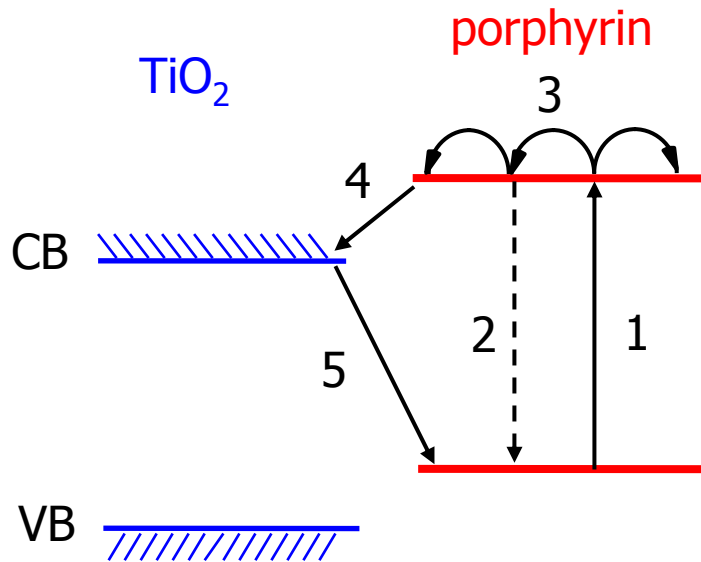
TiO₂ porphyrin



Requirements:

- efficient light absorption
- large exciton diffusion length
- efficient charge separation at interface
- escape of charge from recombination
- mobile charges

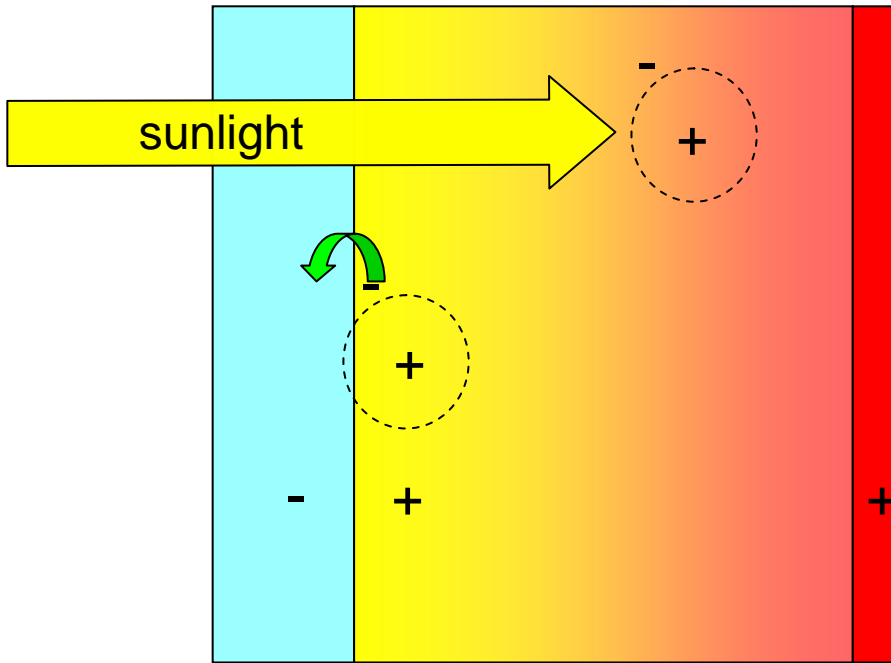
Relevant processes



- 1: photo-excitation
- 2: (non)radiative decay
- 3: exciton diffusion and annihilation
- 4: interfacial electron transfer
- 5: interfacial charge recombination

