

Analytical Solutions For Mixed Gas Injection in CO₂ Sequestration and Enhanced Coalbed Methane Recovery

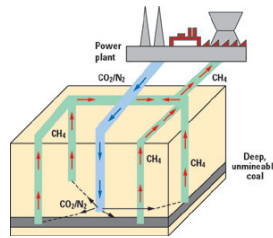


Carolyn Seto, Kristian Jessen and Franklin M. Orr, Jr.

Coal Reservoirs

Motivation

Injection of CO₂ into deep unminable coal seams is an option for geological sequestration of CO₂. In these systems, adsorption of gas on the internal surfaces of the coal is the storage mechanism. In many of these reservoirs, large amounts of CH₄ are adsorbed onto the coal. CO₂ is preferentially adsorbed compared to CH₄, which offers the possibility of desorption and increased recovery of CH₄ from coal without lowering pressure in the coal bed. The interaction between CO₂ sequestration and enhanced CH₄ production make coal reservoirs interesting candidates for sequestration.



Concept of integrated sequestration and electricity generation scheme (SPE 71749). Flue gas sequestered from gas burning power plant are used to enhance CH₄ recovery for power generation.

Understanding the complex interplay between adsorption, phase behaviour and convection is important in creating efficient and physically accurate techniques for predicting the fate of injected CO₂ in the subsurface.

Mixed Gas Injection

Flue gases from industrial power generation activities are significant point sources of anthropogenic CO₂. In some settings, it is expensive to obtain high purity CO₂ streams, and it is possible that mixtures of CO₂, N₂ and sulphur oxides may be injected. However, additional costs associated with compression of impurities may negate the expense of separation prior to injection. Understanding the effect of these other gas species in the injection gas stream on recovery efficiency and timing is important for engineering sequestration projects that maximise recovery and CO₂ storage while minimising costs.

Mathematical Model

For convection-dominated flows, 1D analytical solutions can be used to model multicomponent, multiphase flow in coal reservoirs.

$$\frac{\partial G_i}{\partial \tau} + \frac{\partial H_i}{\partial \xi} = 0 \quad z_i(\xi, 0) = \begin{cases} z_i^{inj} & \xi = 0 \\ z_i^{init} & \xi > 0 \end{cases}$$

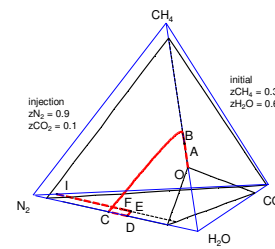
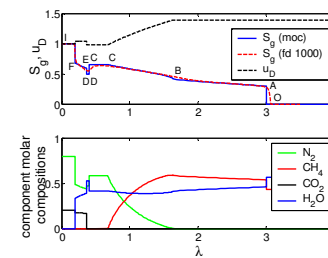
$$i = 1 \dots N_c \quad j = 1 \dots N_p \quad u(\xi = 0) = 1 \quad \tau > 0$$

$$G_i = \sum_{j=1}^{N_p} x_{i,j} \rho_j S_j + \frac{1-\phi}{\phi} a_i \quad H_i = u_D \sum_{j=1}^{N_p} x_{i,j} \rho_j f_j$$

For a multicomponent system, a system of coupled transport equations results, which can be decomposed into an eigenvalue problem. In such an analysis, the eigenvalues represent wave velocities at which the composition propagates. The corresponding eigenvectors represent the directions of variation in composition space. Solutions connecting initial and injection conditions can consist of segments of continuous variation, discontinuous variation and zones of constant state. Unique physical solutions are constructed by applying a velocity rule and entropy conditions.

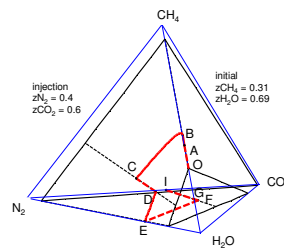
Solution structures for injection gas mixtures have shock and spreading features of pure gas injection solutions (Seto *et al.*, 2006). More strongly adsorbing gases displace less strongly adsorbing gases through a self sharpening displacement or shock (CO₂ displacing CH₄). Less strongly adsorbing gases displace more strongly adsorbing gases by means of a continuous variation or rarefaction (N₂ displacing CH₄).

