



Stanford University

Global Climate & Energy Project

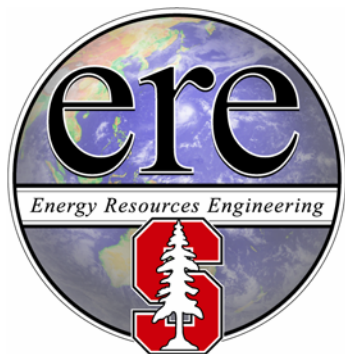
GCEP Research Symposium
September 18-20, 2006



CO₂ Sequestration and Enhanced Coalbed Methane Recovery

Tony Kavscek

Energy Resources Engineering



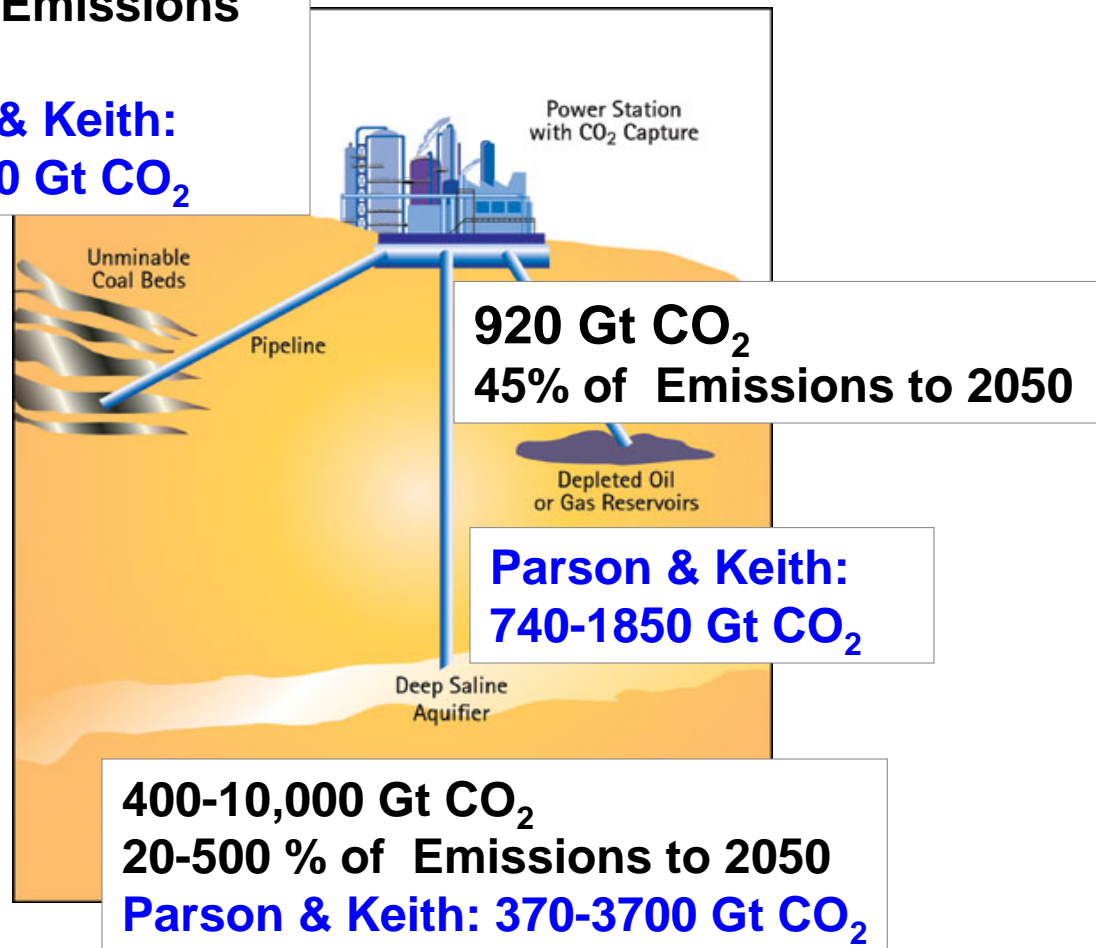


Why study coalbeds?



IEA: 40 Gt CO₂
<2% of Emissions
to 2050

Parson & Keith:
370-1100 Gt CO₂



- add certainty to volume estimates
- store other contaminants
- smother fires
- significant challenges for monitoring
- coal is intrinsically interesting

IEA: Comparative potentials at storage costs of up to \$20/t CO₂



What are we trying to do?



- Elucidate fundamental mechanisms relevant to enhanced methane production and gas storage in coal because poor predictions result, in part, from incomplete knowledge of physical mechanisms.
- Our niche is our ability to probe simultaneously sorption, transport, and permeability to gas
 - experimentally
 - analytically
 - numerically





Today's Presentation

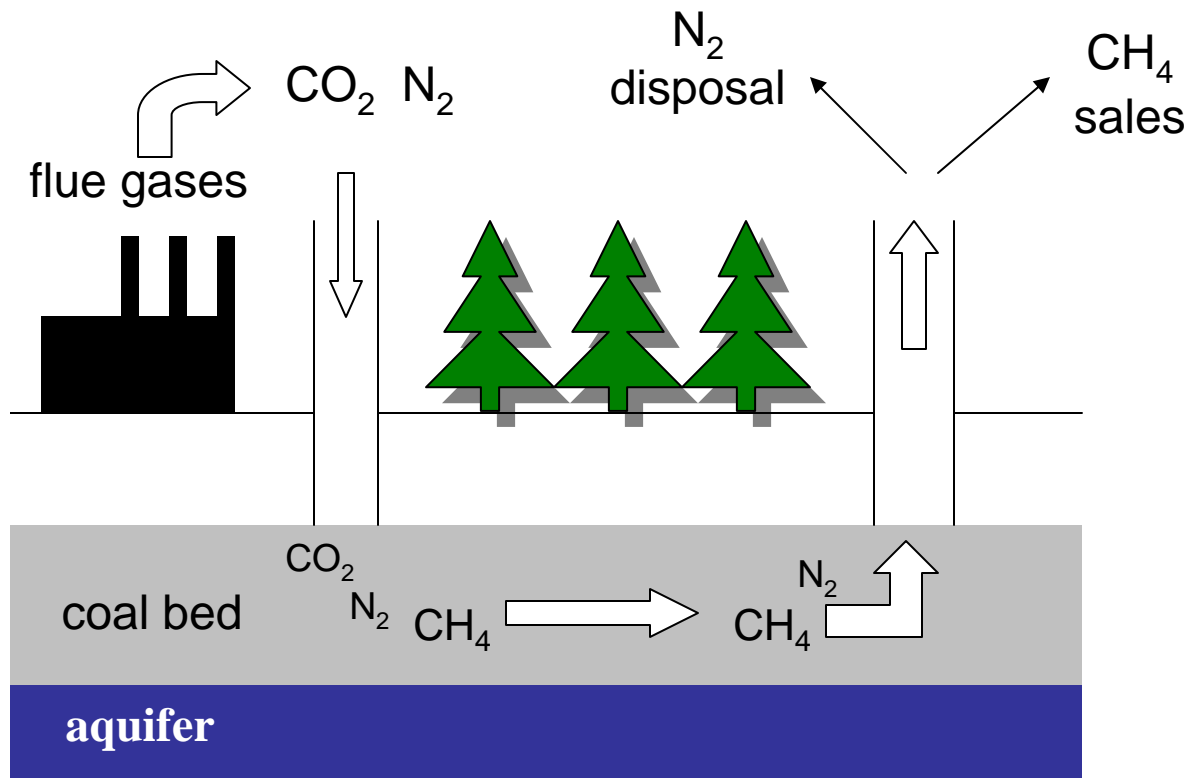


image source: C.J. Seto

- desorption through reduction of partial pressure
- 80-90% recovery of methane

- **How much gas adsorbs on coal?**
- **How do we probe flow mechanisms?**
- **Is coal permeability sensitive to the type of gas?**
- **How does water flow?**



How much gas adsorbs?