Bilayer is inefficient due to small exciton diffusion length

TiO₂ organic material

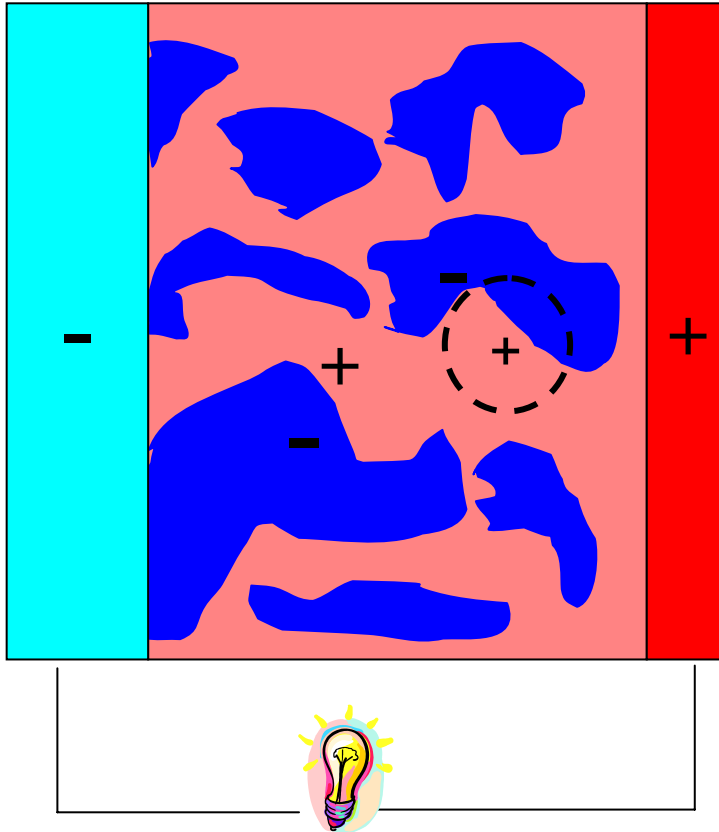


only small fraction of excitons reaches interface

exciton diffusion length $\Lambda_E = \sqrt{D_E \tau_E} < 10 \text{ nm}$

light penetration depth $\Lambda_{hv} \sim 100 \text{ nm}$

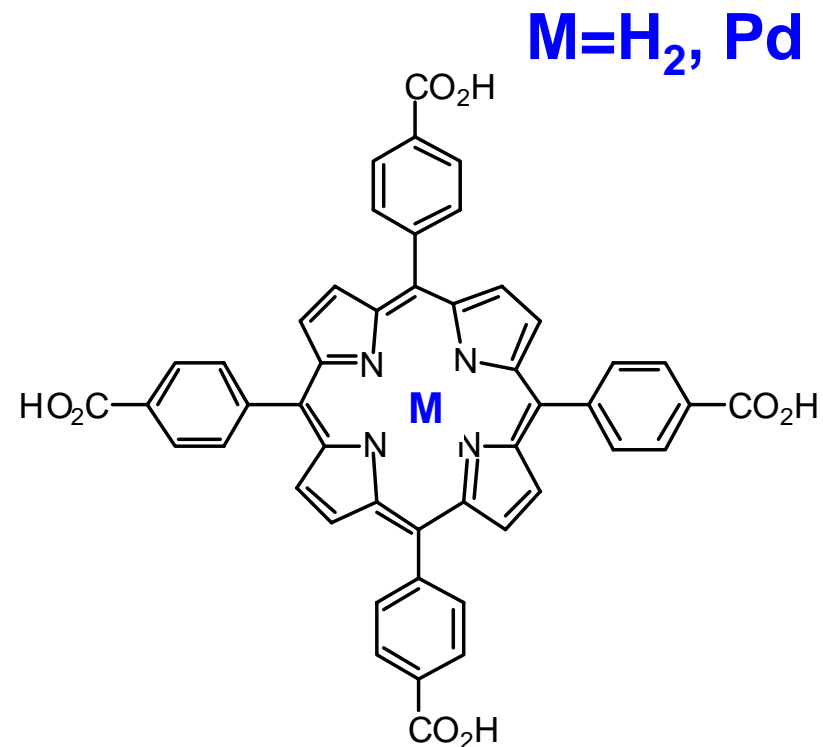
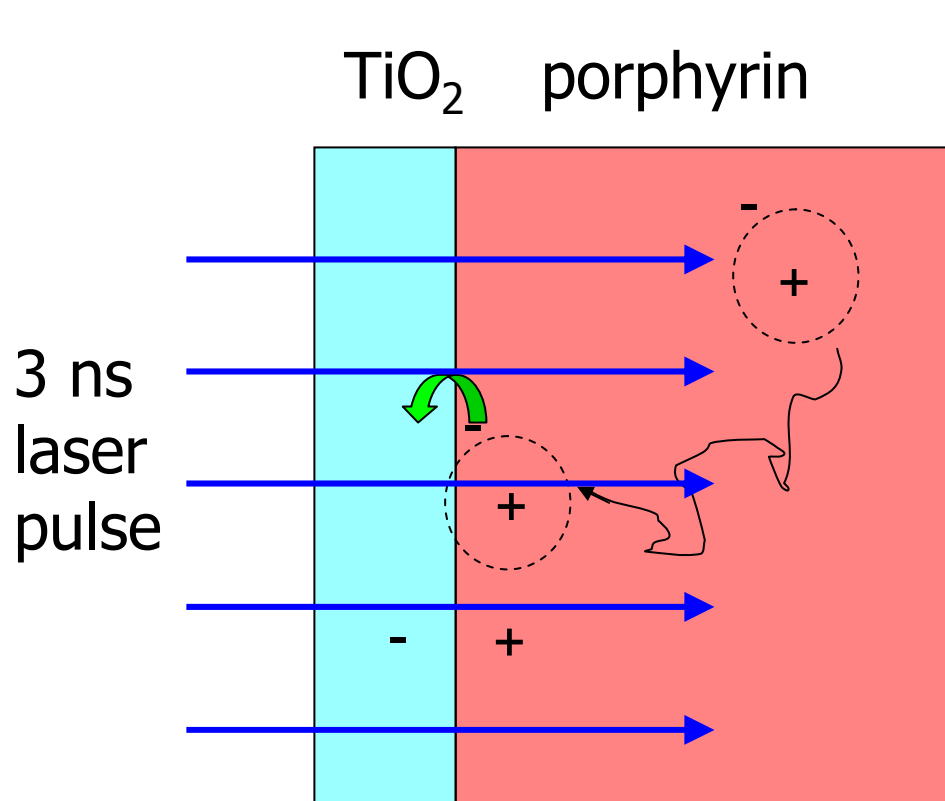
In heterogeneous structure Λ_E can be short



Disadvantages:

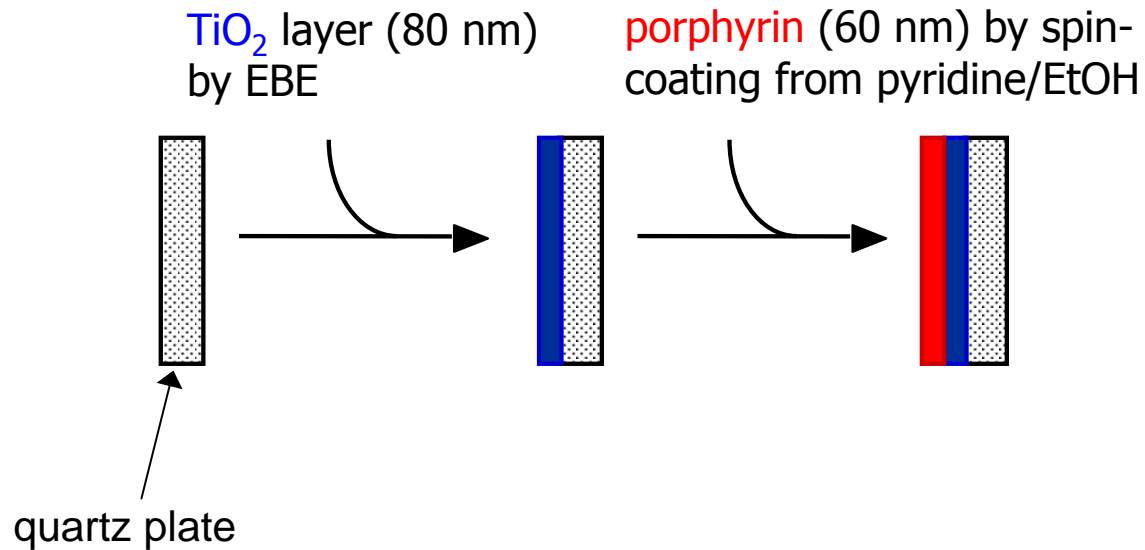
- percolation paths needed
- recombination losses
- exciton quenching on trapped charges

Exciton diffusion and charge separation in bilayer model systems: effect of Pd

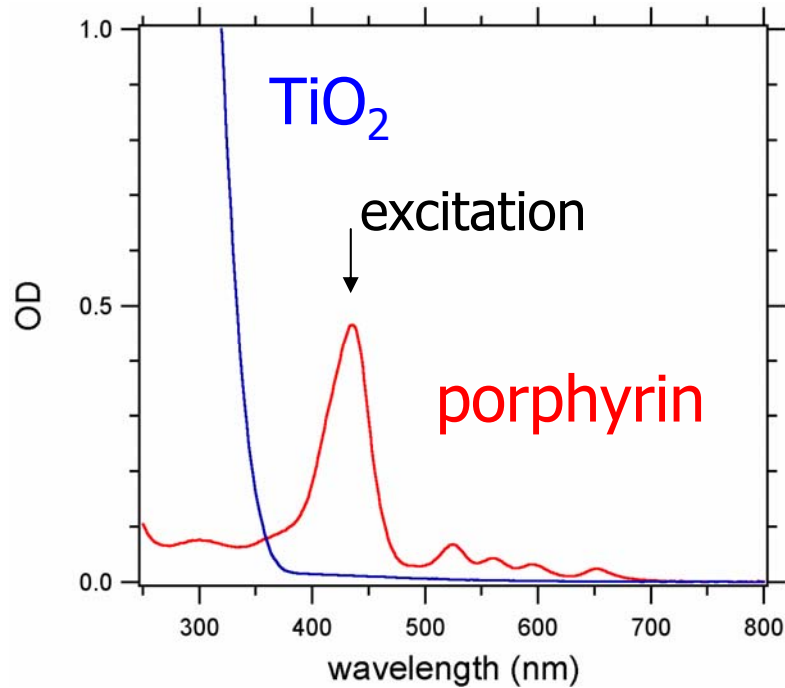


5,10,15,20-tetrakis(4-carboxyphenyl) porphyrin (TPPC)

