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Conversion of Hydrocarbons into Syn-Gas Stimulated by Non-thermal Atmospheric Pressure Plasma

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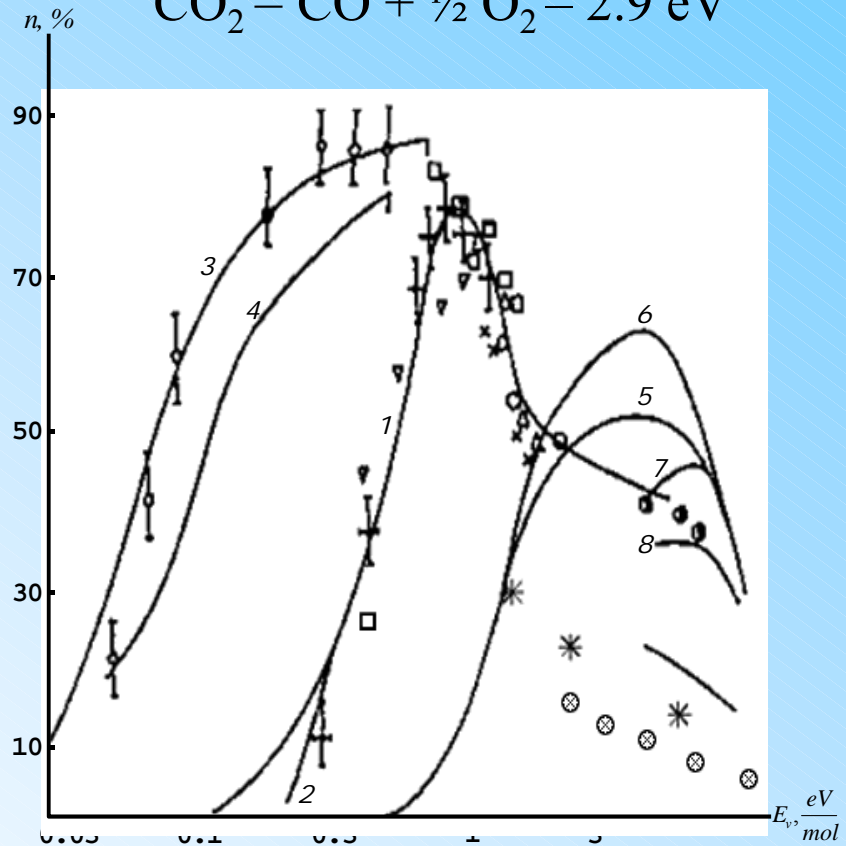
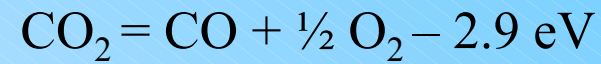
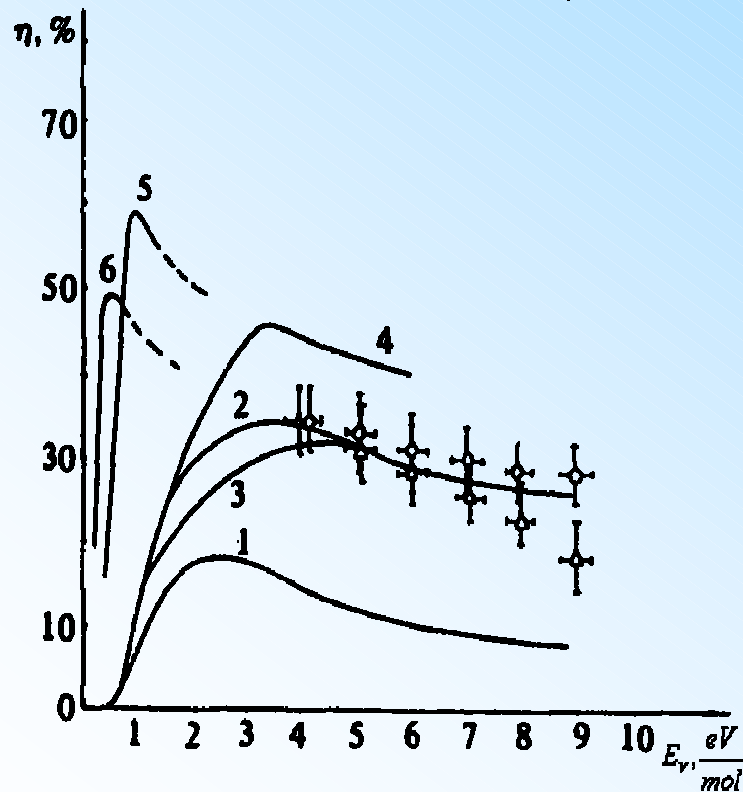
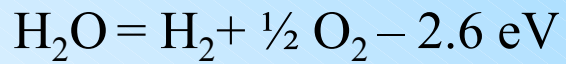
- Plasma Catalysis Vs. Plasma Processing
- Methane Conversion to Hydrogen
- Gliding Arc in Tornado
- Experiments and Modeling
- Plasma Catalytic Conversion of Hydrocarbons as Hydrogen Production Technology





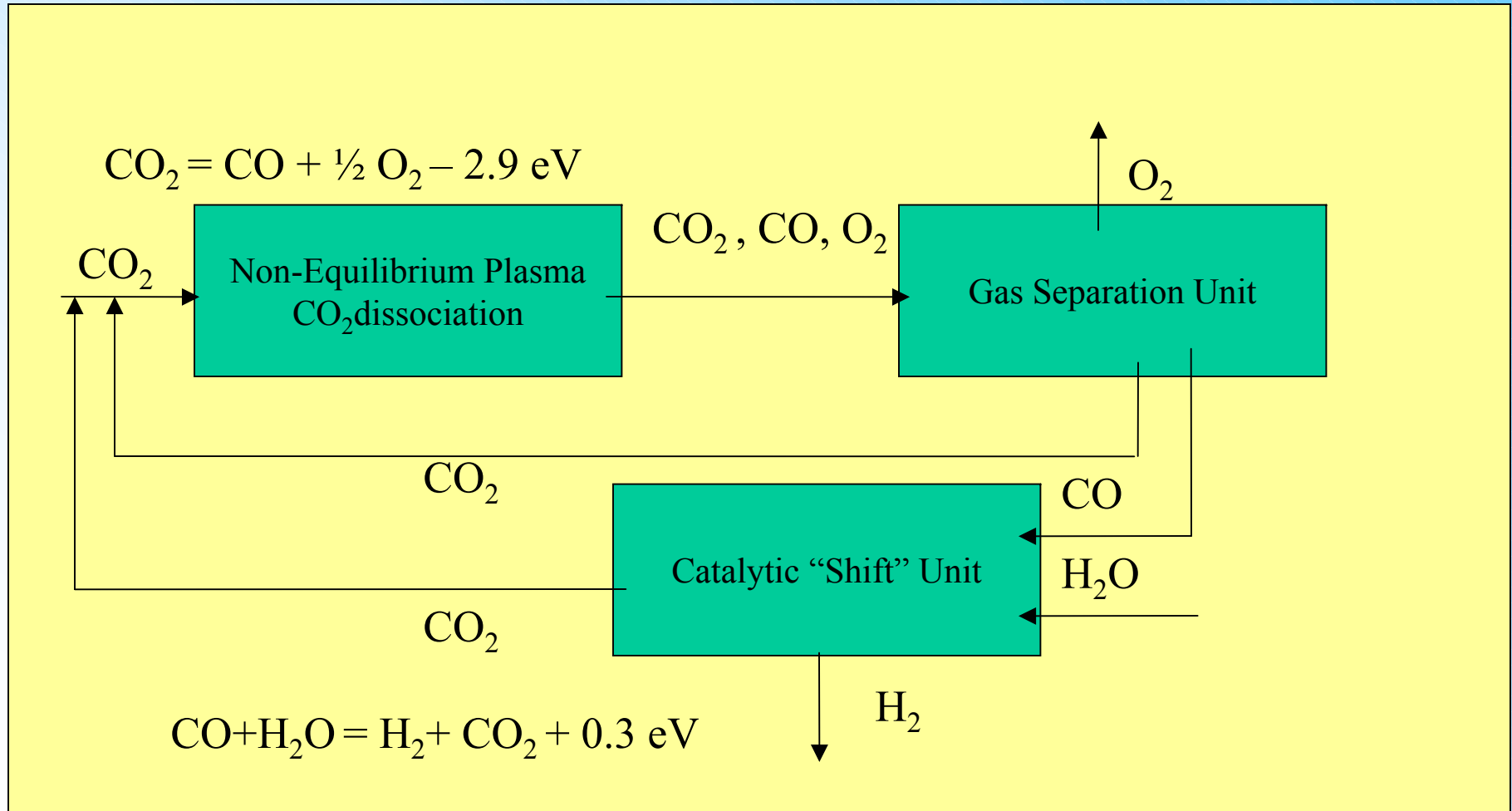
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Plasma-Chemical Hydrogen Production from Water





Two Step Plasma Cycle for Hydrogen Production





Plasma Catalysis Vs. Thermo-Catalytic Partial Oxidation



Thermo-Catalytic Conversion:

- High Temperature Requirements (>1100K)
- Large Specific Size of Reactor
- Special Materials Requirements and Reactor Design
- Sulfur from Natural Gas Causes Catalyst Poisoning
- Low Conversion at Moderate Equivalence Ratios (3.0-3.5)