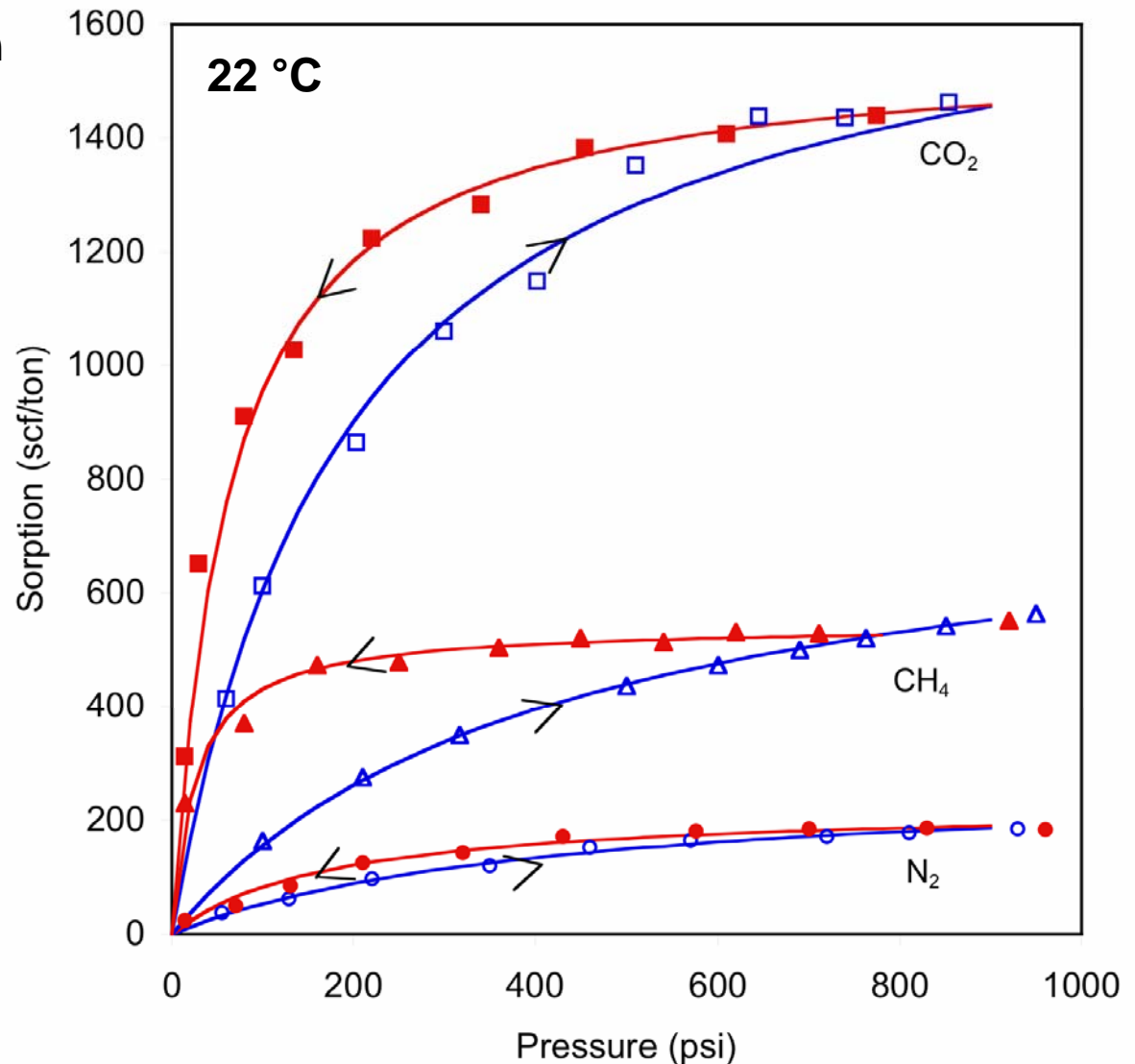
CO₂, CH₄, N₂ Sorption



Ground Powder River Basin (WY) Coal



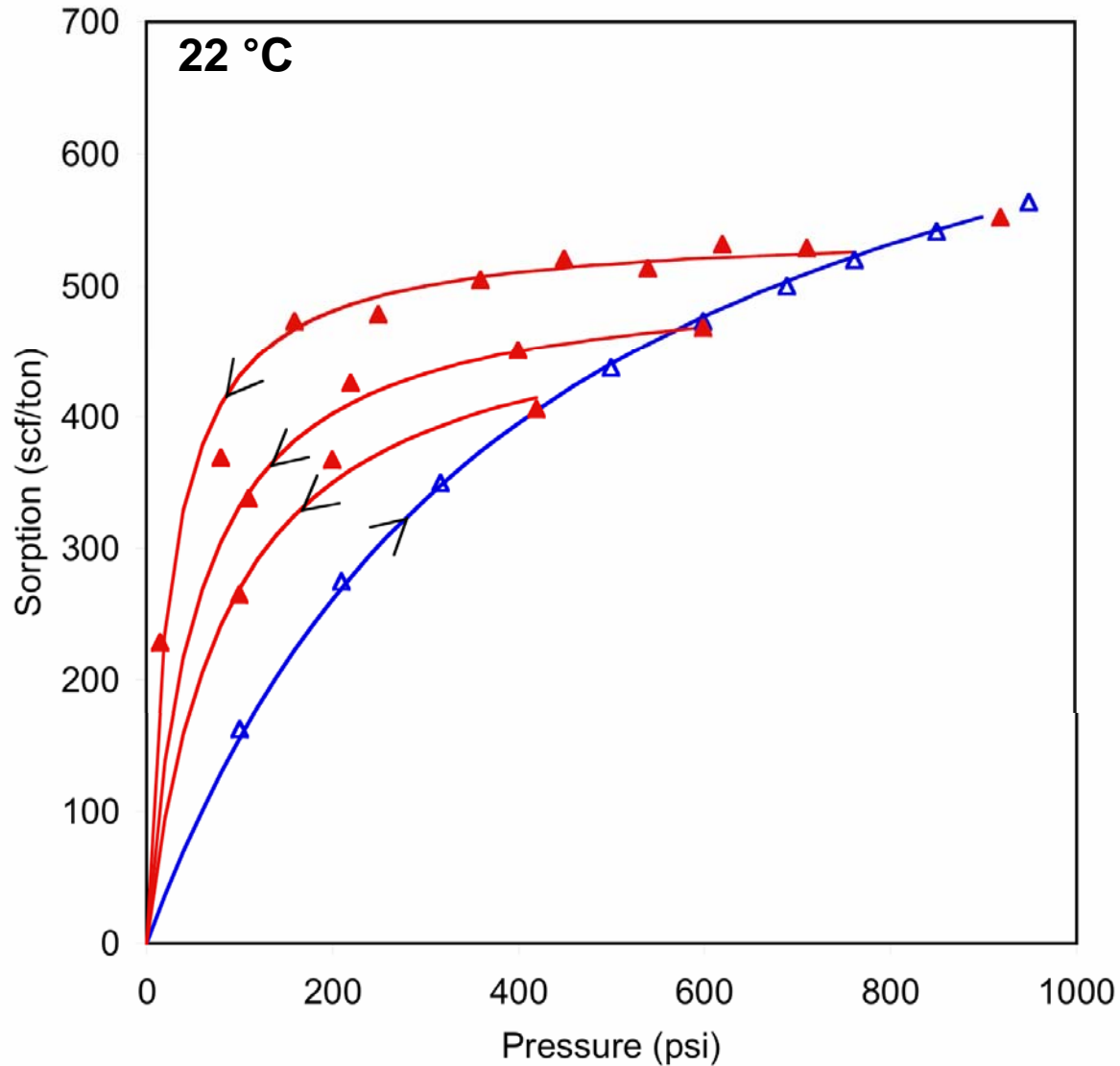
- Pure components are well fit by Langmuir isotherm
- CO₂ adsorbs preferentially
- adsorption hysteresis for all gases