Layer preparation

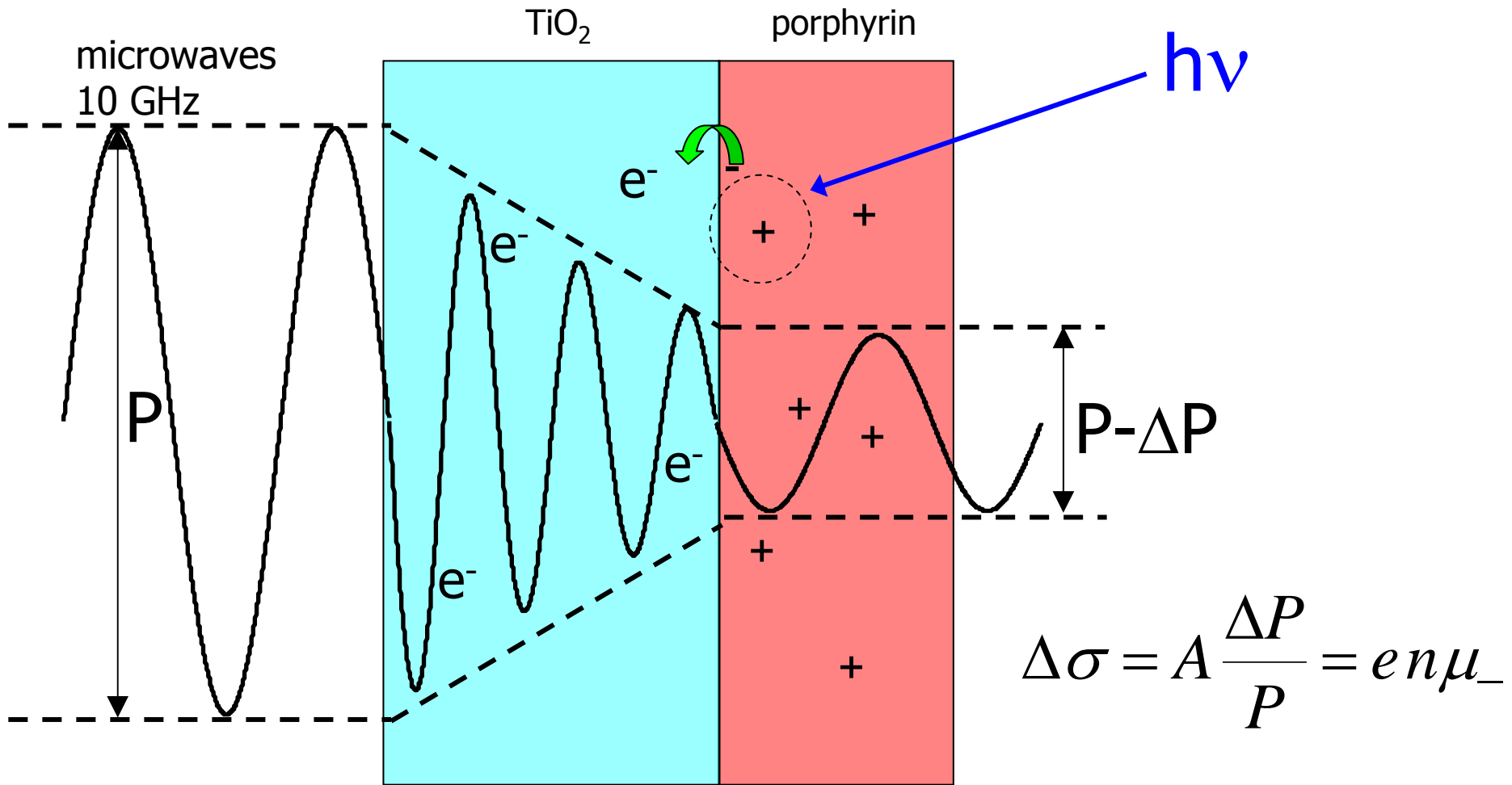


Absorption spectra

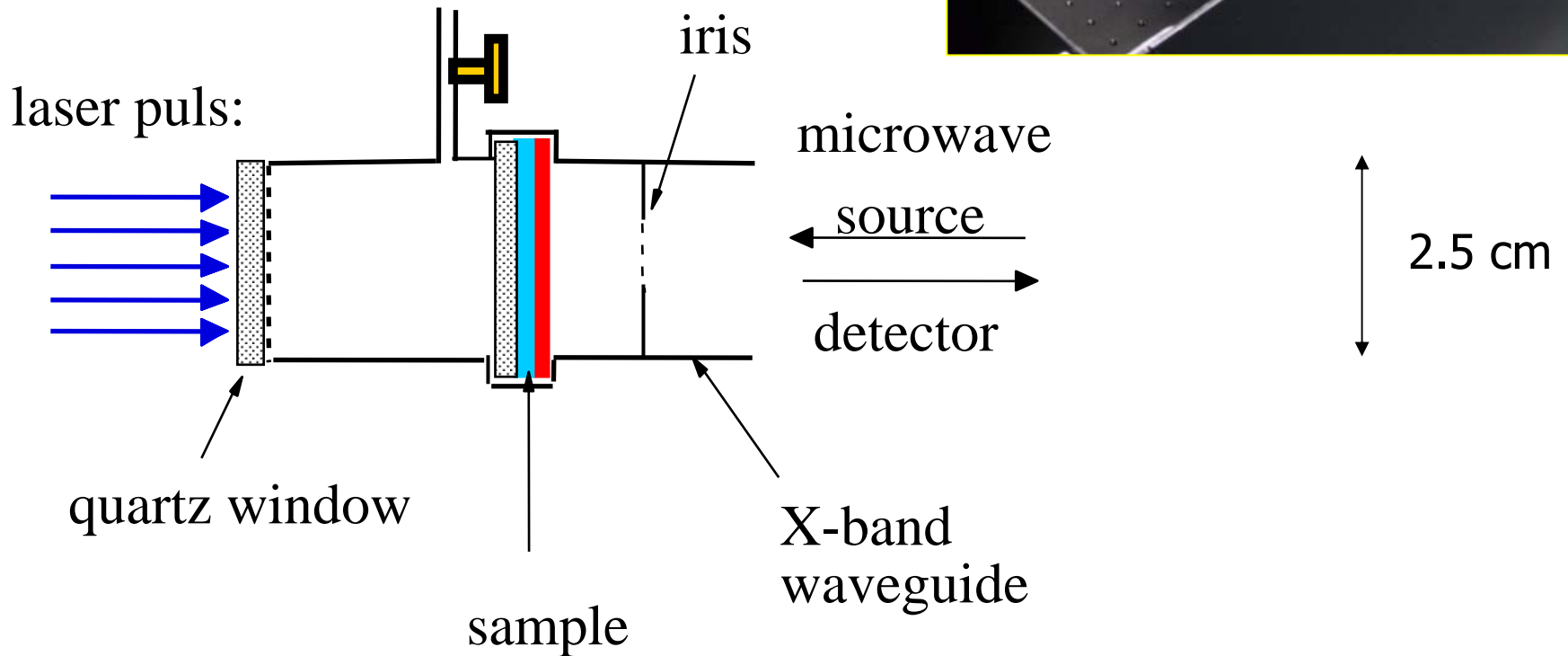
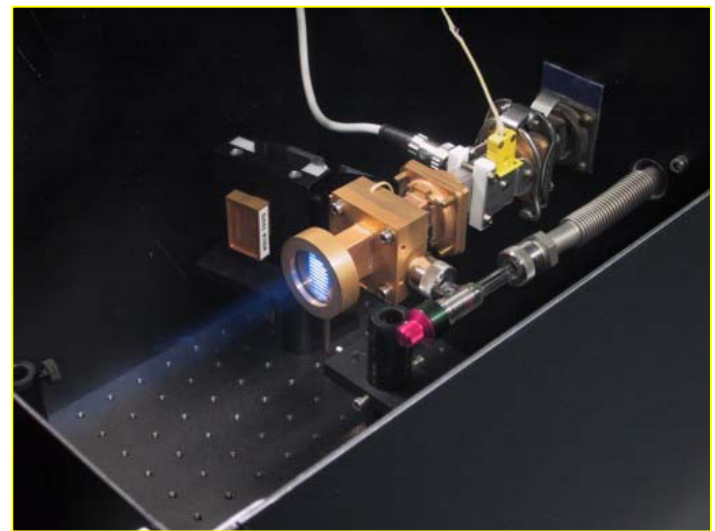


Selective excitation of porphyrin (400-700 nm) or TiO₂ (300 nm)