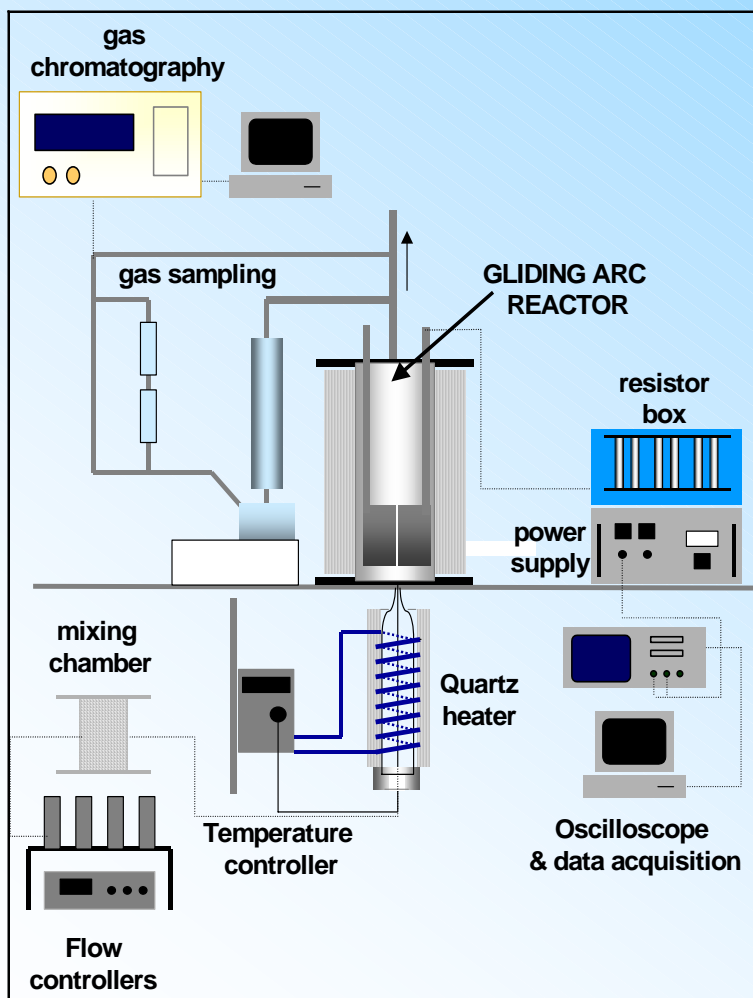
Plasma-Catalysis:

- Low Temperature Operation (~750K)
- Large Specific Productivity
- Lower Temperature Requirements
- No Sensitivity to sulfur or other impurities
- Possibility to Operate at High Equivalence (3.5-4.5)

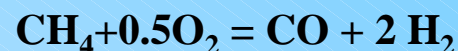


Plasma As Catalyst

Plasma Catalytic Hydrogen Production from Natural Gas



Plasma PO optimal parameters:



optimal equivalence ratio = 3.3,

$$[\text{O}_2] / [\text{CH}_4] = 0.6$$

Preheating temperature = Internal, 750K

Conversion = 92%

Electric energy cost :

experimental = 0.06 kWh/m³

modeling EQ = 0.11 kWh/m³

modeling NE = 0.07 kWh/m³

Output Syn-Gas energy = 3.00 kWh/m³

power for 100,000 barrel/day of Liquid Fuel:

experimental = 4.5 MW

modeling EQ = 8.2 MW

modeling NE = 5.2 MW



Which Plasma to Choose?



**Thermal
Plasma**

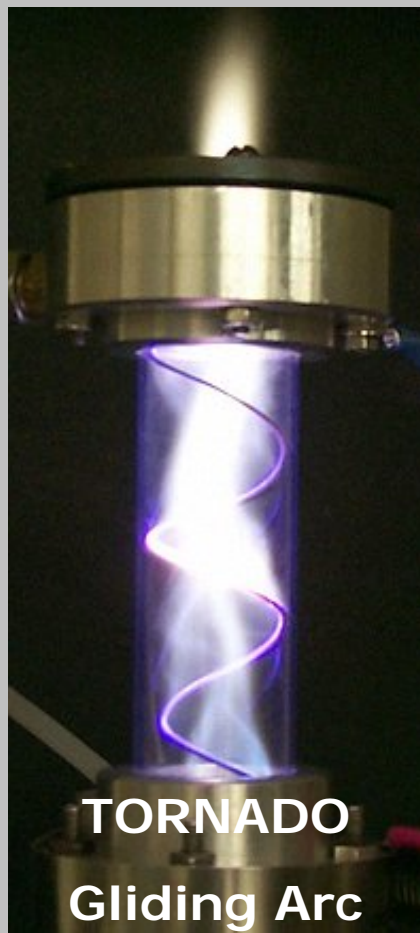
**Non-Thermal
Plasma**



X-ray view of the sun



ICP Torch



**TORNADO
Gliding Arc**



Aurora



Corona Discharge



Gliding Arc as Transitional Non-Equilibrium Plasma:



THERMAL PLASMA

- Very High Plasma power and density.
- High Gas temperature.
- No selective chemical process can be achieved.

NON-THERMAL PLASMA

- Low gas temperature and very high electron temperature.
- Low Power Density
- Chemical Selectivity can be achieved.

MAJOR CHALLENGES :

- Power Density & Productivity.
- Selectivity.



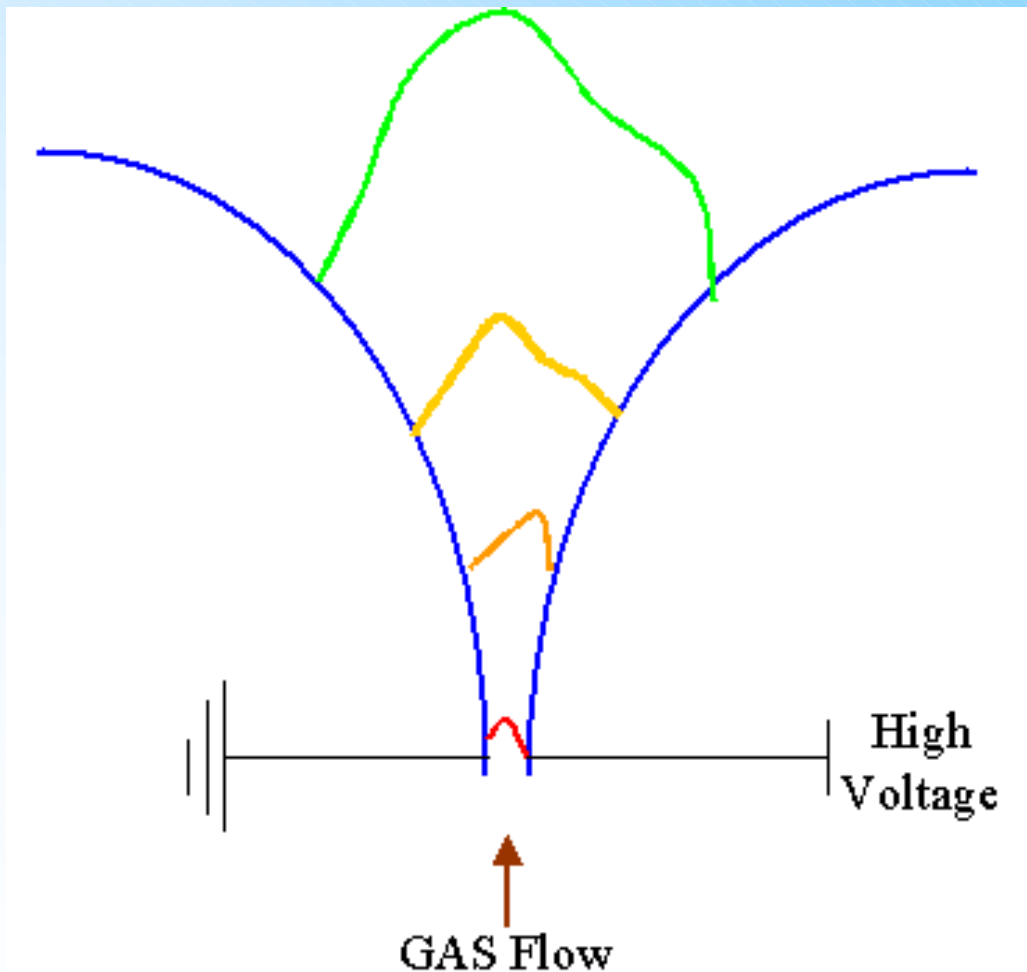
“Gliding Arc in Tornado”



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“GLIDING ARC in Flat Geometry”

Fast Equilibrium to Non-Equilibrium Transition



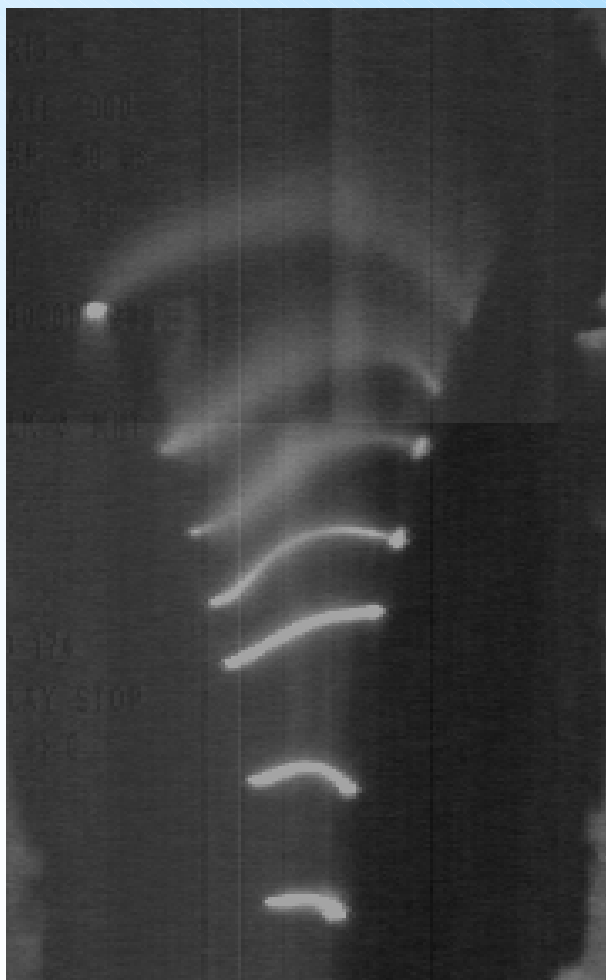
- Arc starts in a narrow gap between diverging electrodes in a gas flow.
- Current increases very fast and the voltage on the arc drops.
- Gas flow forces the arc to move along the diverging electrodes and elongate. The growing arc demands more power to sustain itself.
- Arc cools down and becomes Non-Equilibrium to sustain itself with less power.