CH₄ Scanning Loops

Initial pressure influences desorption hysteresis

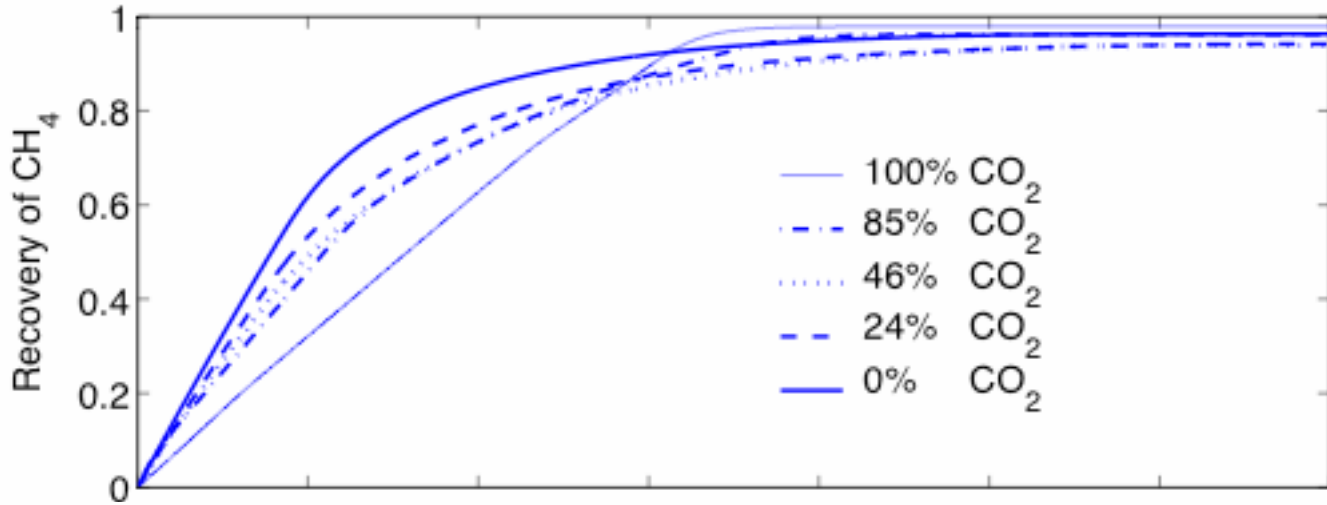




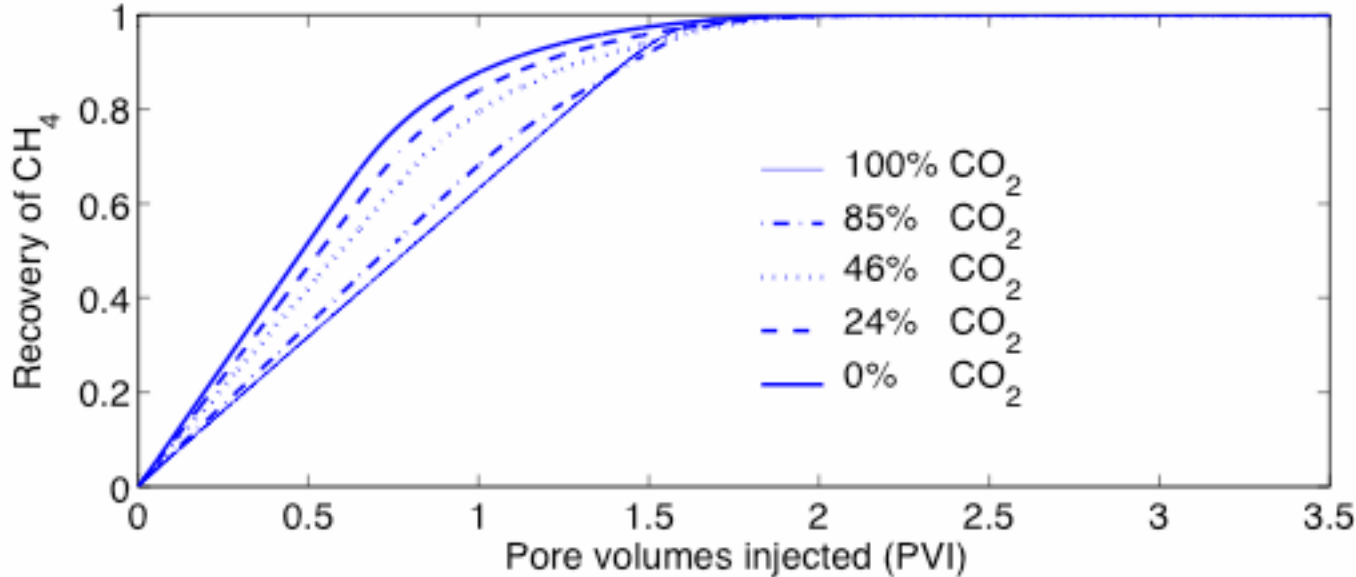
Competitive Adsorption and Recovery



experiment



simulation



$X\% \text{CO}_2 (1-X)\% \text{N}_2$

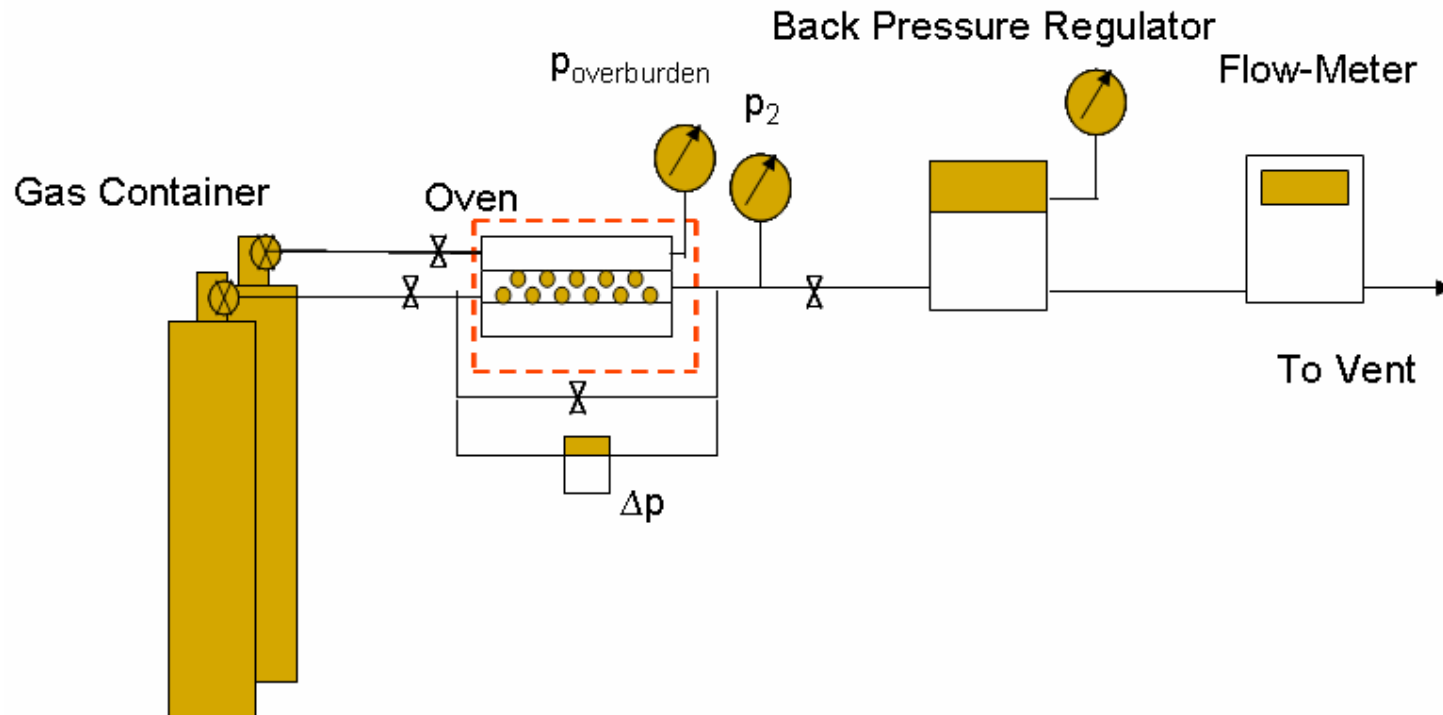


$\text{CH}_4 + \text{CO}_2 + \text{N}_2$



How do we probe mechanisms?

sorption/permeability/gas injection



- Pore pressure: 60~1100 psi
- Net effective stress: 300-400 psi
- Gas mixtures made in the lab by weight



Coal Coring and Processing

Cryogenic versus conventional



PRB coal samples are provided by Montana Bureau of Mines and Geology. The core recovery is less than 2%



Composite Coal



Total length: 21.17 cm

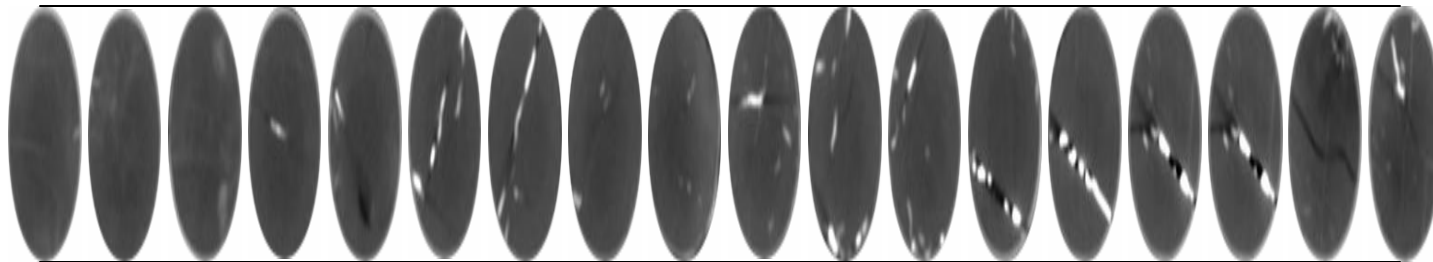
Diameter: 2.79 cm

Porosity: 7%

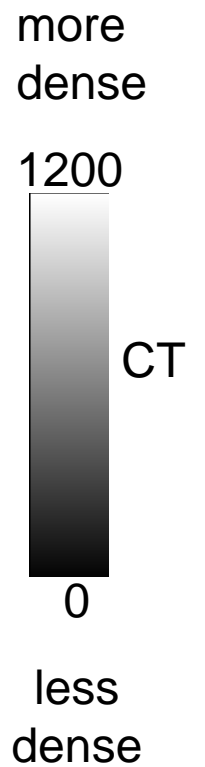
Permeability to He: 1.7 md



Intact composite coal plugs



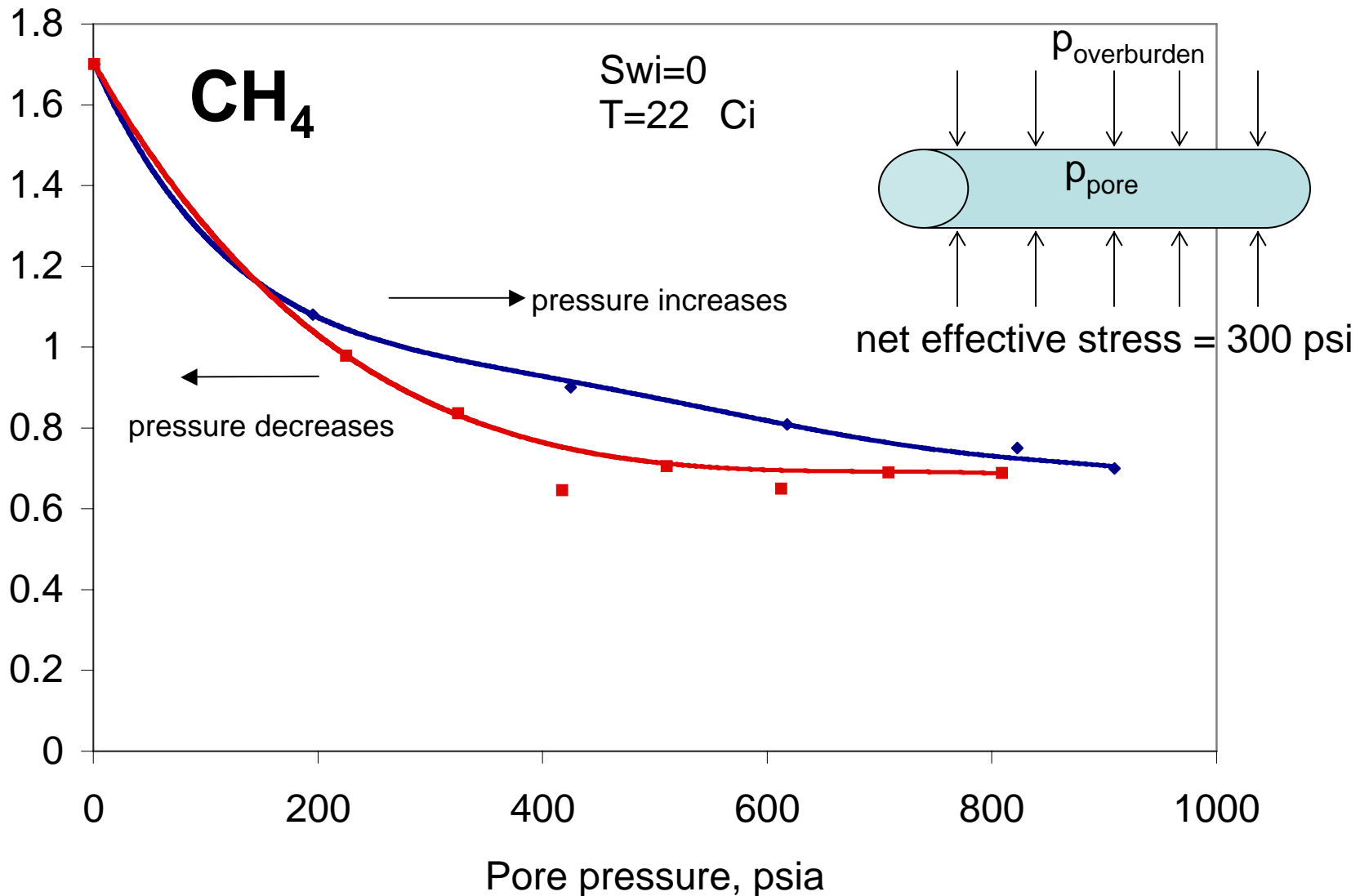
Corresponding CT-images





Is permeability sensitive to gas?