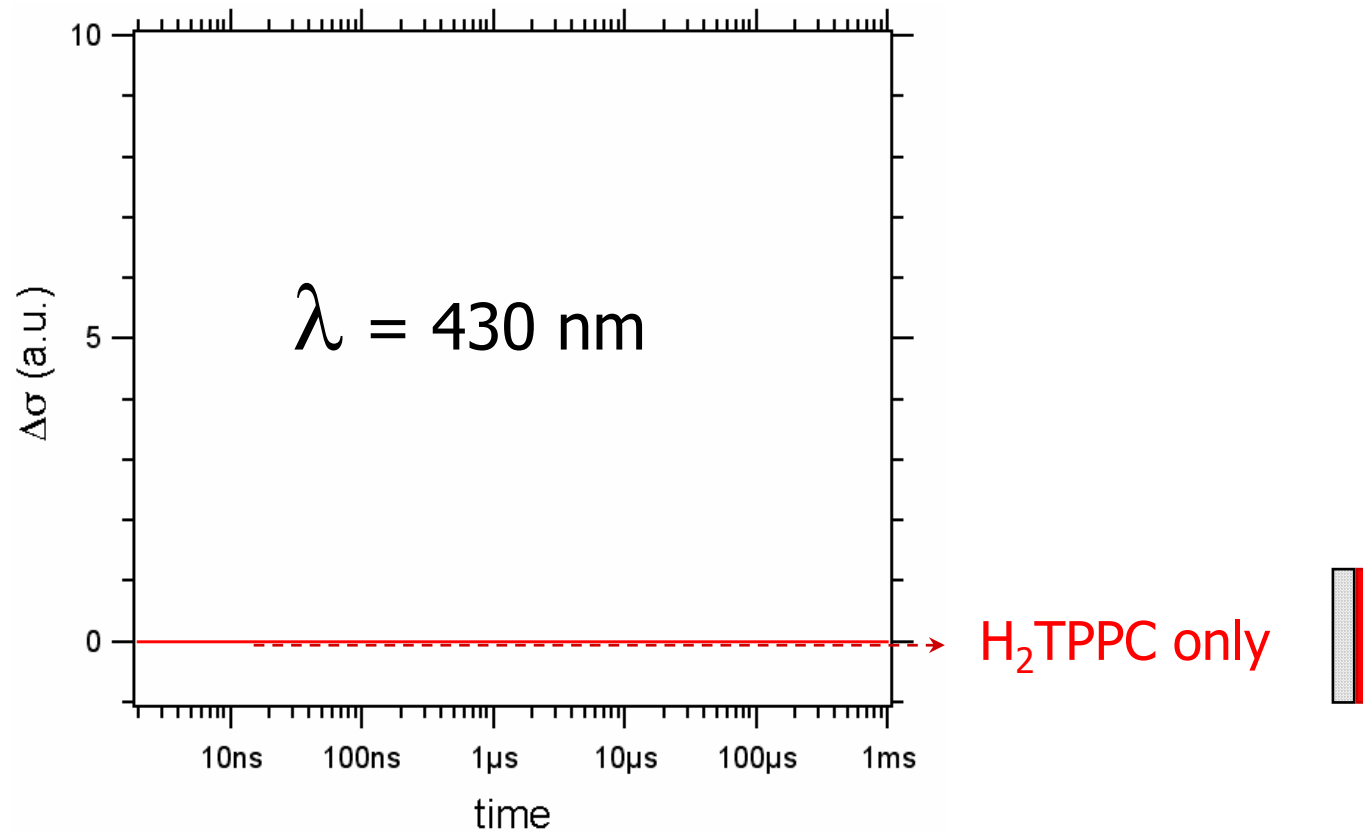
Time-resolved microwave conductivity measurements probe mobile electrons in TiO₂