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“GLIDING ARC in Flat Geometry”

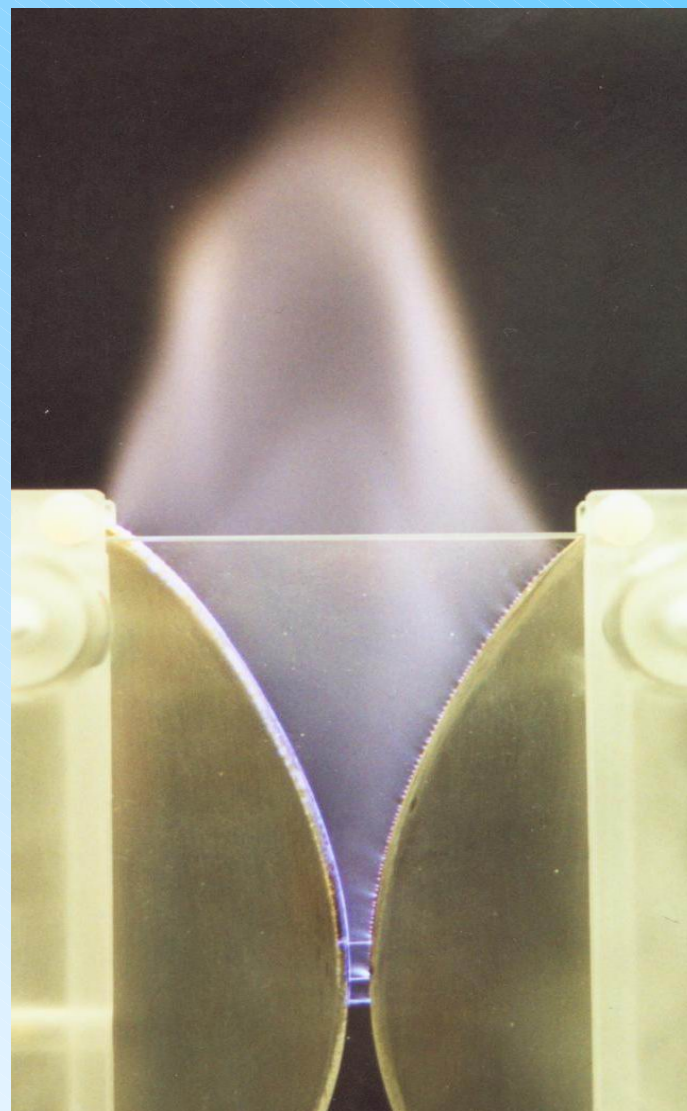
Fast Equilibrium to Non-Equilibrium Transition



Extinction

Elongation

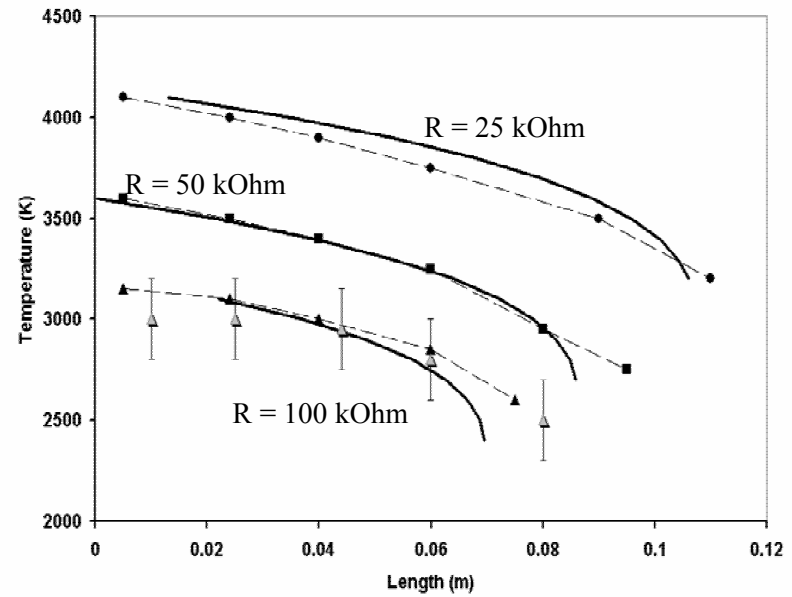
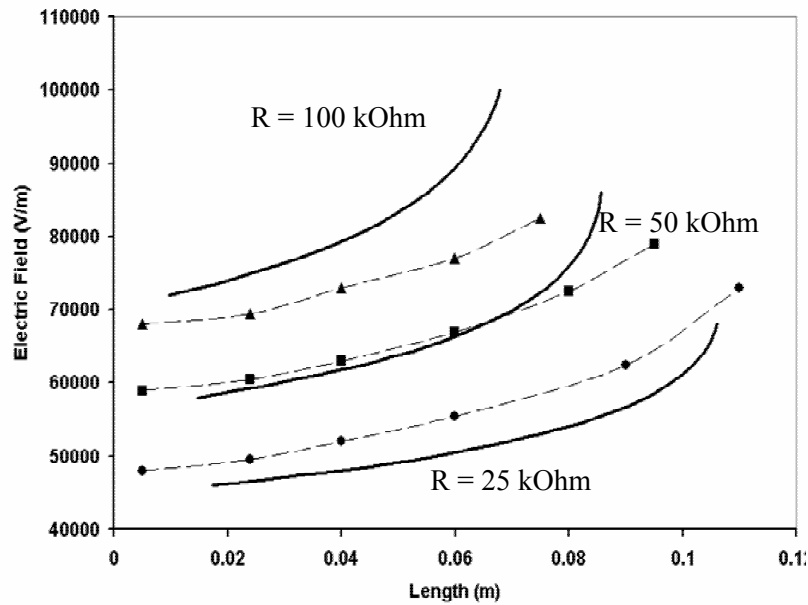
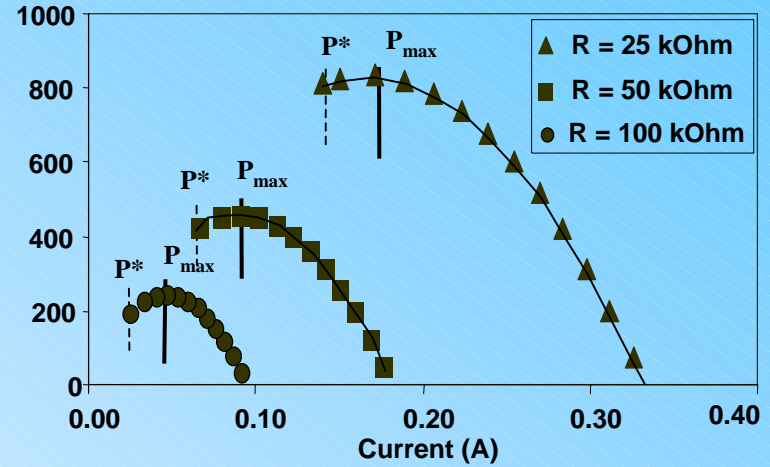
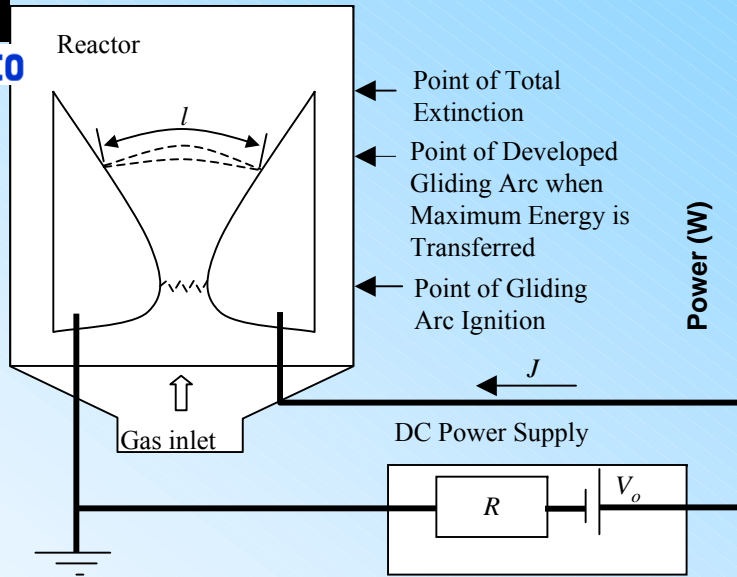
← Initial Breakdown →





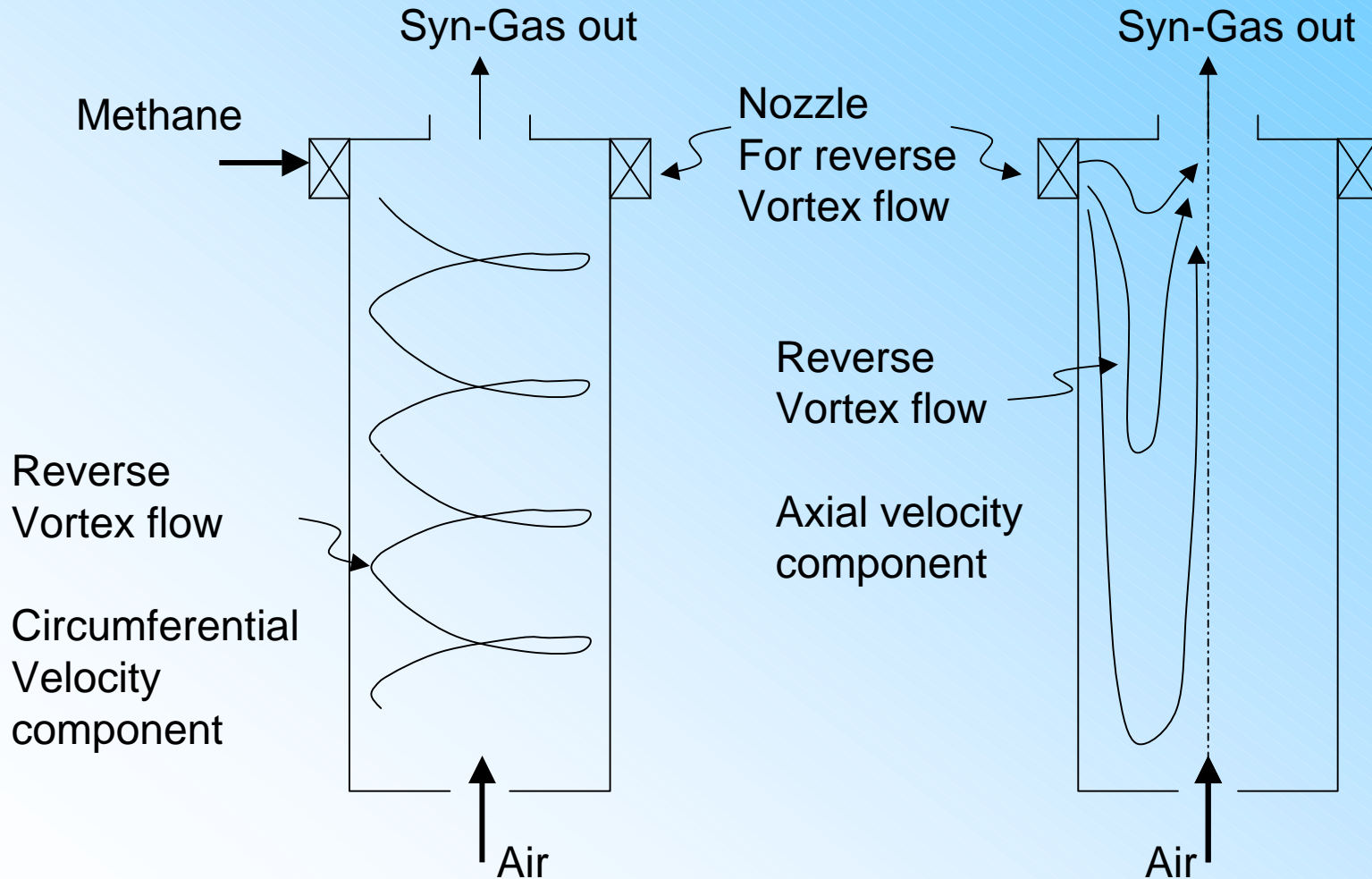
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Non-Equilibrium Gliding Arc