hysteresis loading/unloading

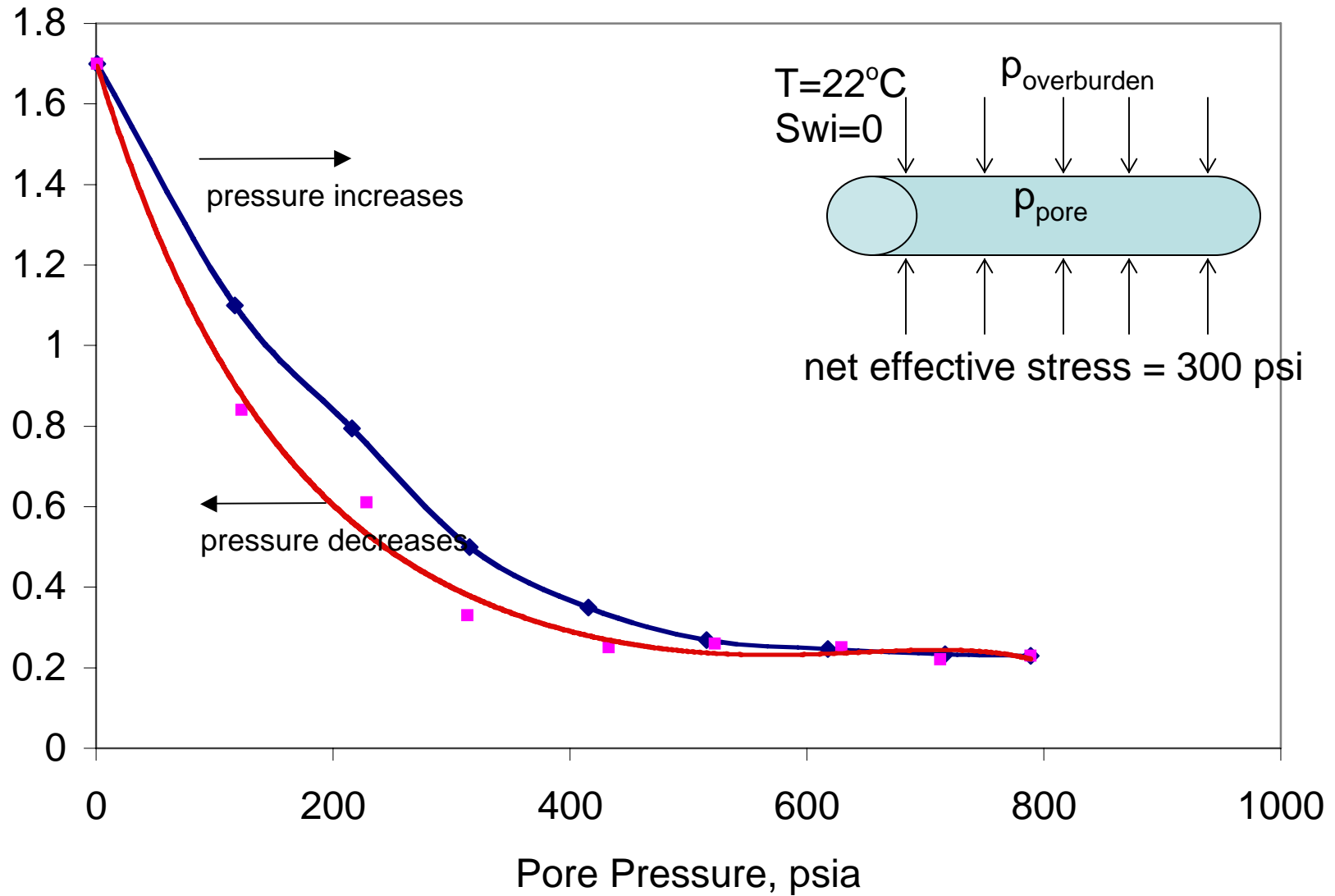




Permeability to CO₂



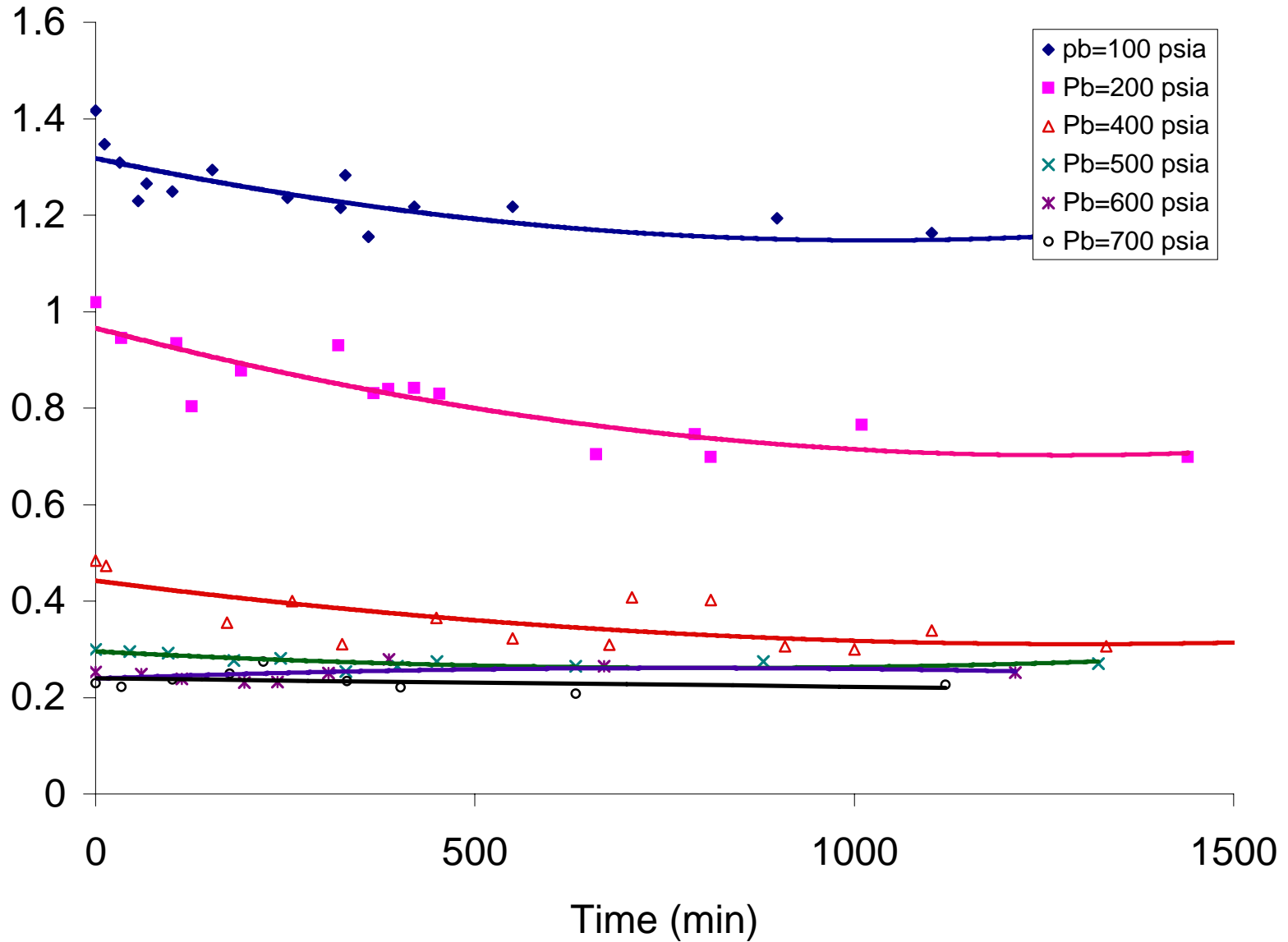
hysteresis loading/unloading





Time to Steady Flow

increasing pressure



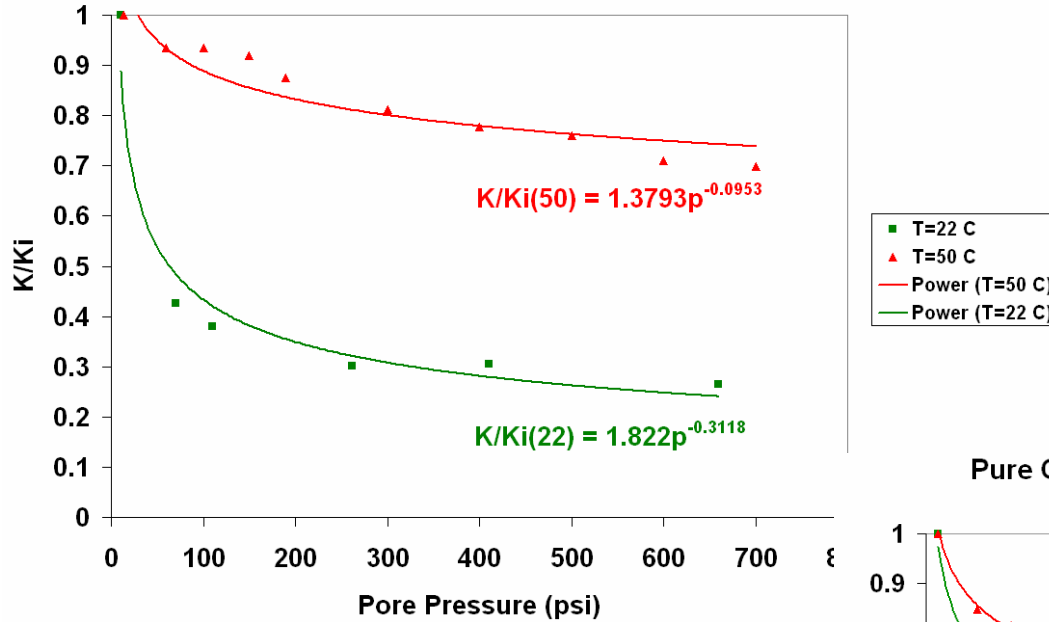


Effect of Temperature

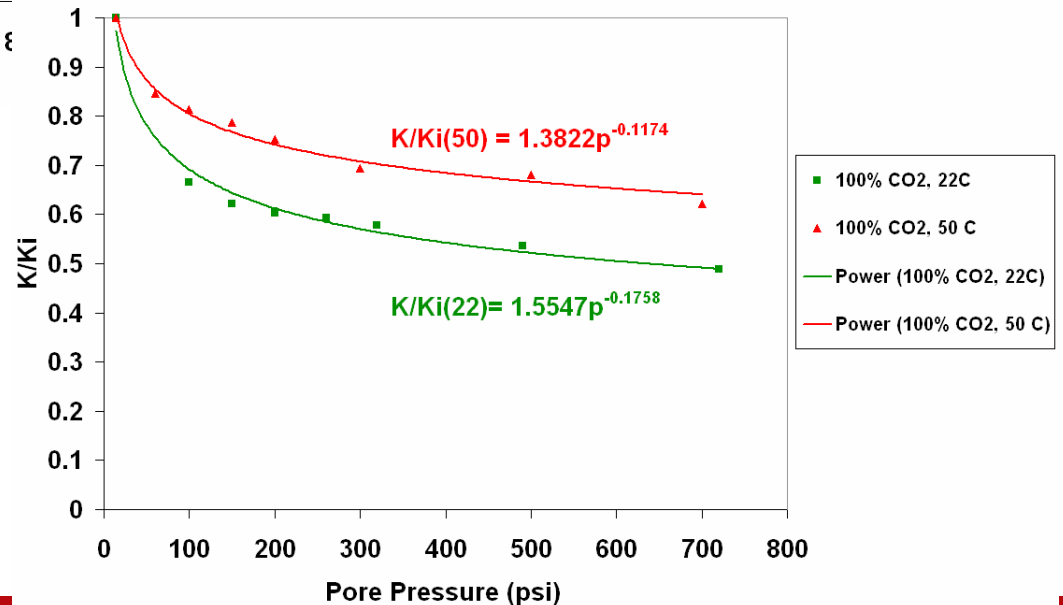
generally, adsorption decreases with T



Pure CH4, net overburden pressure 400 psi



Pure CO2, net overburden pressure=400psi



Ground Powder River Basin (WY) Coal





Today's Presentation

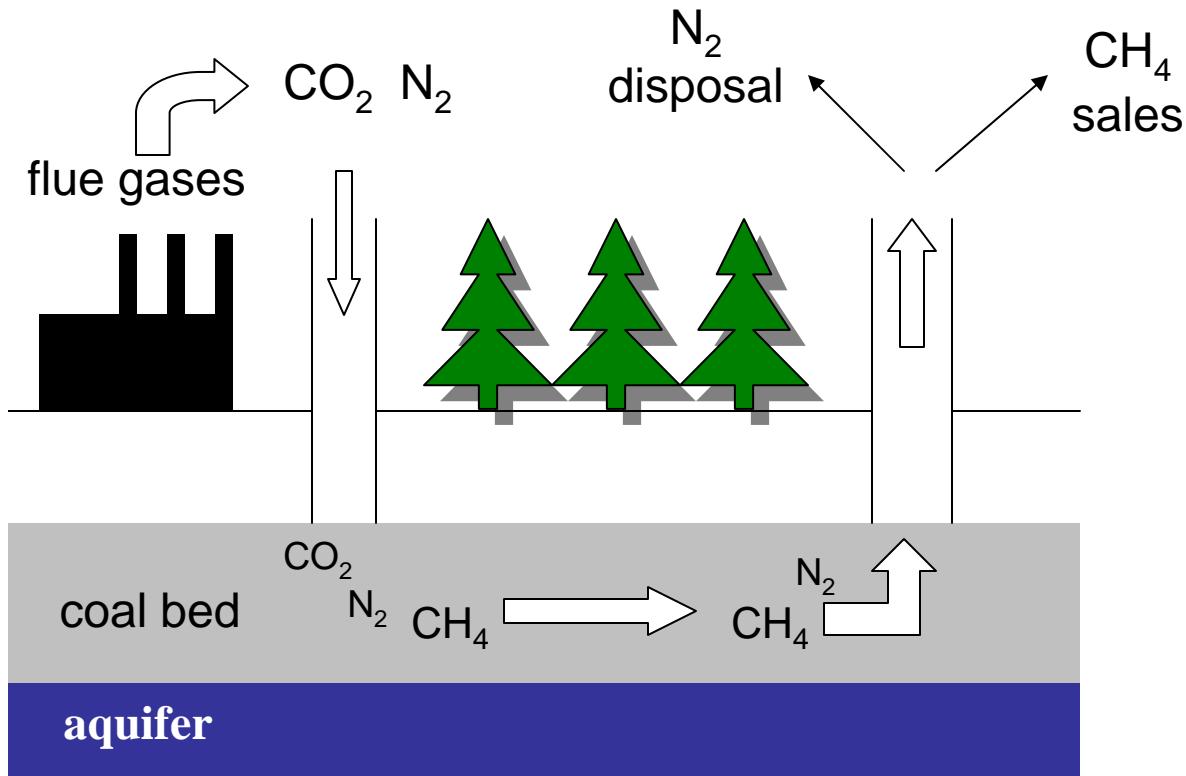


image source: C.J. Seto

- desorption through reduction of partial pressure