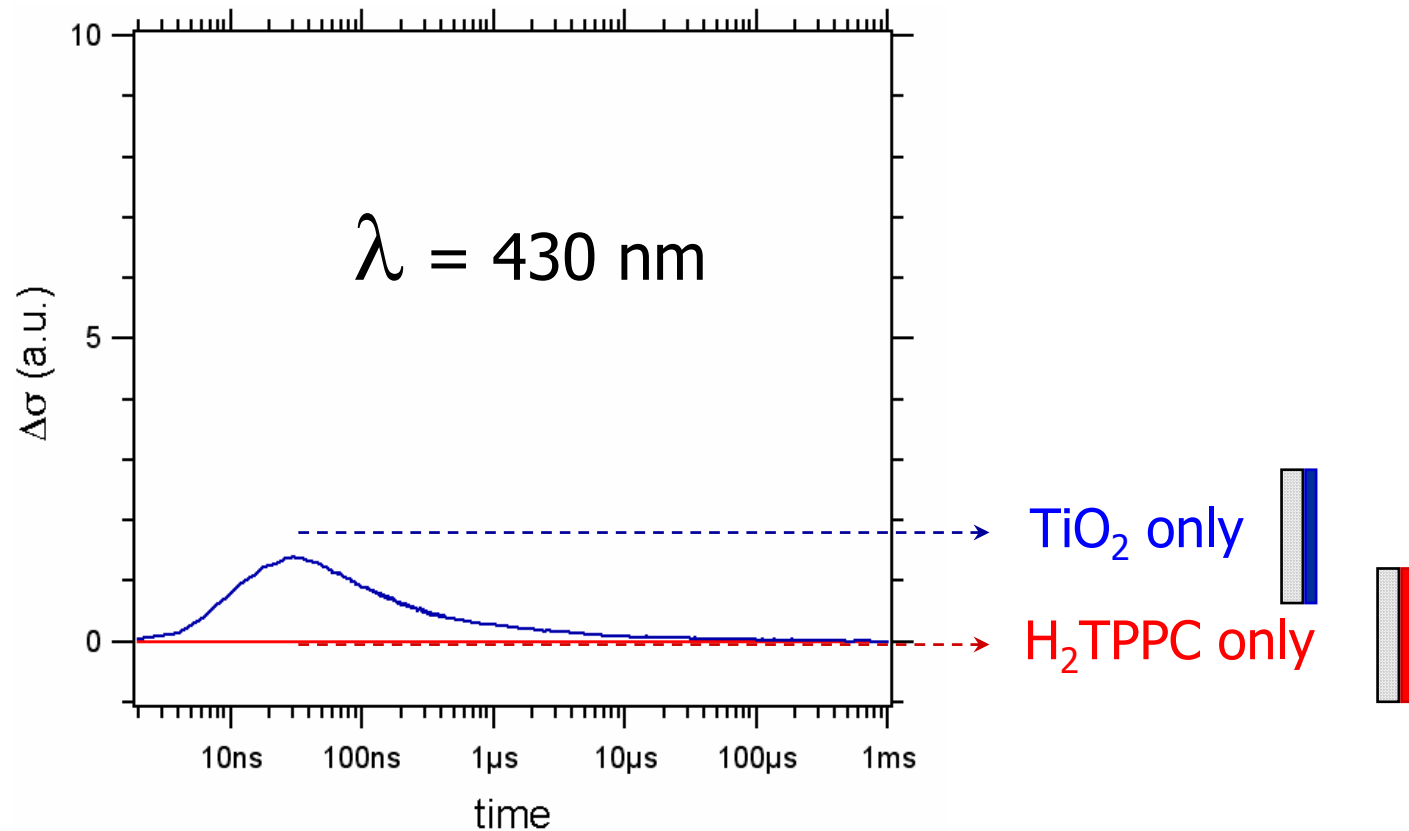
Measurement cell



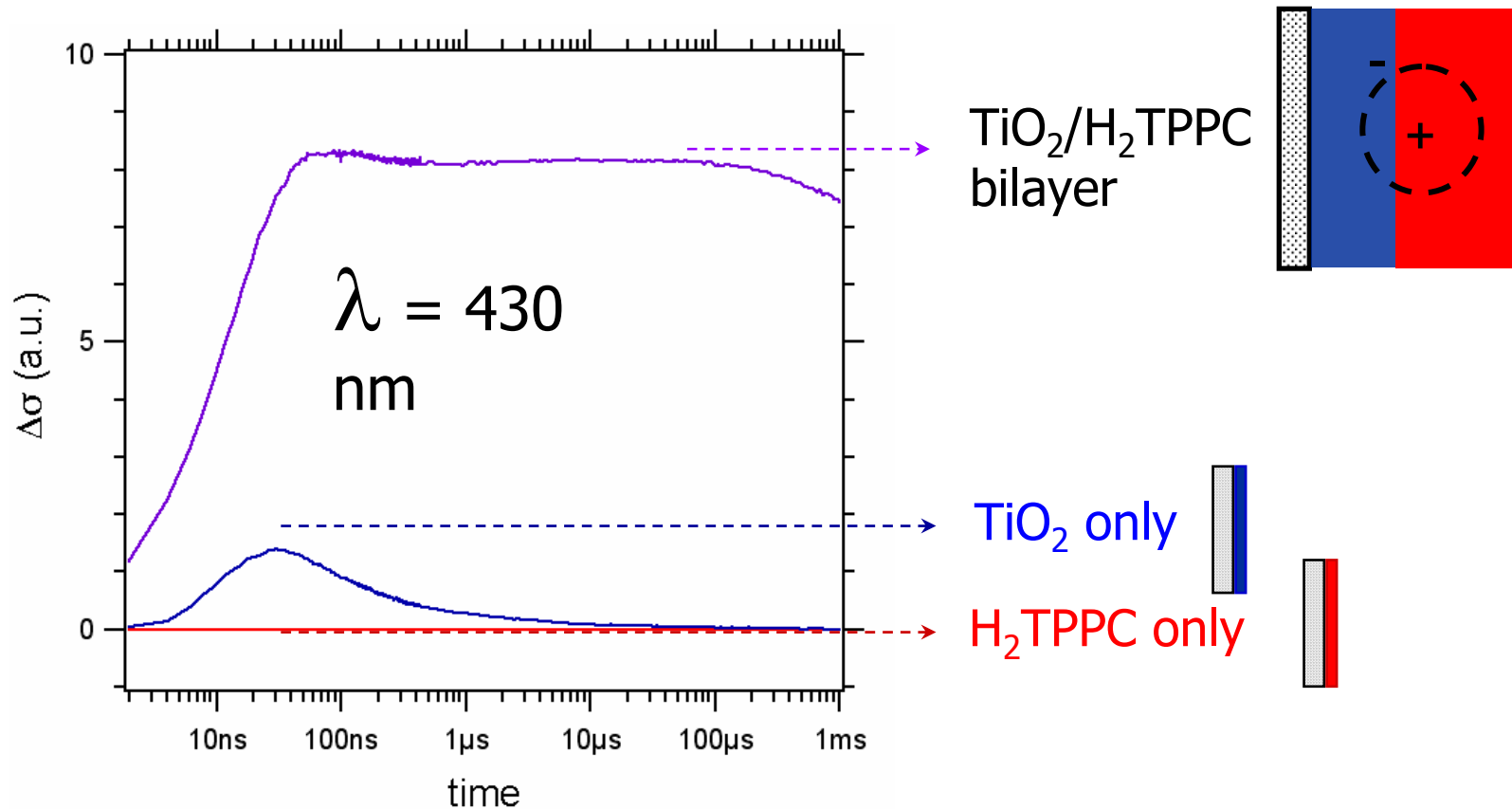
Porphyryn on quartz: no photoconductivity



TiO₂: small short lived photoconductivity

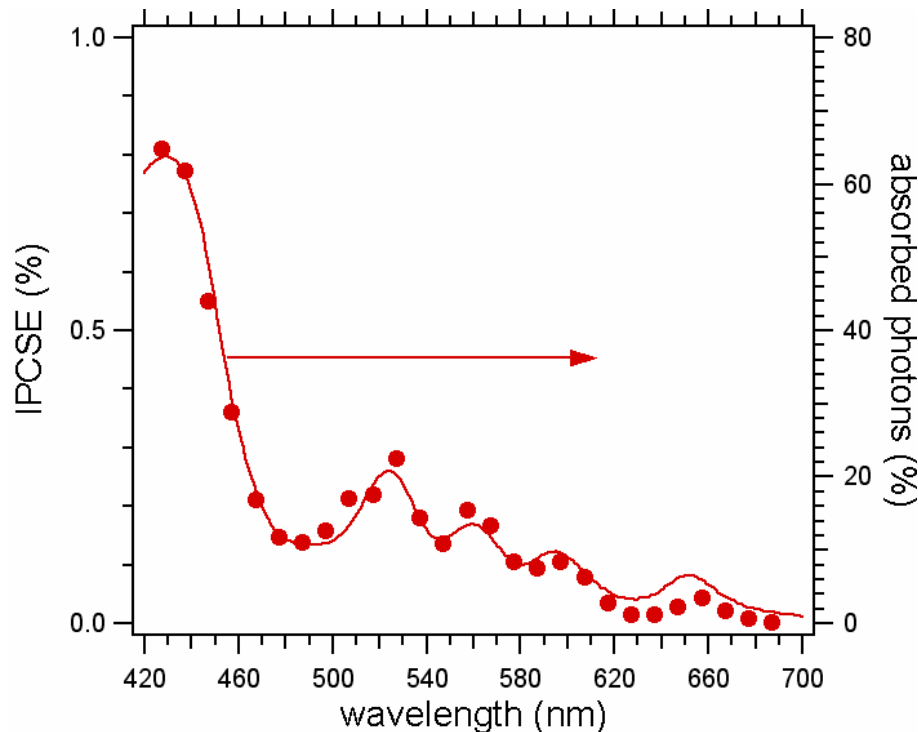


Porphyrin/TiO₂ bilayer: long-lived photoconductivity



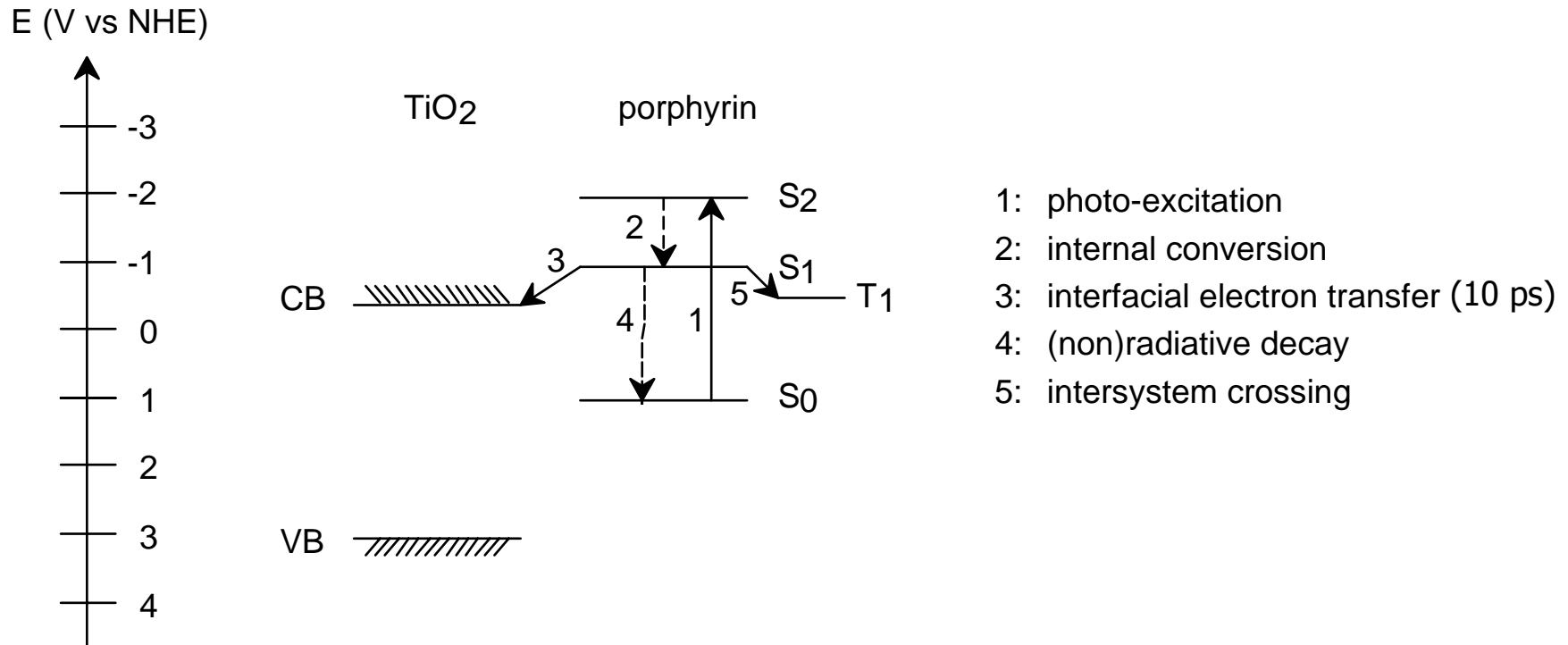
Porphyrin/TiO₂ bilayer: wavelength dependence