Reverse Vortex Flow (Tornado) Stabilization of Gliding Arc



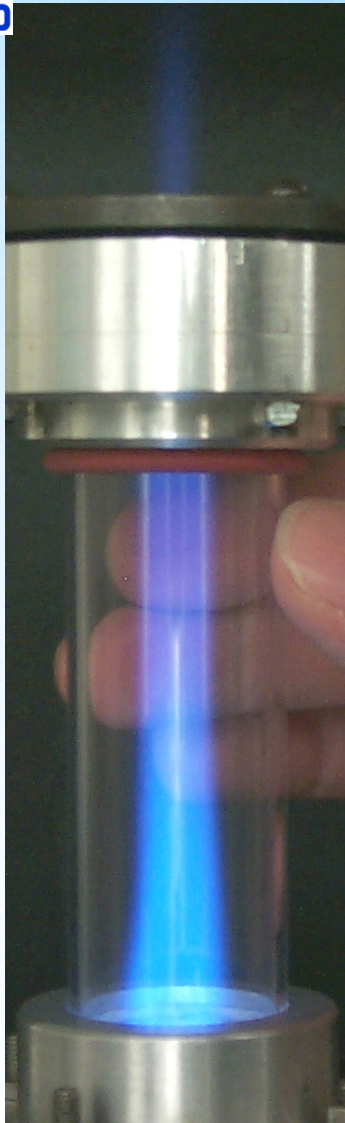


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Gas Burner with Reverse Vortex Flow, “Tornado” Stabilization of Flame



$\varepsilon \sim 0.5$



$\varepsilon \sim 1.1$



$\varepsilon \sim 1.8$

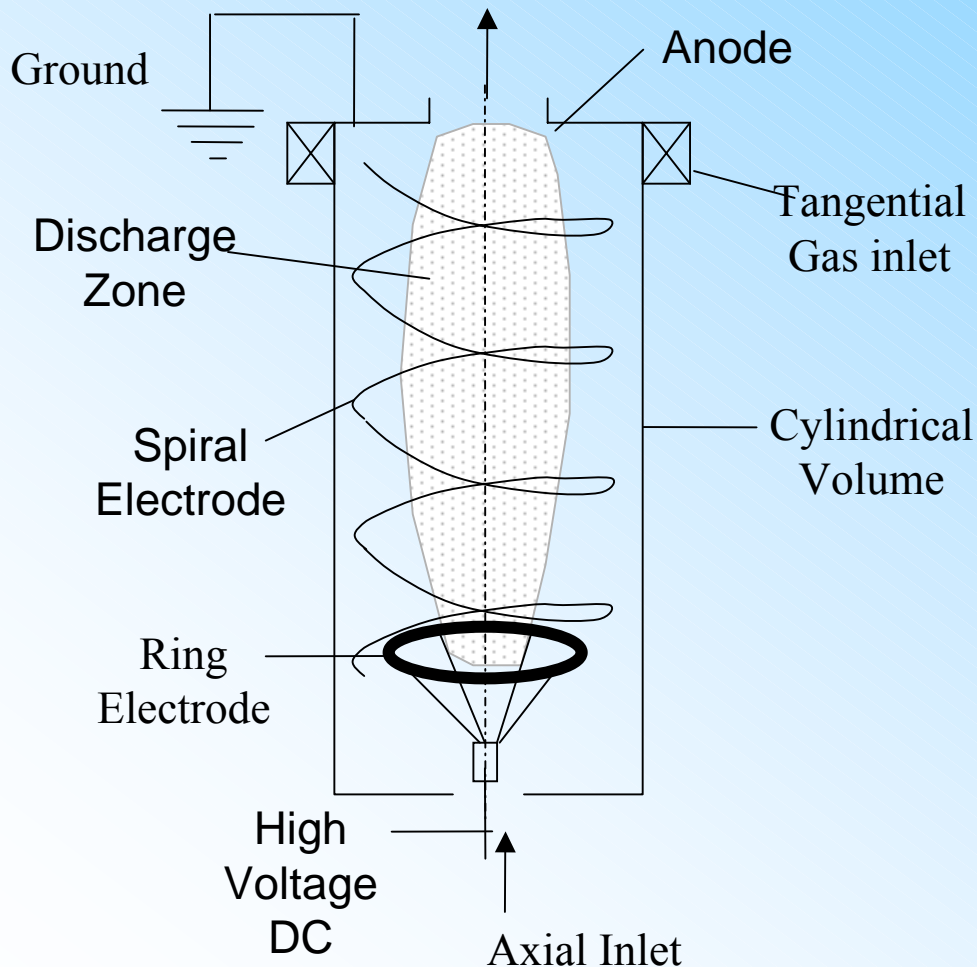




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“THE GLIDING ARC IN TORNADO”

Gliding Arc in Tornado Flow



- Gliding Arc in Tornado works in a Reverse Vortex Flow setup.

- A circular and spiral electrode is placed in the plane of the flow act as diverging High Voltage DC Electrodes.

- The flow conditions and the characteristics of the power supply determine the shape of the spiral electrode.

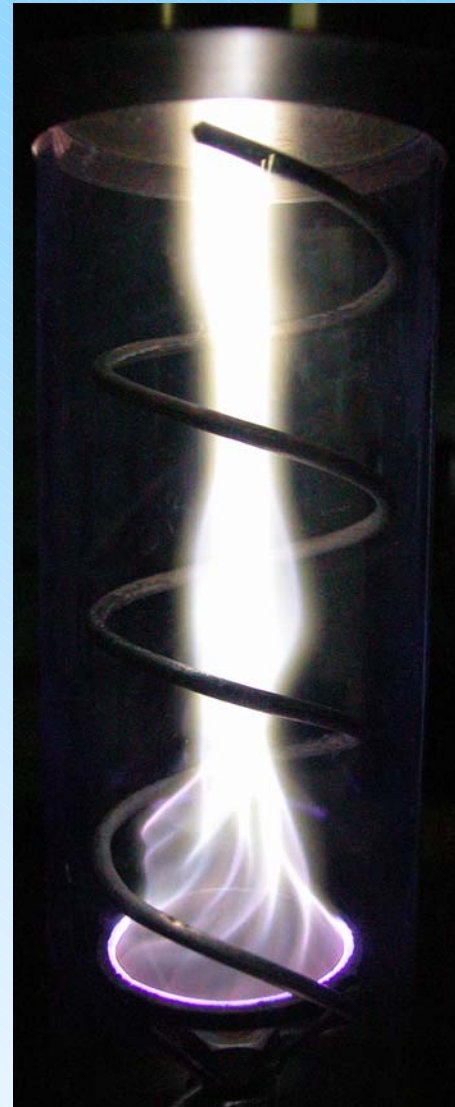
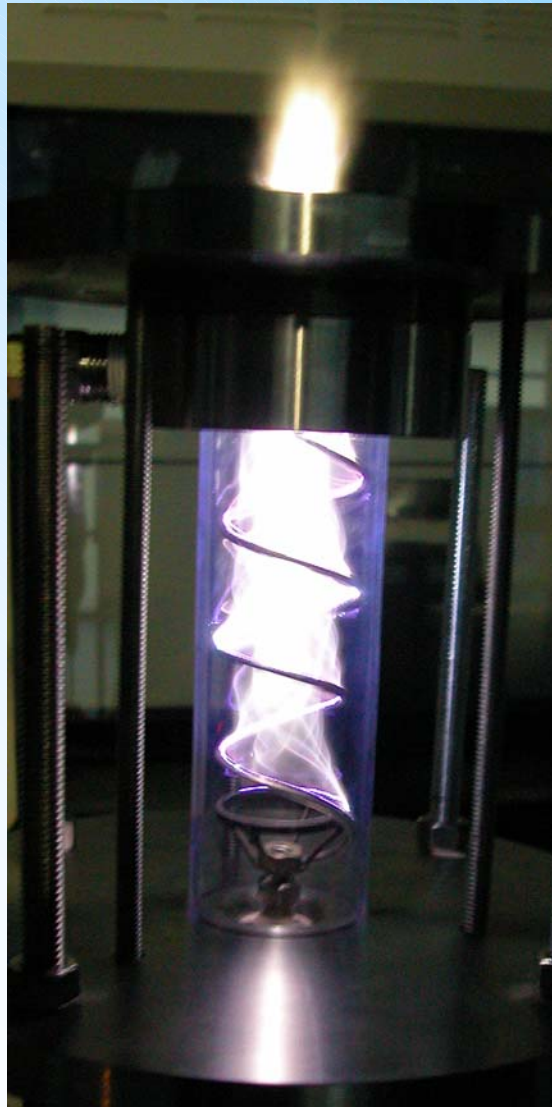
Schematic Diagram for GAT reactor.