- 80-90% recovery of methane

- How much gas adsorbs on coal?
- How do we probe flow mechanisms?
- Is coal permeability sensitive to the type of gas?
- How does water flow?



How does flow through coal?



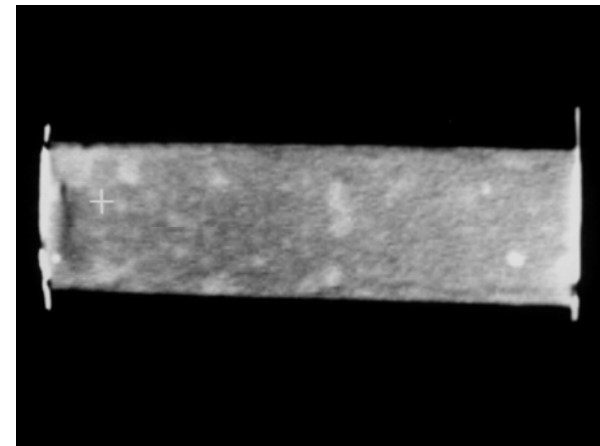
- X-ray computerized tomography (CT) scanner
 - X-ray radiation through the core captured by detectors
 - More dense materials/fluids yield greater attenuation



- Porosity, saturation calculations

- Porosity
$$\phi = \frac{CT_{wet} - CT_{dry}}{CT_{water} - CT_{air}} = \frac{V_{void}}{V_{rock}}$$

- Saturation
$$S_w = \frac{CT_{obj} - CT_{dry}}{CT_{wet} - CT_{dry}} = \frac{V_w}{\phi V_{rock}}$$