$$\text{IPCSE} = \frac{\text{no. of charges}}{\text{no. of incident photons}}$$



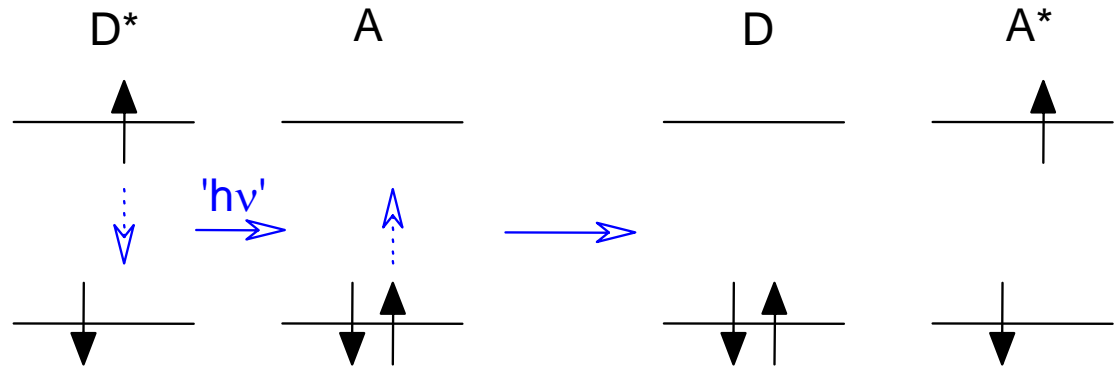
- No preferential charge separation at specific wavelength
- IPCSE < 0.8 % ($F_a=0.6$ at 430 nm)
- Exciton diffusion length only 1-2 monolayers of porphyrin

In H₂TPPC photoconductivity results from singlet excitons

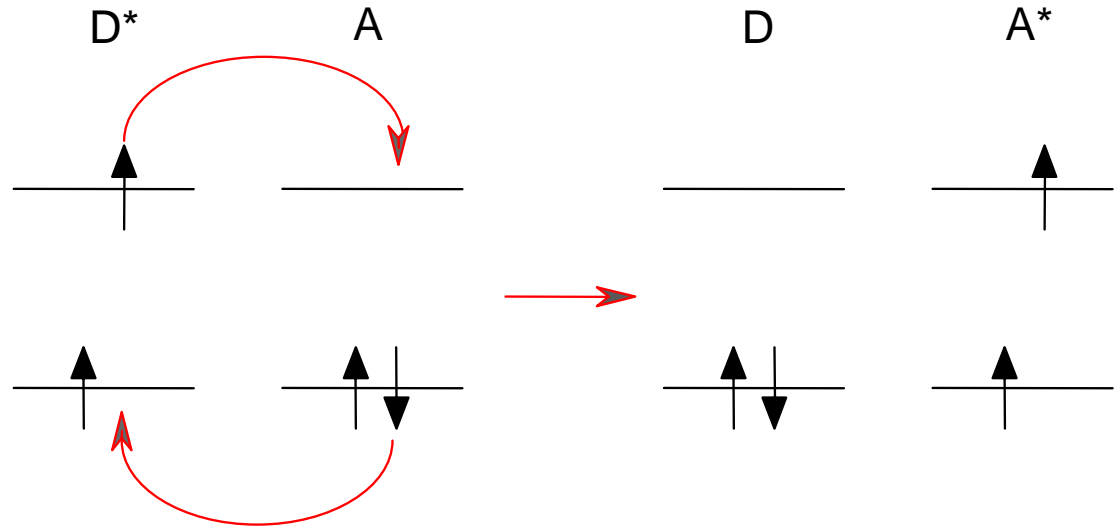


Singlets ($\tau=11$ ns) undergo intersystem crossing to triplets ($\phi=0.78$, $\tau=0.4$ ms) which diffuse via slow Dexter mechanism

Singlet:
Förster energy transfer

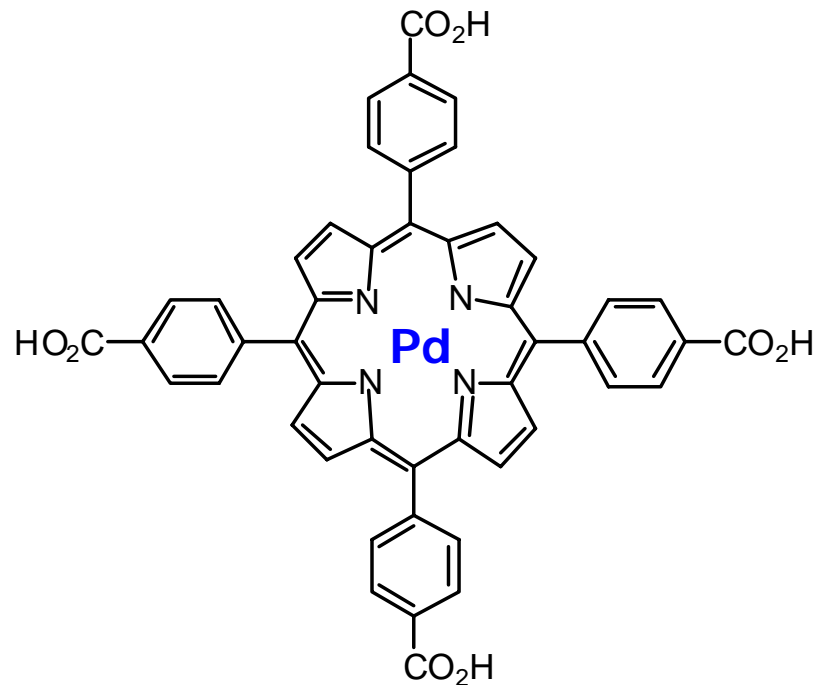


Triplet:
Dexter energy transfer

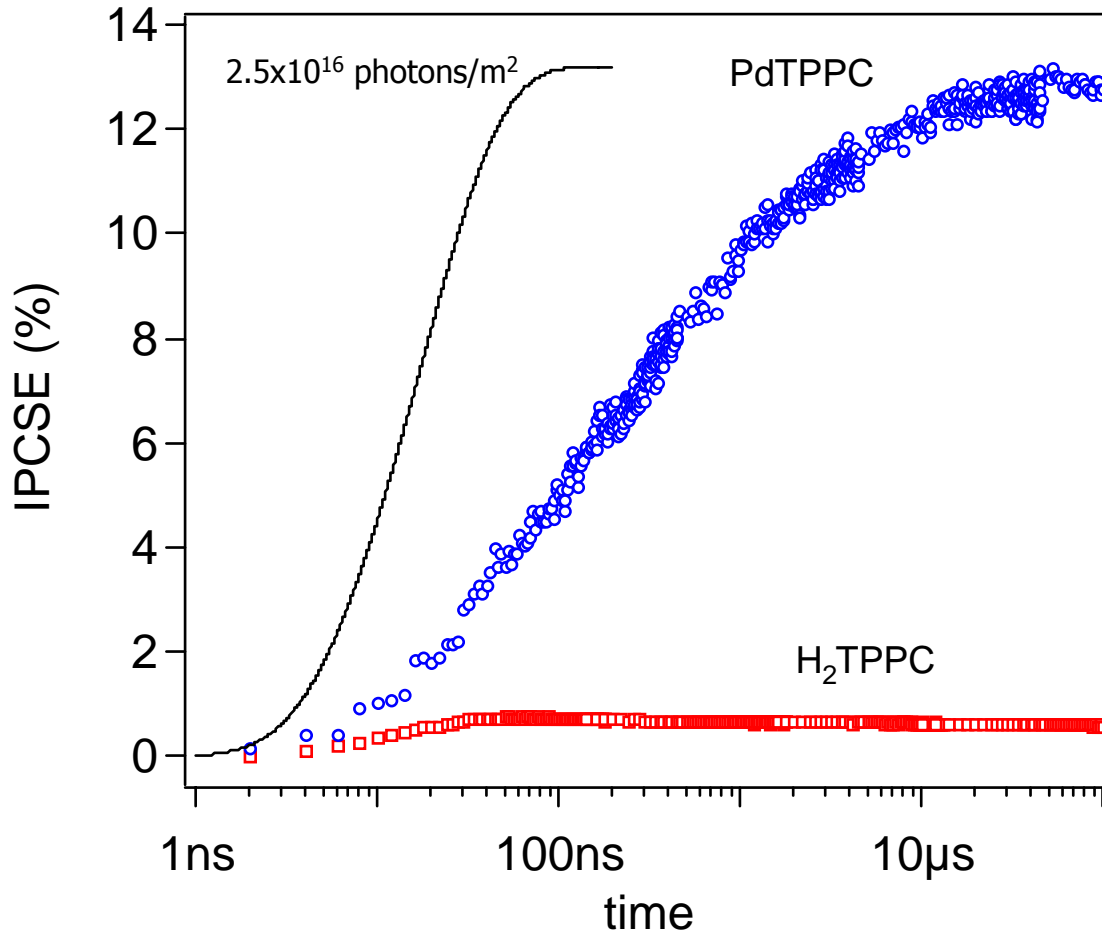


Mixing of singlet and triplet by spin orbit coupling due to heavy metal

- Excitons in Pd substituted porphyrin:
- long lifetime due to triplet character
 - diffusion via singlet character