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Gliding Arc “Tornado”

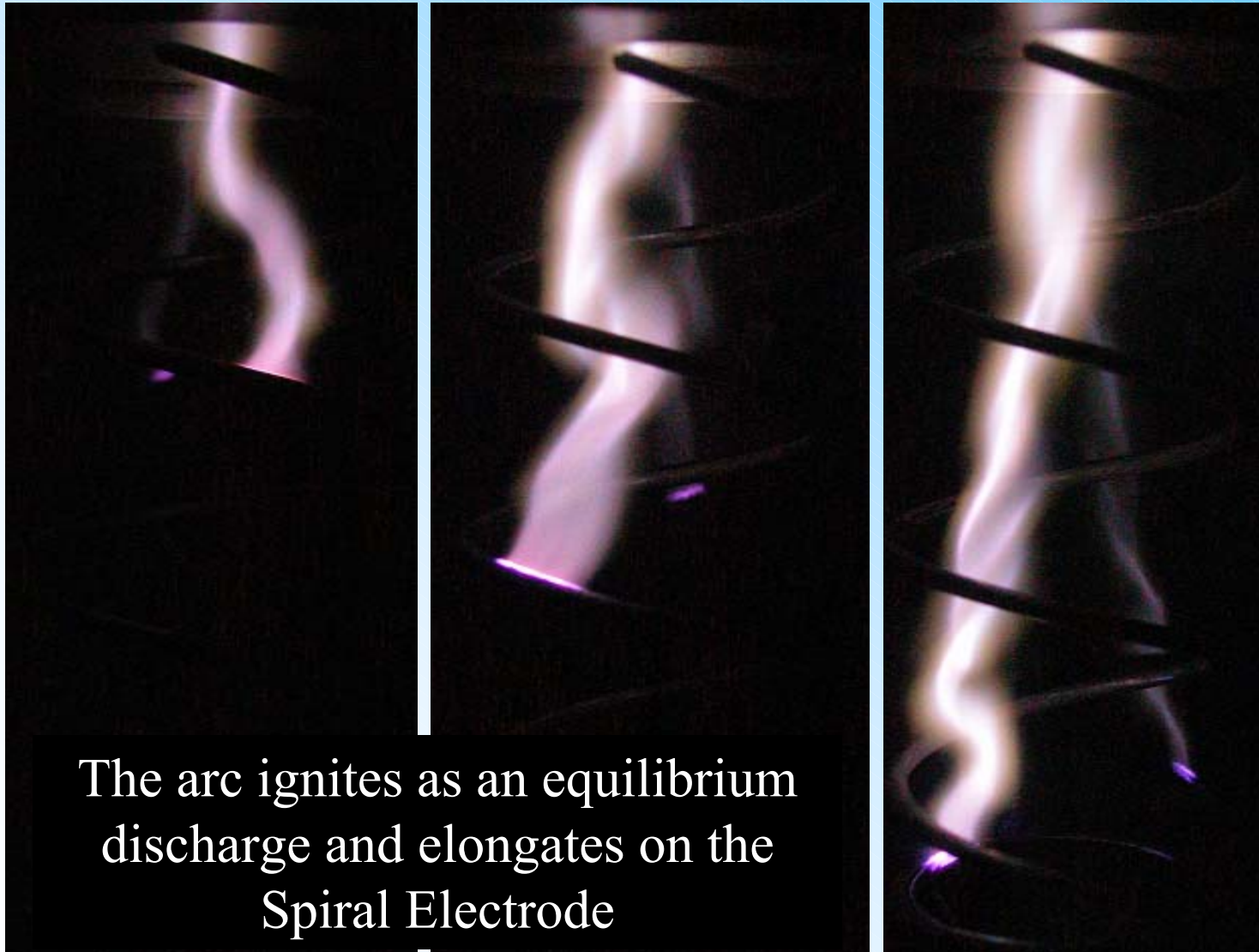
From the Spiral to the Ring





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Gliding of Arc on Spiral Electrodes in Reverse Vortex Flow

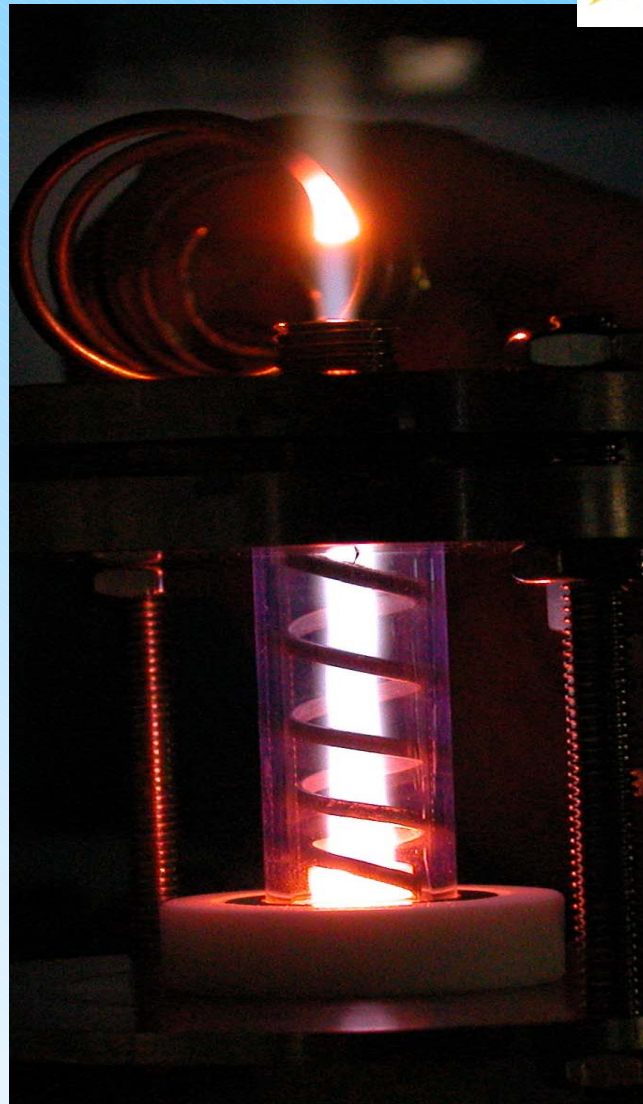
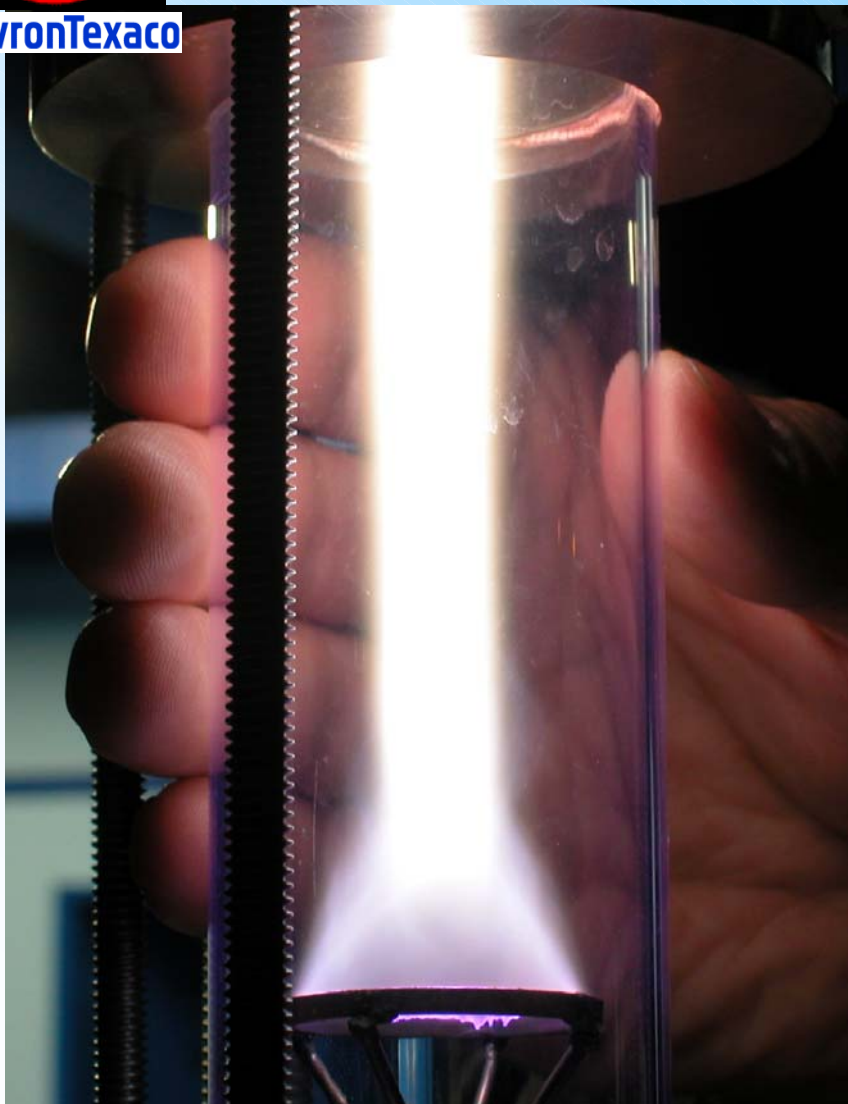


The arc ignites as an equilibrium discharge and elongates on the Spiral Electrode