800

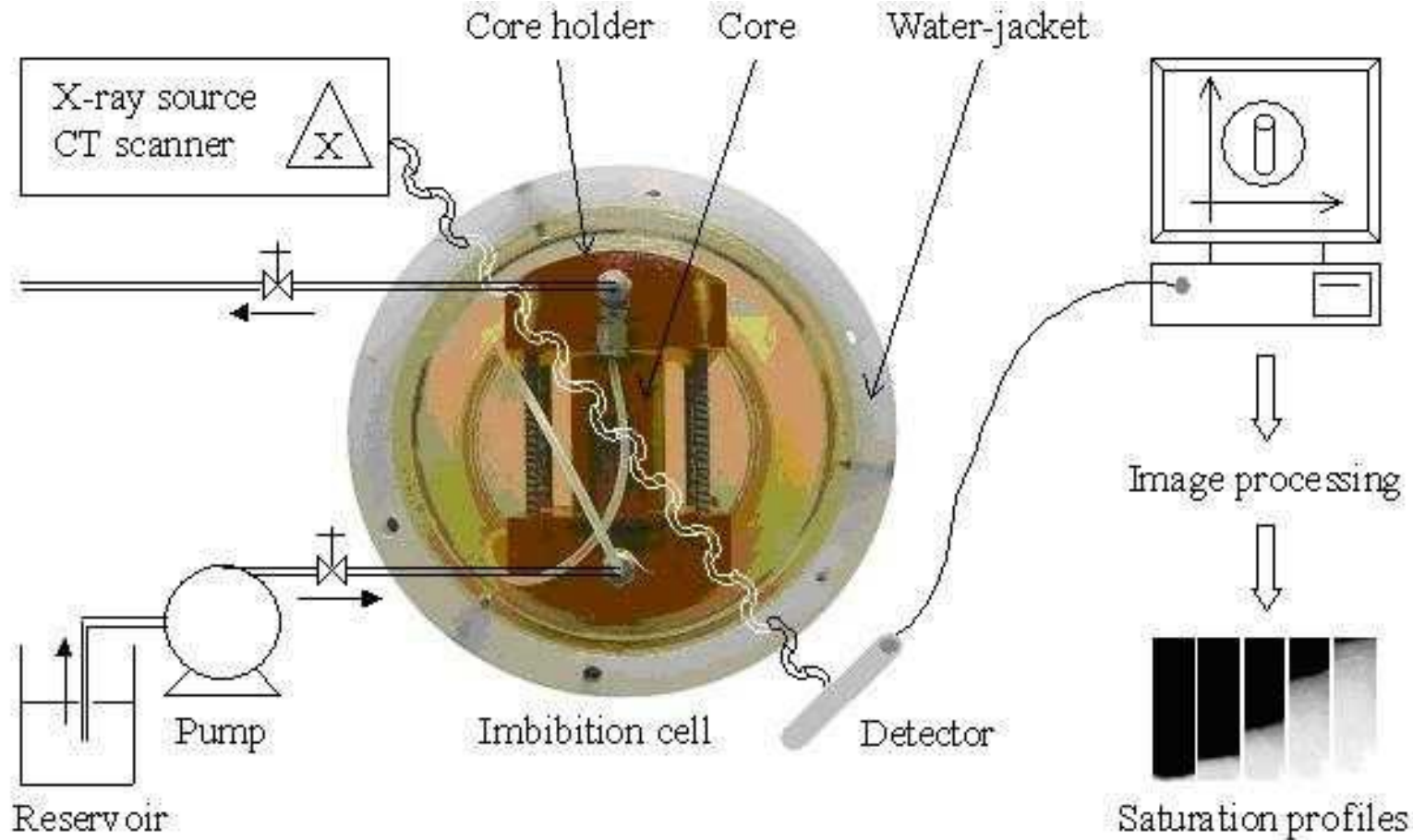
+800

+0

0

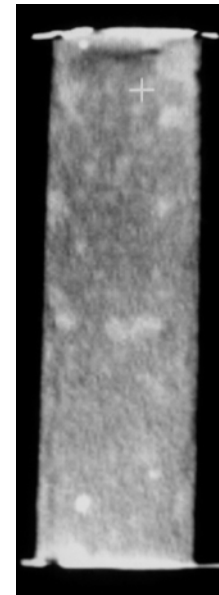
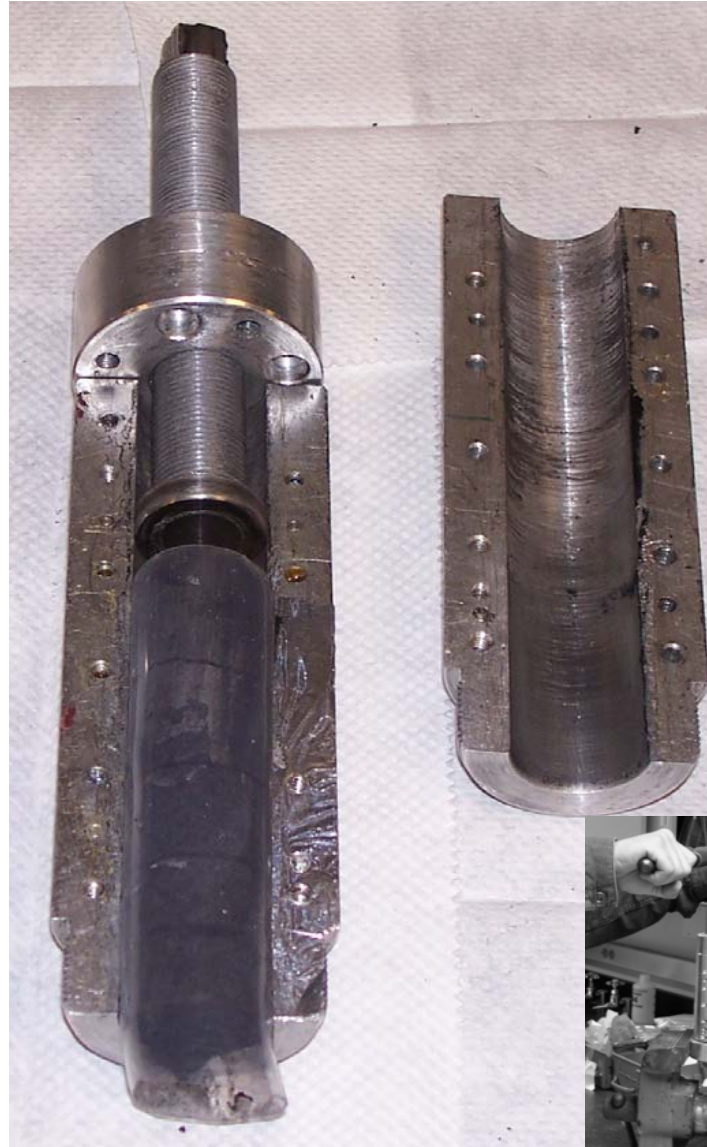


CT Scanning Process





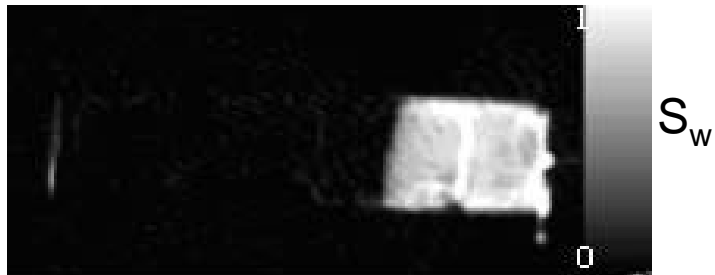
"Core" Preparation



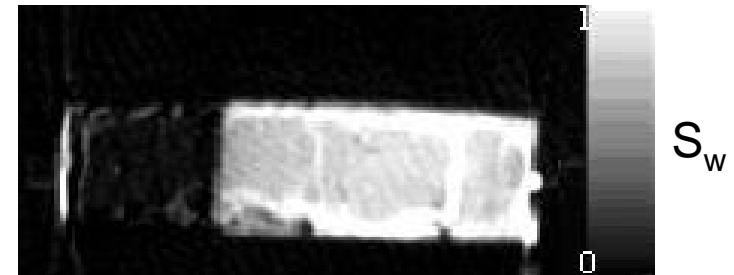


Spontaneous Water Imbibition

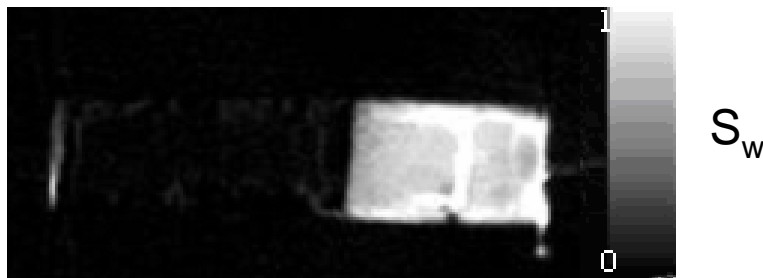
pH=2



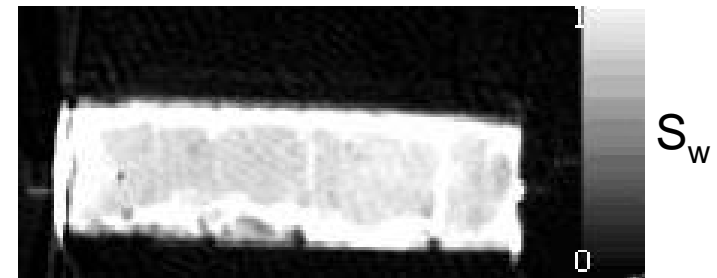
15 mins



180 mins



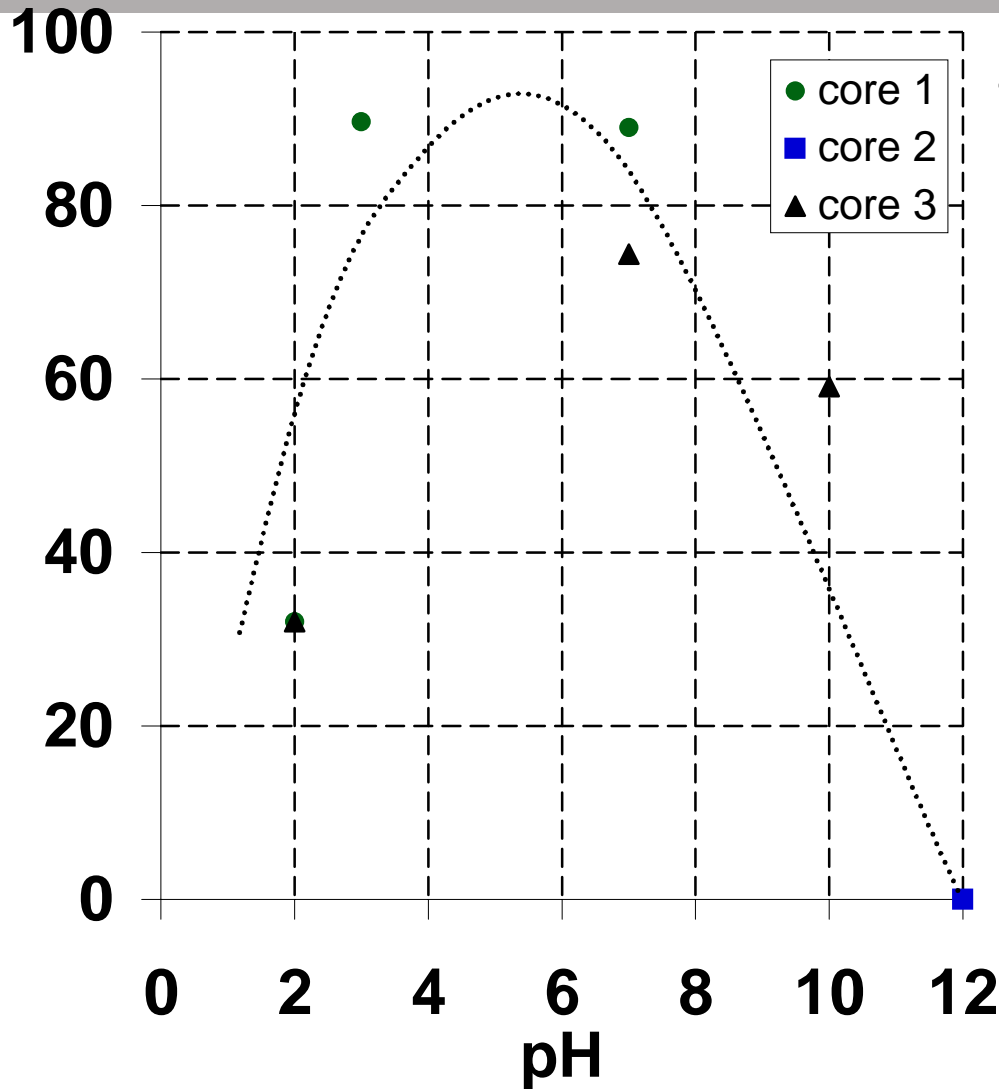
60 mins



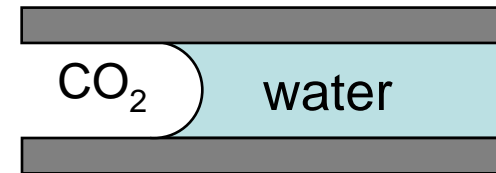
post forced imbibition



Wettability Aspects



- coal wettability varies with pH and CO₂ content





Relative Permeability is Foundational to Modeling



$$\frac{\partial G_i}{\partial \tau} + \frac{\partial H_i}{\partial \xi} = 0$$

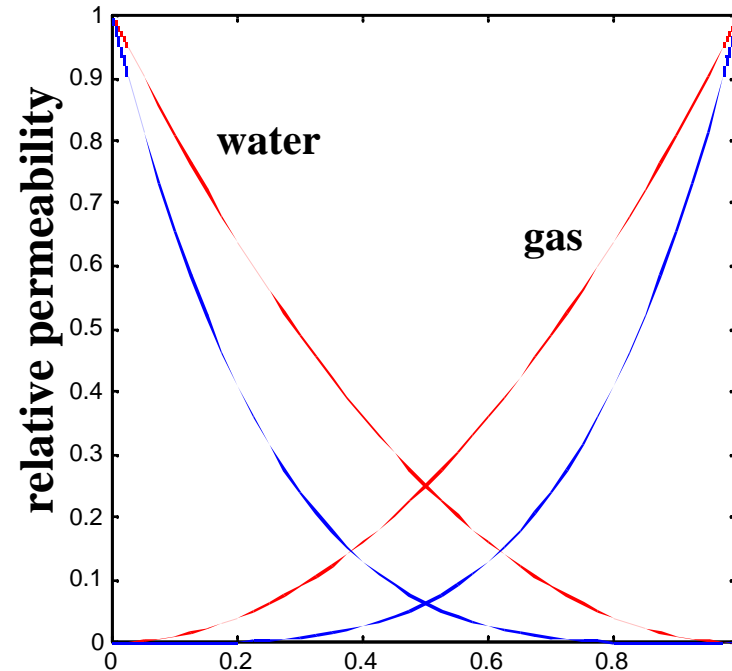
$$G_i = \rho_{DG} y_i S_g + \rho_{DL} x_i S_L + \frac{1-\phi}{\phi} a_{iD}$$

$$H_i = (\rho_{DG} y_i u_g + \rho_{DL} x_i u_L)$$

$$u_\beta = -\frac{k k_{r\beta}}{\mu_\beta} \nabla p_\beta$$

assumptions

- 1D flow
- local equilibrium
- no dispersion or diffusion
- homogenous porous medium
- negligible gravity or capillary forces