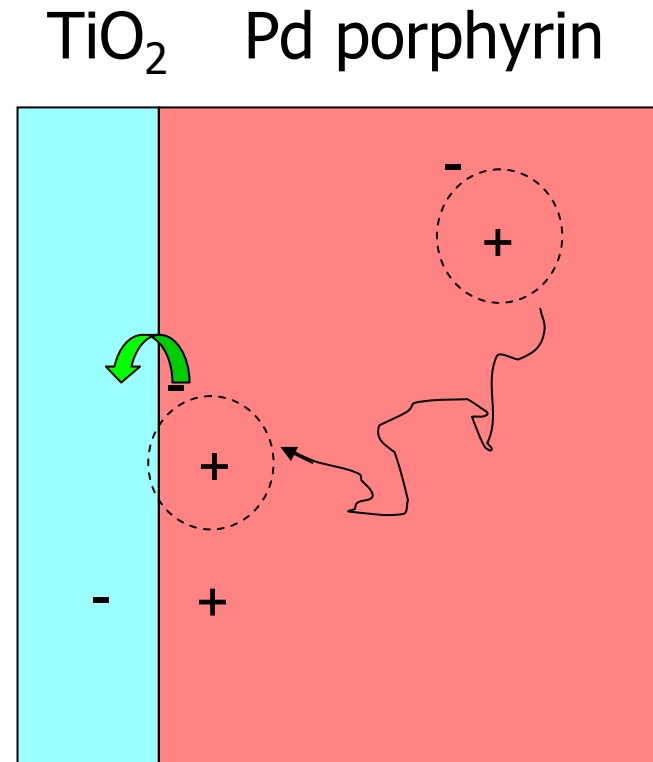
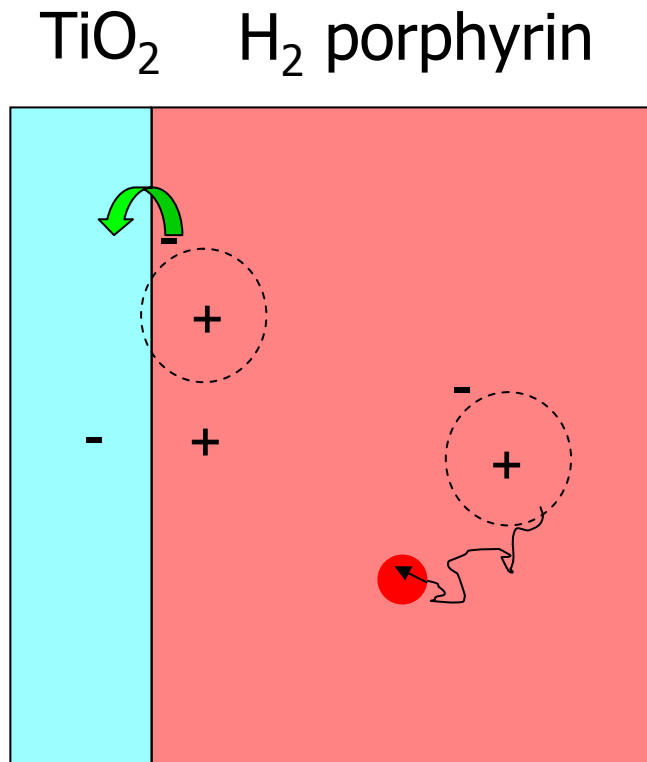


Pd substitution enhances IPCSE on longer times to 12% due to diffusion of excitons with mixed singlet/triplet character



	H ₂ TPPC	PdTPPC
τ_S (ns)	11	$\sim 10^{-2}$
τ_T (ms)	0.4	0.29
ϕ_T	0.78	~ 1
k_P (s ⁻¹)	0.0067	61
$E(S_0)$ (eV)	-5.5 [-1.1]	-5.7 [-1.3]
$E(S_1)$ (eV)	-3.6 [0.8]	-3.5 [0.9]
$E(S_2)$ (eV)	-2.6 [1.8]	-2.7 [1.7]
$E(T_1)$ (eV)	-4.1 [0.3]	-3.9 [0.5]

Enhanced diffusion length in PdTPPC



In PdTPPC diffusion via singlet and triplet

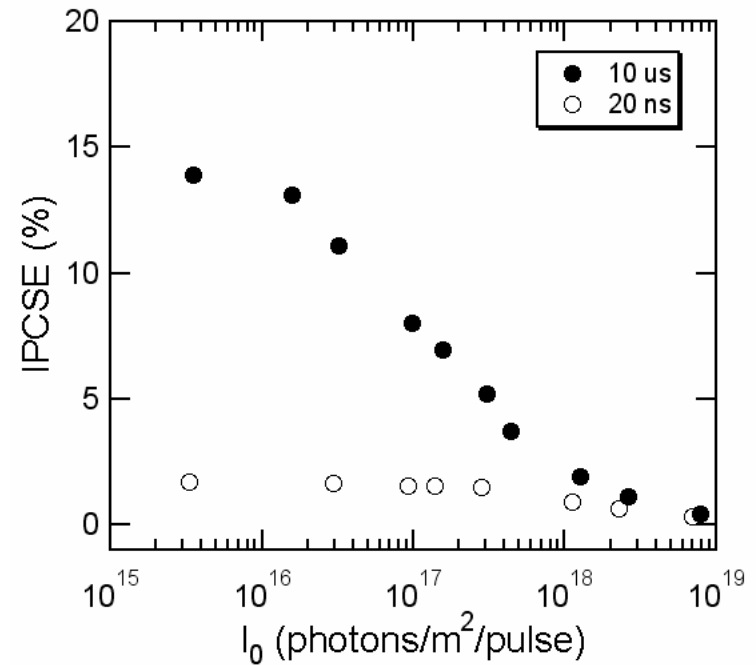
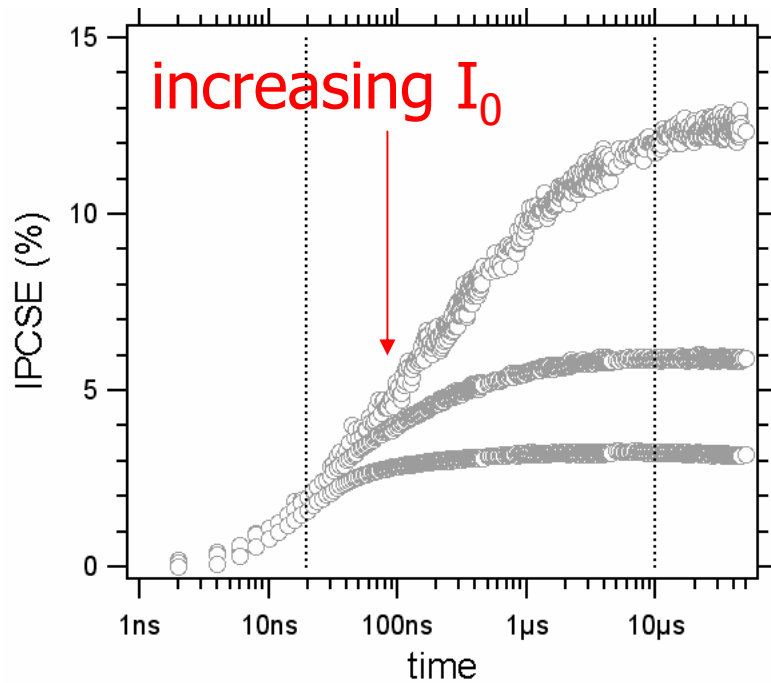
$$\Psi_{exc} = c_S \Phi_S + c_T \Phi_T$$

$$\Lambda = \sqrt{D\tau}$$

$$D = \delta^2 k_{hop} = \delta^2 \left[P_s^2 k_{Forster} + (1 - P_s) k_{Dexter} \right] \quad \tau = \frac{1}{P_s k_{rad} + k_{nrad}}$$

$$\Lambda = \sqrt{\frac{\delta^2 \left[P_s^2 k_{Forster} + (1 - P_s) k_{Dexter} \right]}{P_s k_{rad} + k_{nrad}}}$$