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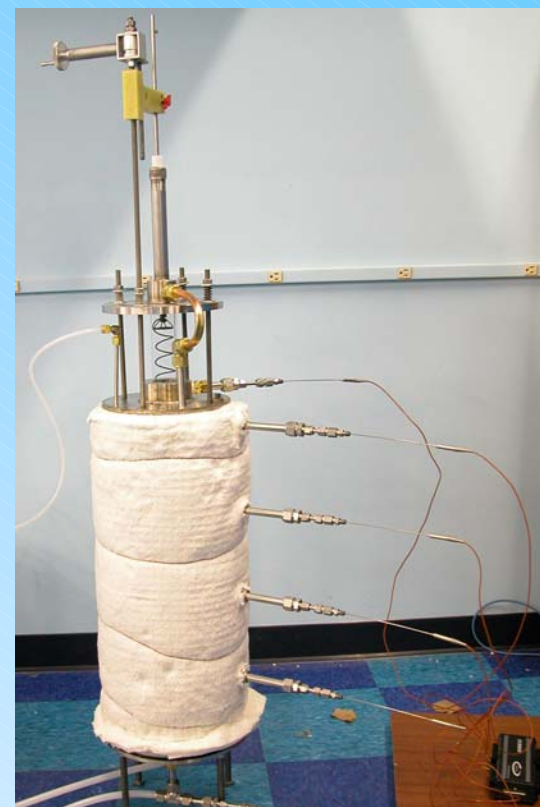
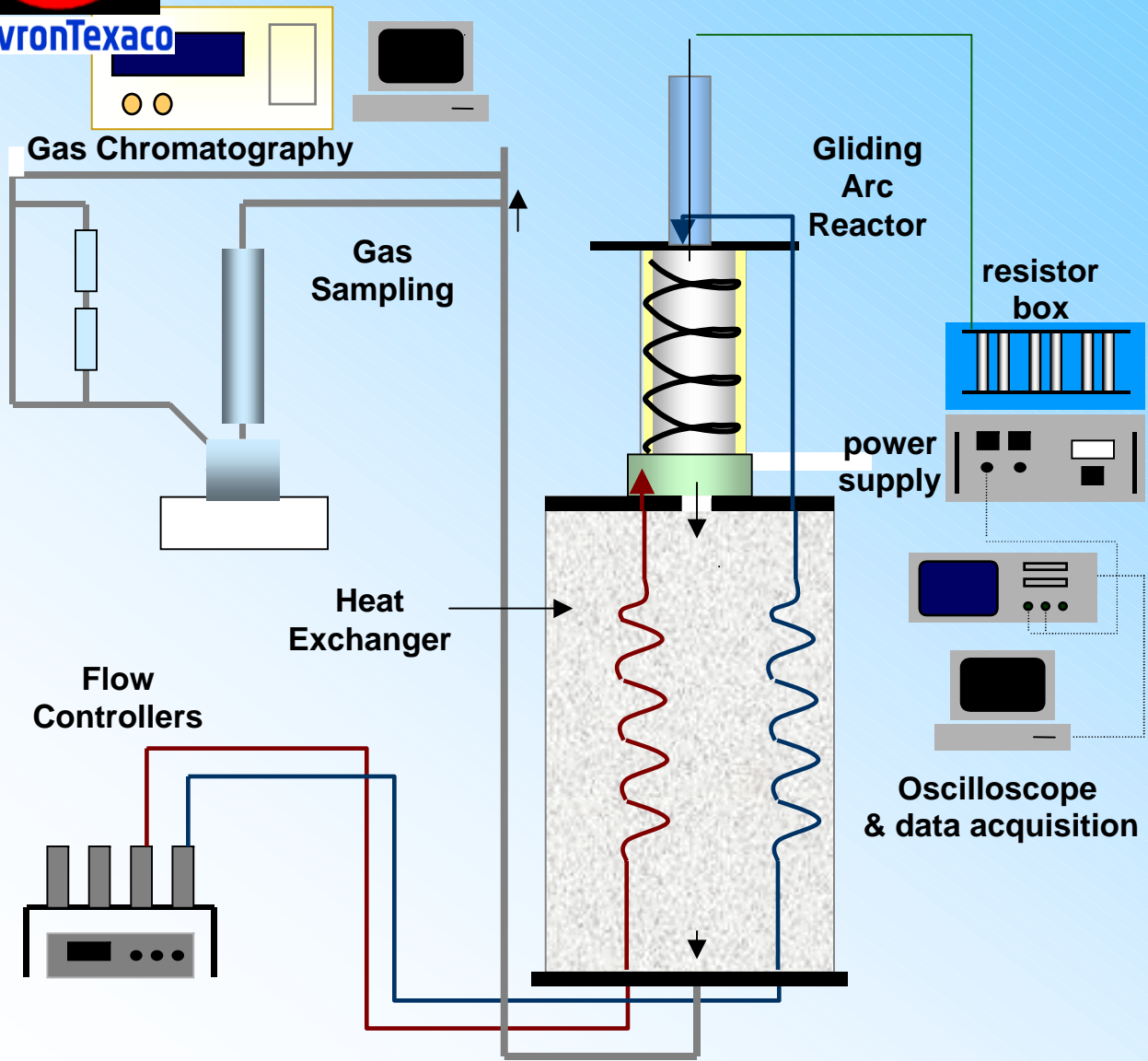
“It Can Melt a Metal Rod But You Can Touch It”





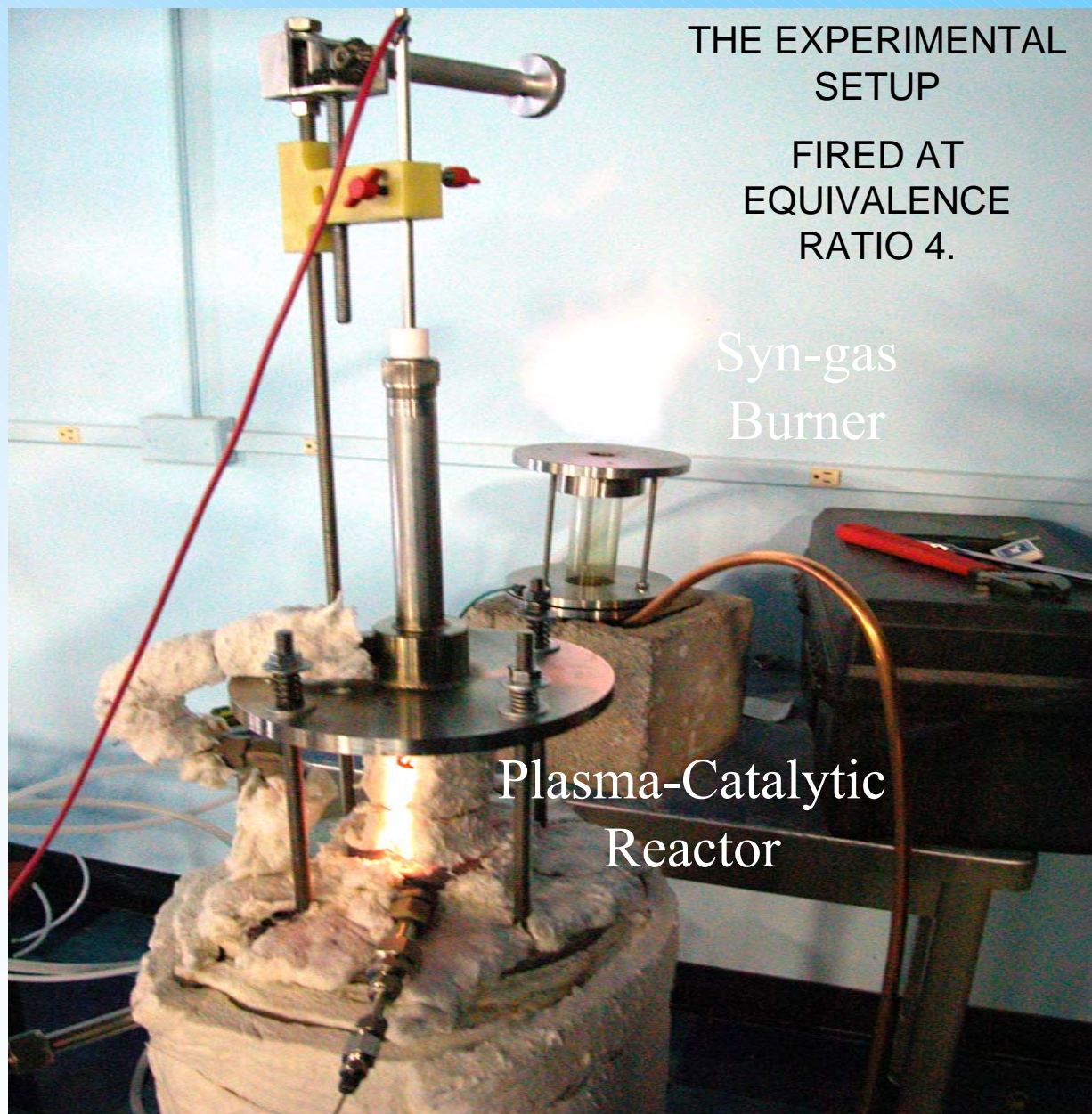
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EXPERIMENTAL SETUP





Plasma Catalytic Methane Partial Oxidation



THE EXPERIMENTAL
SETUP

FIRED AT
EQUIVALENCE
RATIO 4.

Syn-gas
Burner

Plasma-Catalytic
Reactor



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Chemical Kinetics Simulation



The entire process includes three main stages:

- 1.) Electric discharge stage
- 2.) Mixing of air and post discharge methane.
- 3.) Post discharge regime .

□ The neutral species chemistry was described by the GRI-MECH 2.11 kinetic scheme including 65 species and 200 reactions. In the post discharge region the gas mixture is allowed to react within a certain residence time, with small heat losses.