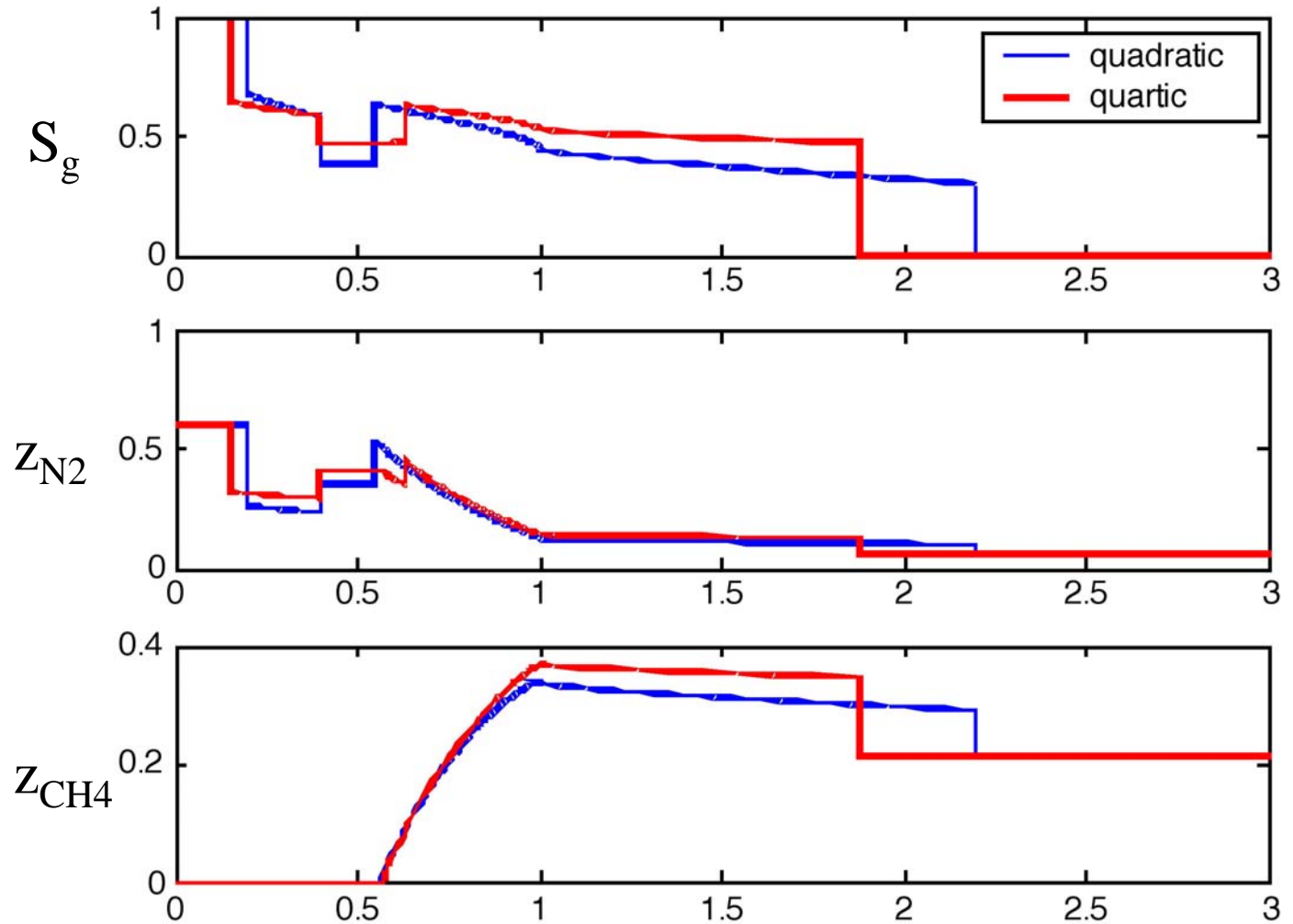
Sensitivity to Rel-Perm



40% CO₂ 60% N₂

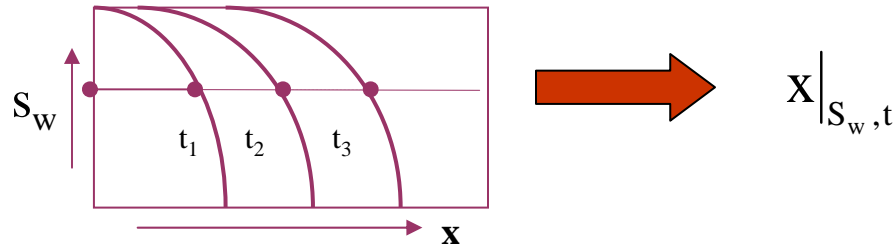


CH₄ + CO₂ + N₂



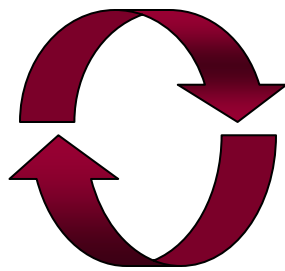


Relative Permeability from Imbibition Data



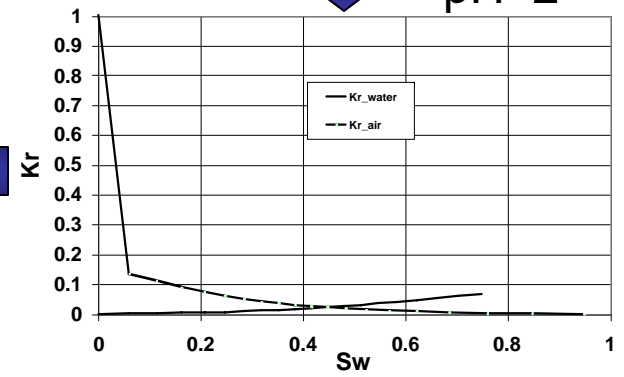
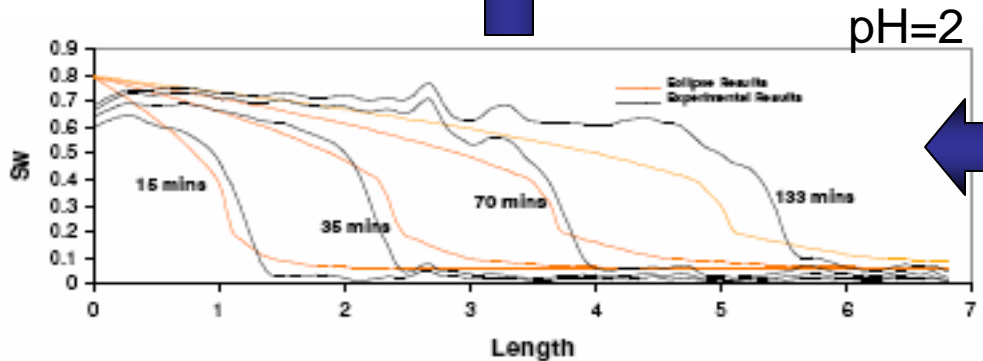
Simulated Annealing

Is difference minimized ?



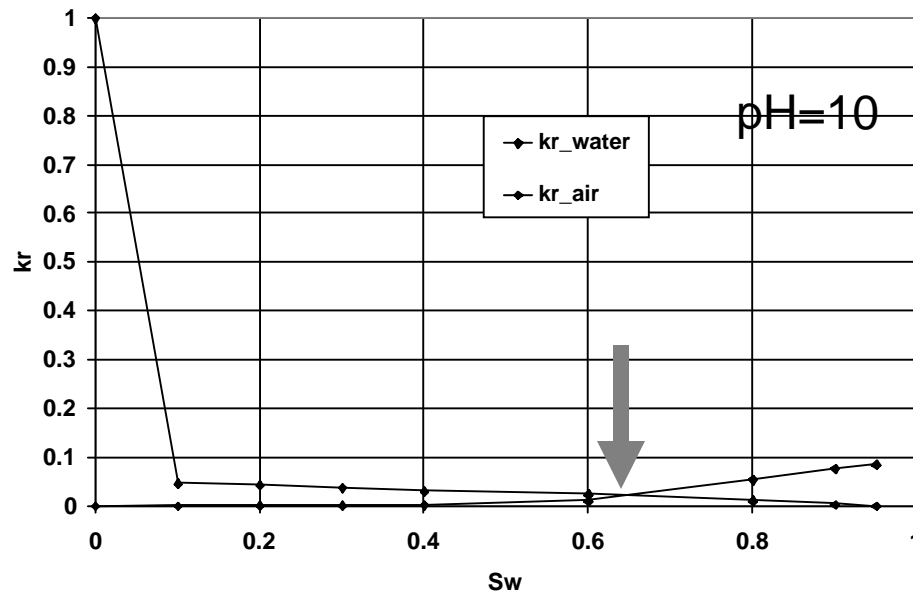
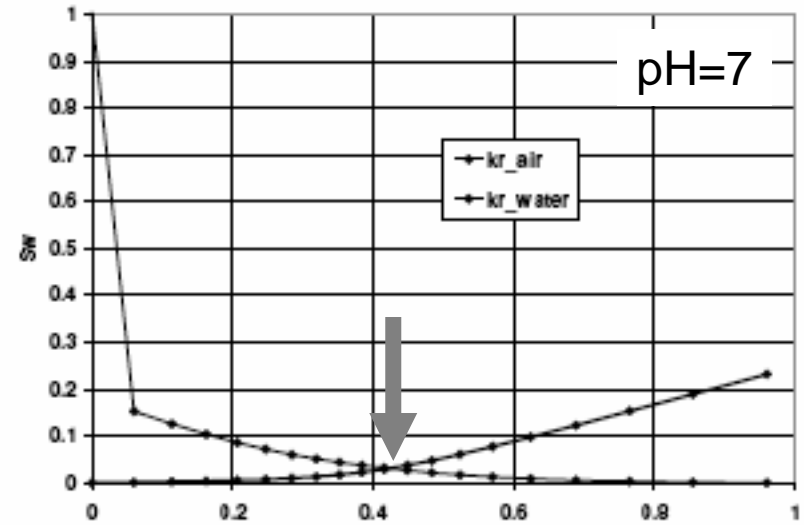
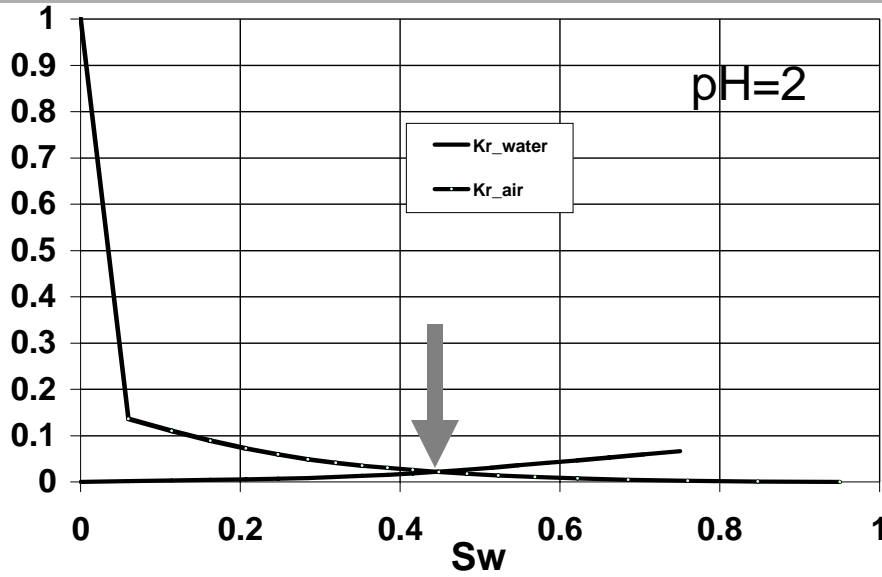
B-Splines Coeff