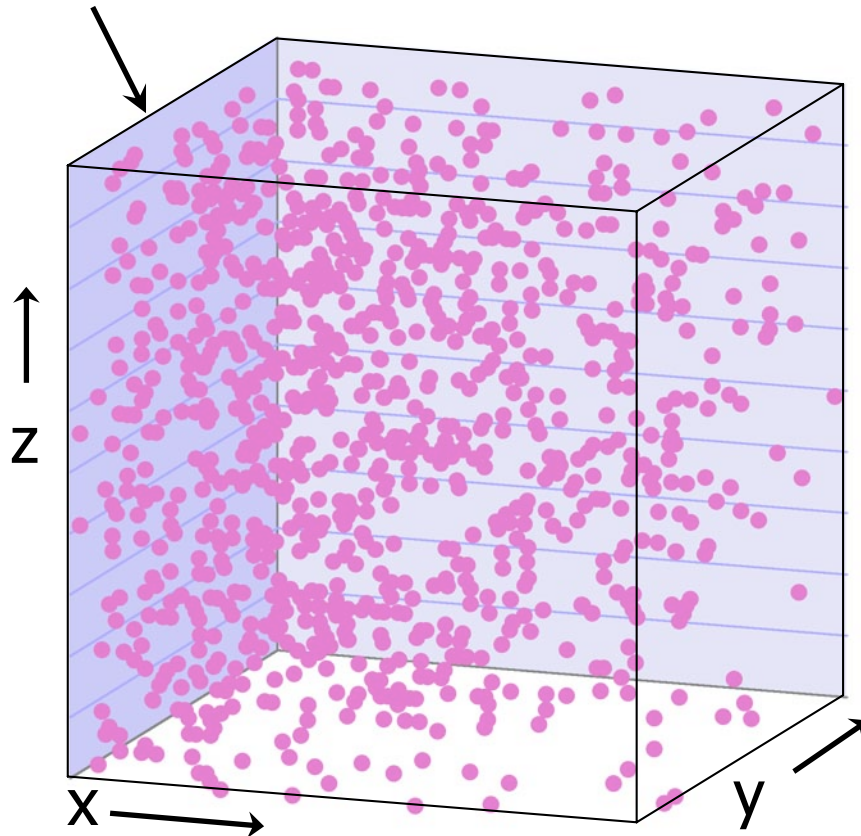
Effect of laser intensity on conductivity in PdTPPC/TiO₂ bilayer



Strong influence of $I_0 \rightarrow$ bimolecular exciton-exciton annihilation

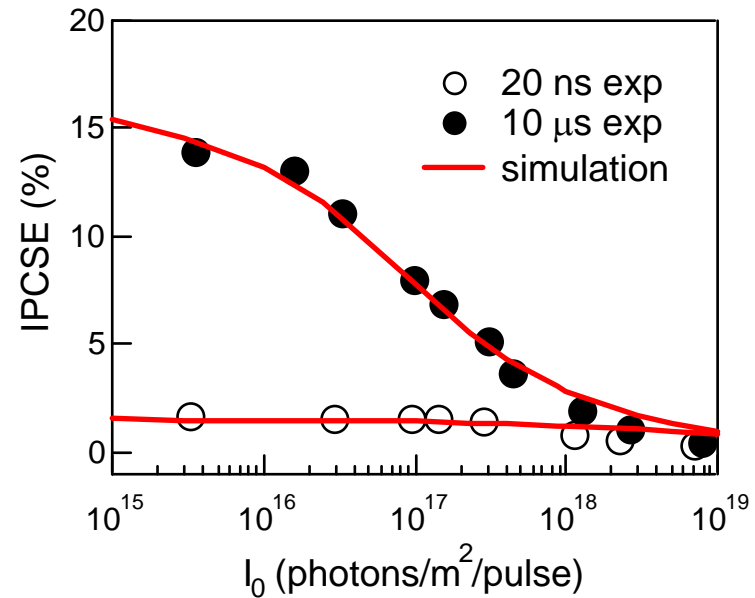
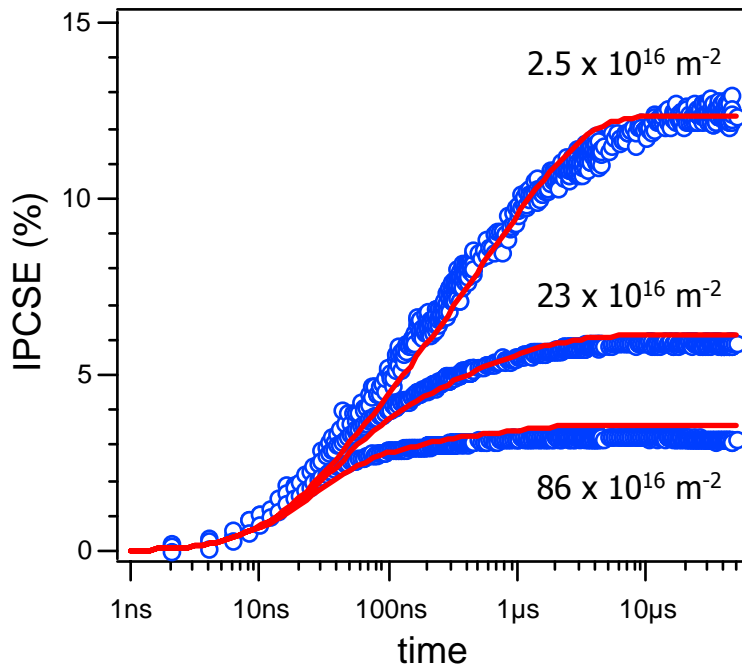
Monte Carlo computer simulations

injection in TiO₂



- 3D excitation profile; Lambert-Beer along x
- Exciton mean lifetime τ and diffusion coefficient D
- Excitons perform 3D random walk
- Annihilation if 2 excitons within distance R_{ann}
- Electron injection with chance ϕ_{inj} at TiO₂ interface
- Output: charge separation efficiency vs time

Triplet exciton diffusion in PdTPPC simulated



- $D=8 \times 10^{-11} \text{ m}^2/\text{s}$ (5 ns hopping time for $d=1.5 \text{ nm}$) , $\tau \geq 10 \mu\text{s}$
- Exciton diffusion length $\geq 28 \text{ nm}$!
- $\phi_{inj} = 0.4$
- $R_{ann} = 1.5 \text{ nm}$

$D = 8 \times 10^{-11} \text{ m}^2/\text{s}$ for triplets in PdTPPC

literature:

$D = 2.8 \times 10^{-8} \text{ m}^2/\text{s}$ for singlets in Ic porphyrin derivative

$D = 10^{-8} - 10^{-6} \text{ m}^2/\text{s}$ for singlets in conjugated polymers

$D = 10^{-12} \text{ m}^2/\text{s}$ for triplets in ZnTPPC

Conclusion

- Singlet-triplet mixing due to heavy atoms enhances exciton diffusion length
- Simple bilayer devices may become competitive with designs based on mesoporous nanocrystalline materials

Acknowledgments

Dr. J.E. Kroeze, Dr. T.J. Savenije and Dr. J.M. Warman

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