The syn-gas production is characterized by two stages:

- 1.) The short combustion zone.
- 2.) The longer zone, which can be called as the reforming zone

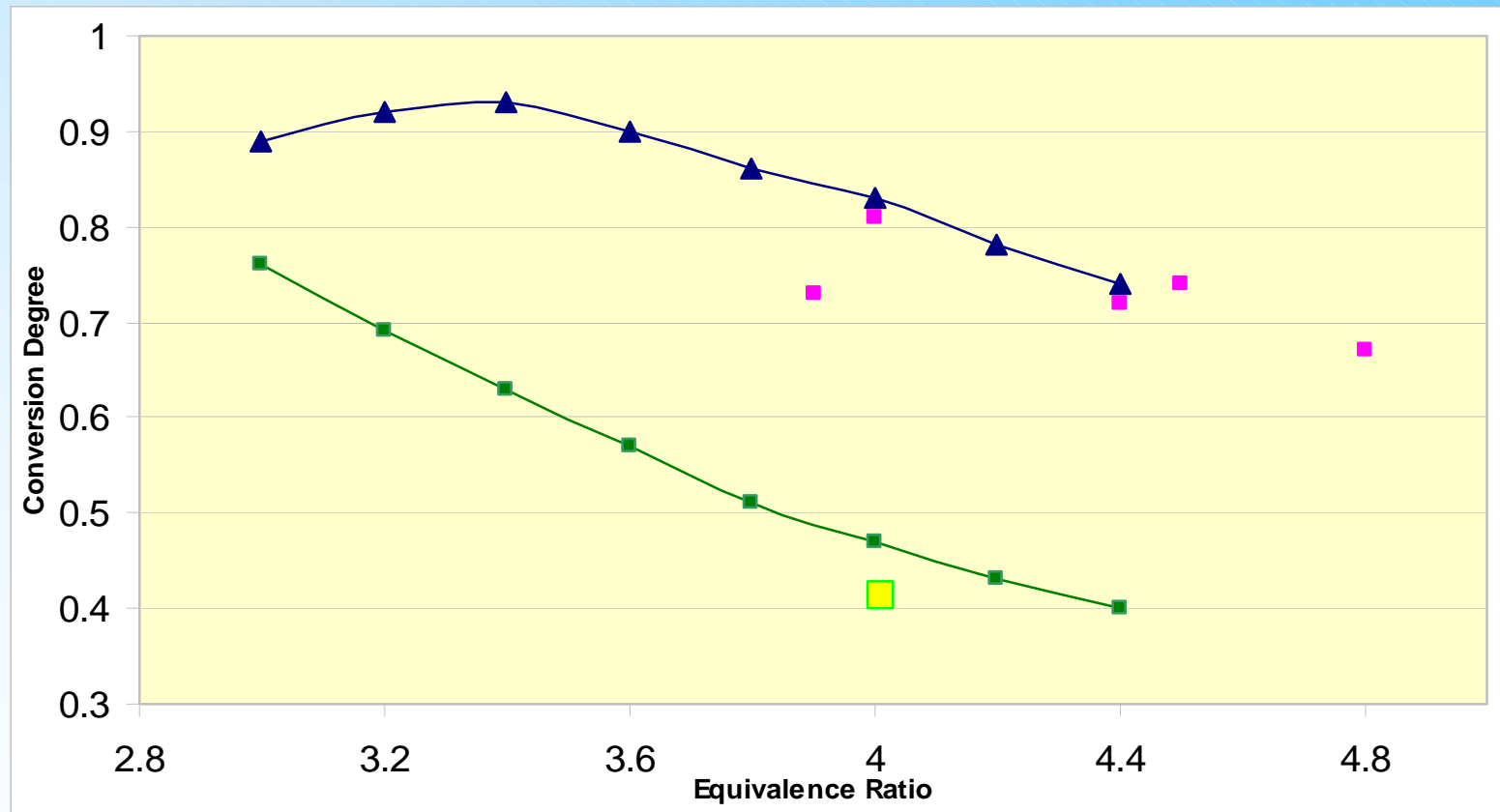
□ At the first stage the concentrations of O and OH radicals are relatively large, and it is characterized by very fast chemistry.

□ At the second stage, much slower chemistry takes place. Water and CO₂ are consumed, yielding hydrogen and CO, the corresponding reaction are endothermic ones, causing the observable temperature decrease.



Simulation Vs Experiments

The conversion degree: $\alpha = ([H_2] + [CO]) / 3[CH_4]$



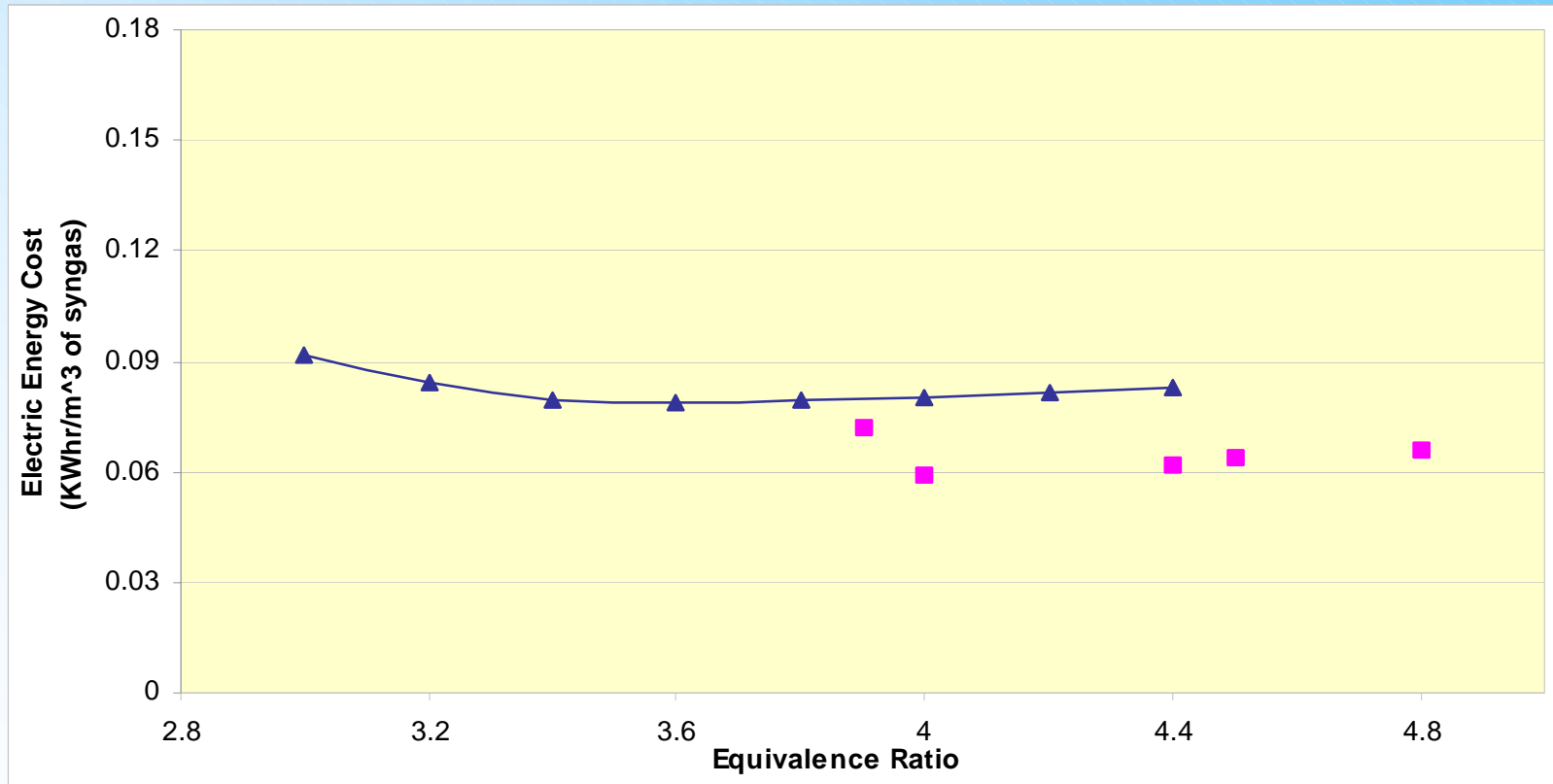
- ▲ Modeling results With plasma
- Modeling results without plasma
- Experimental results with plasma
- Experimental results without plasma



Simulation Vs Experiments



Electric Energy Cost = W_{el} (KW-hr)/ meter cube of
Syn-Gas (Output Syn-Gas Energy = 3.00 kWh/m³)