pH=2





Relative Permeability vs pH





Summary



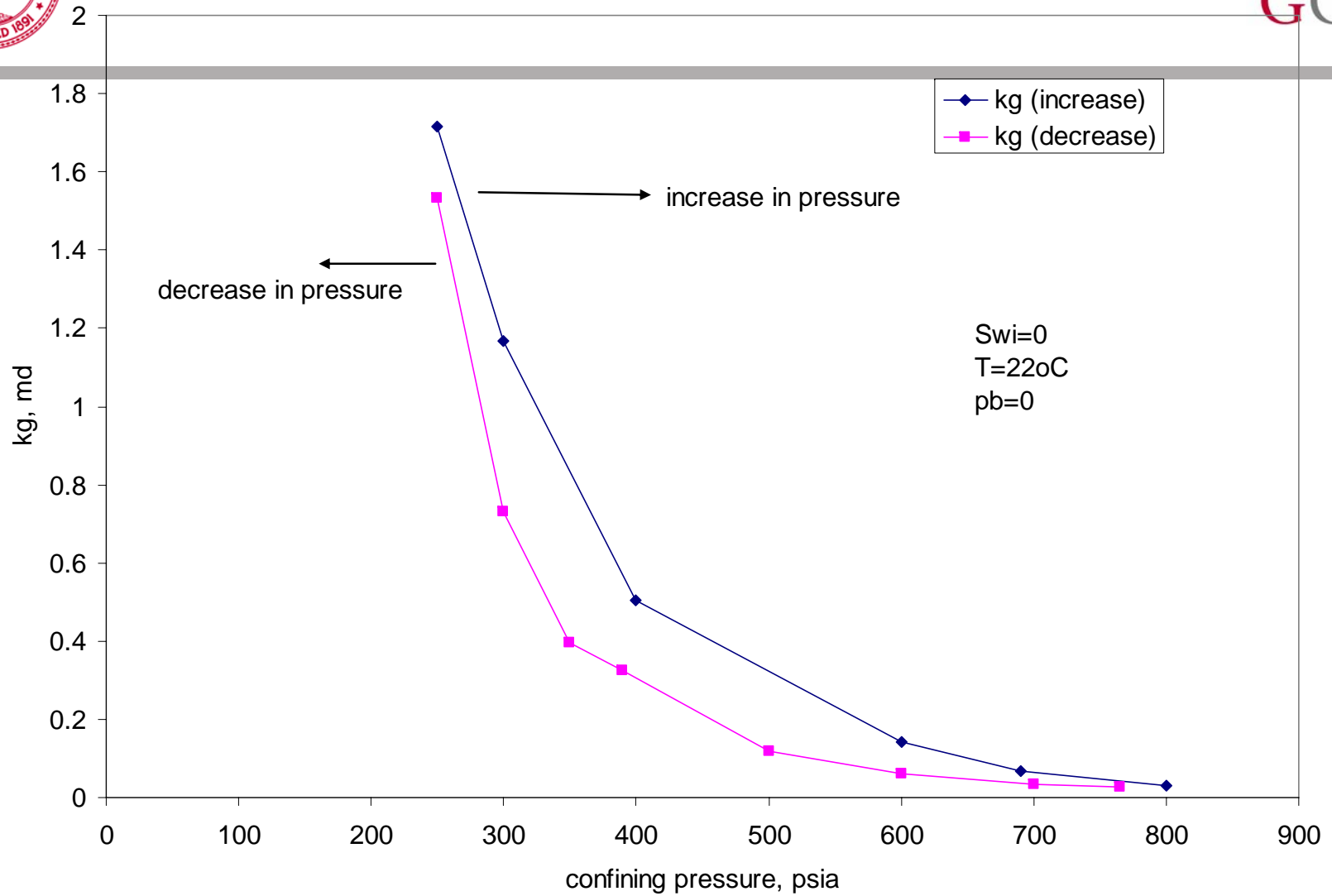
- Gas injection enhances CBM recovery significantly (at least 94% OGIP)
- Coals demonstrate rich dynamical behavior that is sensitive to
 - gas adsorption
 - adsorption hysteresis
 - temperature
 - water pH
- Permeability reduction during CO₂ flow is less severe in intact core samples
- Wettability of coal surfaces varies with the amount (pressure) of CO₂



Acknowledgement



- GCEP
- Collaborators: Tanmay Chaturvedi, Kristian Jessen, Wenjuan Lin, Lynn Orr, Carolyn Seto, Tom Tang.



Effect of confining pressure on permeability to Helium



Why are we studying coalbeds?

methane/CO₂ storage/acid gas disposal



•win / win / win ?

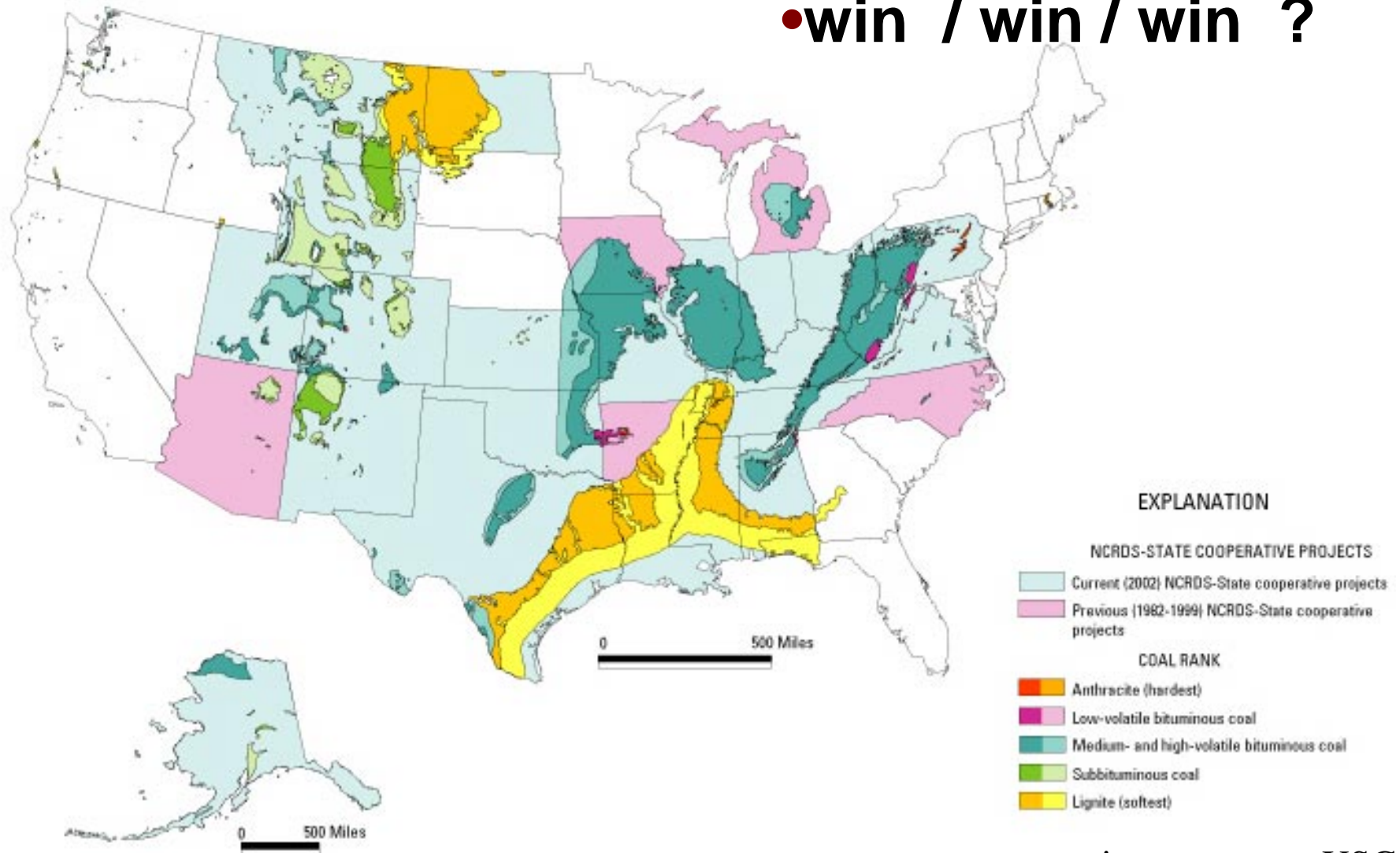


image source: USGS