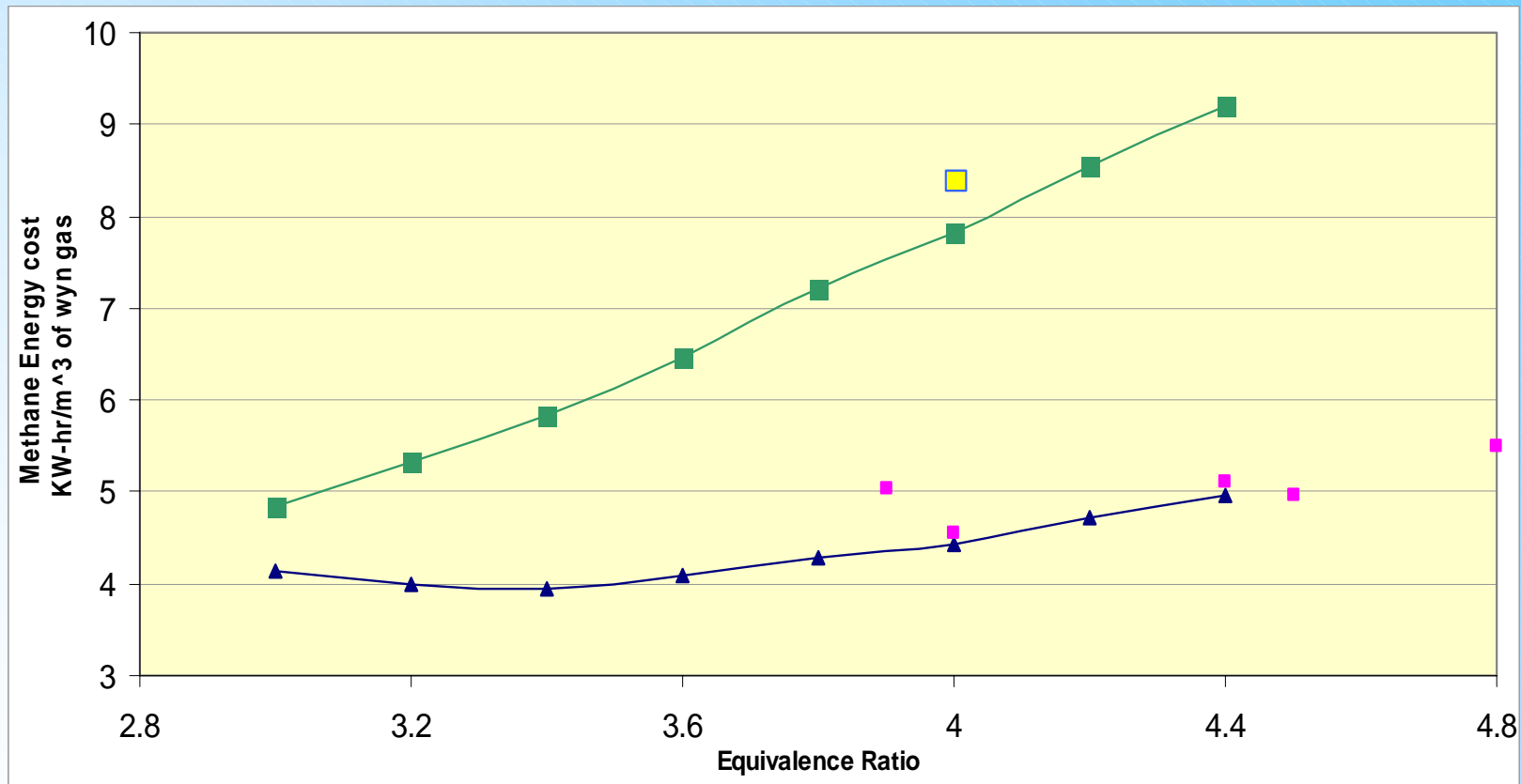


- ▲ Modeling results
- Experimental results



Simulation Vs Experiments

Methane Energy Cost = [CH₄] (KW-hr) per meter-cube of Syn-Gas



▲ Modeling results with Plasma
■ Experimental results with plasma

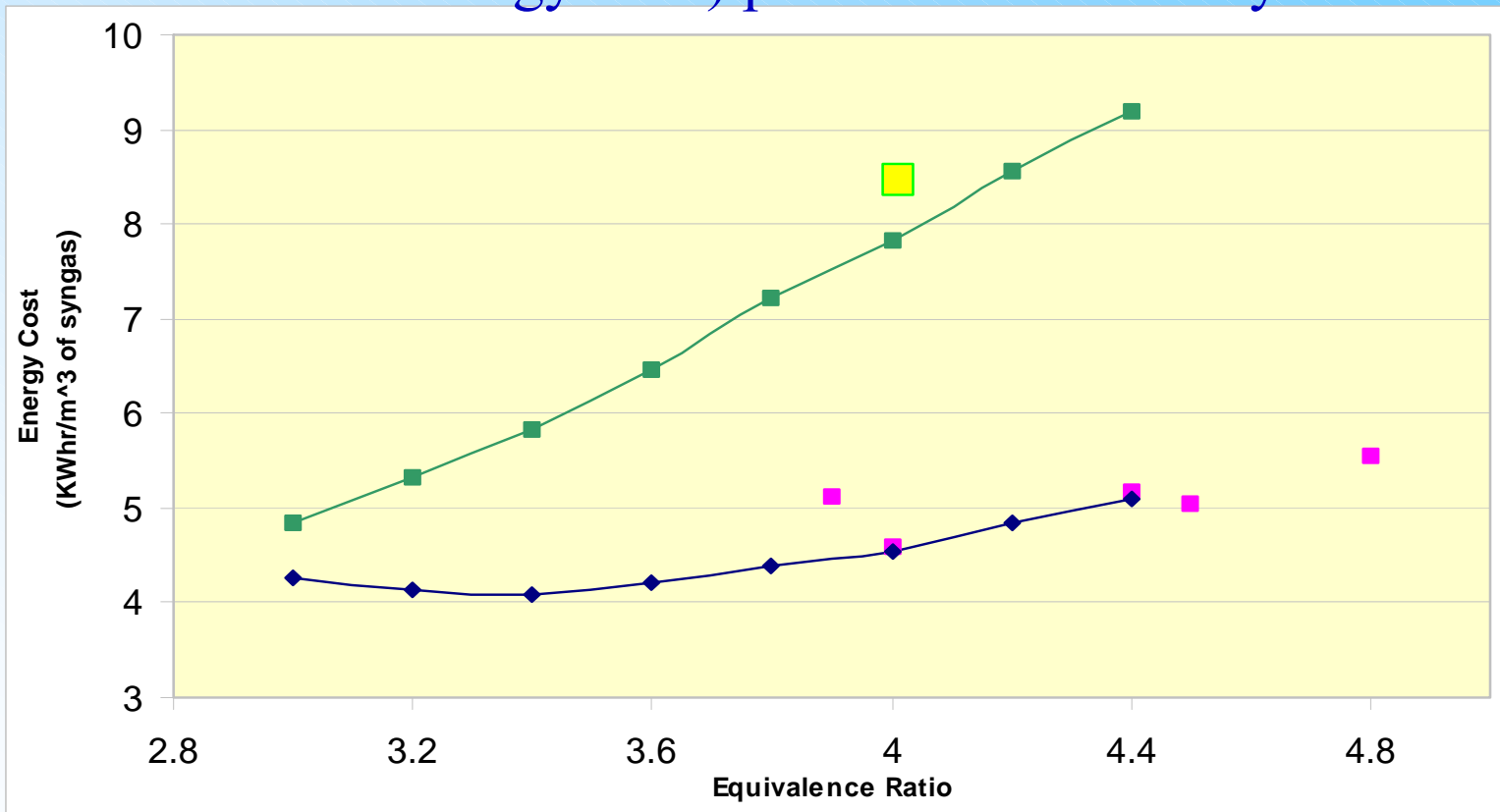
■ Modeling results without plasma
■ Experimental results without plasma



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Simulation Vs Experiments

Total Energy Cost = (Electric Energy Cost + Methane Energy Cost) per meter Cube of Syn-Gas



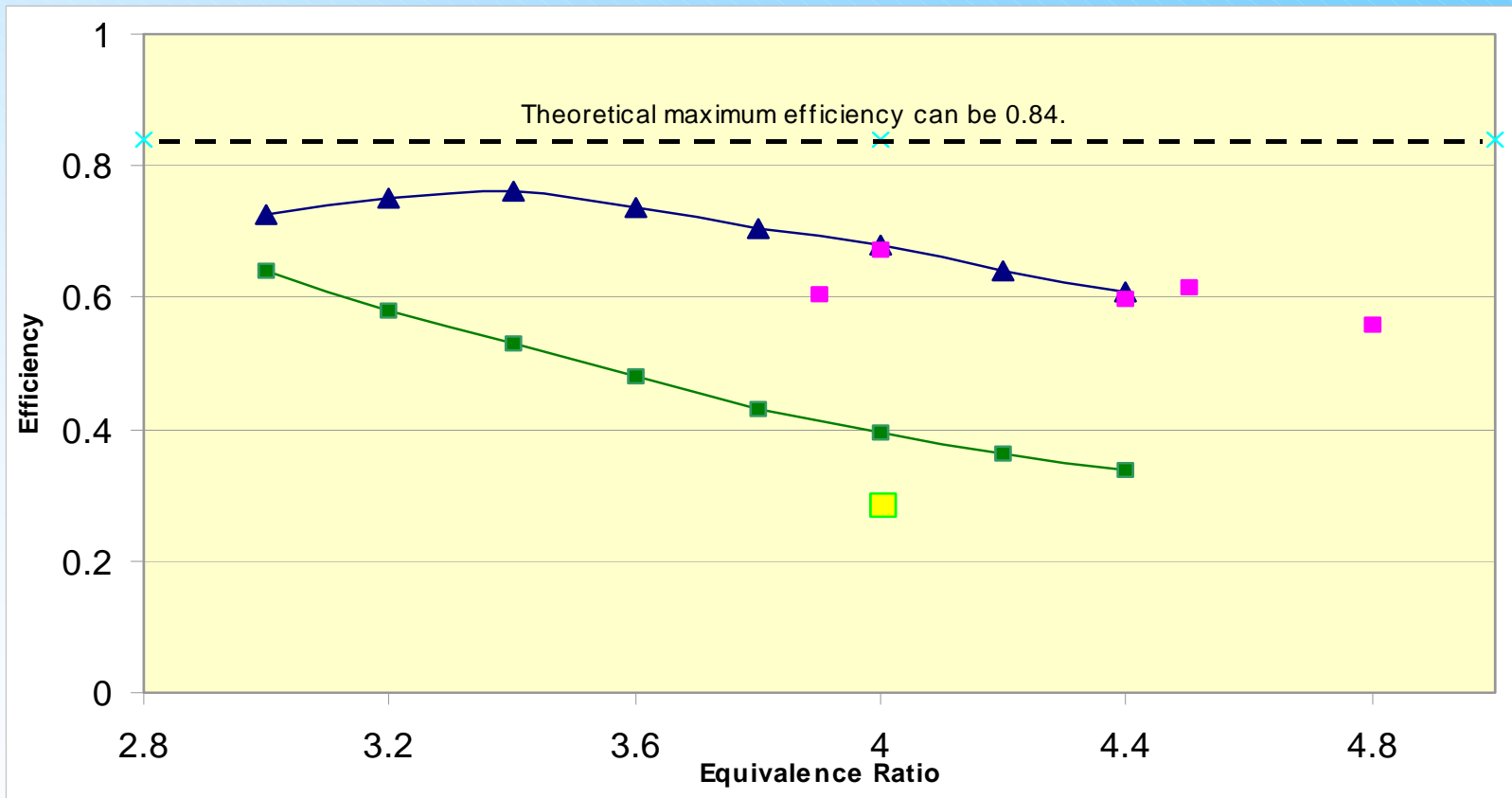
▲ Modeling results With plasma
■ Experimental results with plasma

■ Modeling results without plasma
■ Experimental results without plasma



Simulation Vs Experiments

Efficiency = KW-hr of Syn-Gas Produced / Total Energy Input in KW-hr



▲ Modeling results
With plasma
■ Experimental results
with plasma

■ Modeling results
without plasma
■ Experimental results
without plasma



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Plasma Catalysis Highlights:



- Only 2.0% of Total Energy Consumption Required for Plasma Power
- Electric Energy Cost 0.06 kWh/m³ of syn-gas (energy from syn-gas = 3.0 KW-hr/m³).
- 92% conversion at Equivalence ratio of 3.3.
- Internal Heat Recuperation (Preheating) at 750 K.
- No soot Deposition.
- Large Specific Production rates due to low residence times.
- Effective for Higher Hydrocarbon conversion to Syn-Gas.
- Not Sensitive to Sulfur and Other Impurities.



Best Regards, Chiranjeev
Kalra from the city of
brotherly love