N₂-Rich Injection Gas



Relative to CH₄, N₂ is a less strongly adsorbing gas while CO₂ is more strongly adsorbing. Preferential adsorption of CO₂ from the flowing phase occurs as the displacement propagates through the coal. The injection gas is separated into a slow moving bank of CO₂ and faster moving bank of N₂. CO₂ is isolated from CH₄ production. As the N₂ bank propagates through the coal, N₂ desorbs CH₄ by reducing the partial pressure of CH₄. This results in a mixture of CH₄ and N₂ at the outlet, requiring separation of N₂ from the production stream.

CO₂-Rich Injection Gas

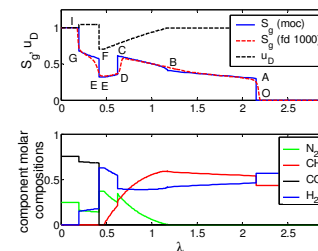


As N₂ propagates through the coal and desorbs CH₄, gas saturation rapidly increases to a maximum by means of a shock, making gas the preferential flowing phase.

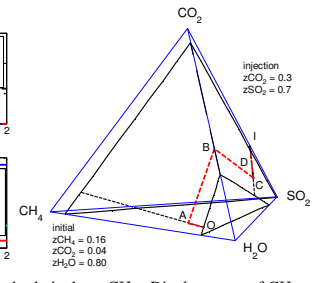
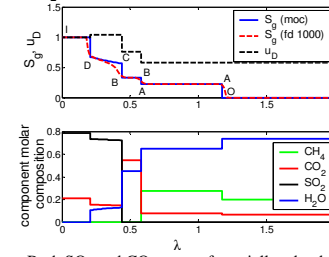
The volume increase from desorption of CH₄ is no longer sufficient to offset the volume decrease due to CO₂ adsorption, resulting in a slower displacement than the N₂-rich injection displacement.

For CO₂-rich injection gas mixtures, a large fraction of injection gas is removed from the flowing phase, forcing a new composition route connecting initial and injection states.

Once CO₂ is adsorbed onto the coal, gas phase saturations created by the remaining N₂ in the injection gas are sufficiently low that it is no longer the preferential flowing phase, resulting in a slow moving intermediate N₂-H₂O binary mixture.



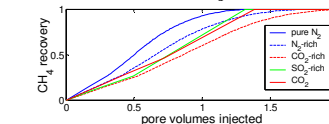
SO₂-Rich Injection Gas



Both SO₂ and CO₂ are preferentially adsorbed relatively to CH₄. Displacement of CH₄ occurs through a series of shocks. Injection gas components are separated into banks based on their relative adsorption strength. The produced CH₄ is free of SO₂.

A significant decrease in local flow velocity occurs due to removal of all injection gas components from the flowing phase as they adsorb onto the coal surface, resulting in the latest breakthrough time for the injection gas scenarios presented.

Effect of Mixed Gas Injection on Recovery



Efficient recovery of CH₄ is achieved for both pure and mixed injection gas compositions. Total recovery times for mixed gas injection are comparable to those for pure gas injection.

The volume increase to the flowing phase caused by CH₄ desorption results in faster initial recovery in N₂-rich displacements, while the volume decrease caused by adsorption of CO₂ and SO₂ in acid gas injection mixtures results in slower initial recovery.

Conclusions and Summary

- Preferential adsorption of gases on the coal surface and differential partitioning of components between gas and liquid phases results in chromatographic separation of injection gas mixtures as they propagate through the reservoir. More strongly adsorbing, less volatile components travel through the coal slower than less strongly adsorbing, more volatile components.
- Efficient recovery of CH₄ is achieved for both pure gas injection and flue gas mixtures. Faster recovery is achieved with N₂-rich gases, but requires separation of the produced gas stream and compression of large volumes of N₂. Later breakthrough times are achieved with CO₂-rich injection gas.
- Injection gas mixture can be optimised to align with the goal of the project, be it accelerating CH₄ recovery or delaying CO₂ breakthrough.

References

Seto, C. J., Jessen, K., Orr, F. M., Jr. "A Four-Component, Two-Phase Flow Model for CO₂ Storage and Enhanced Coalbed Methane Recovery." SPE 102376, presented at the 2006 SPE Annual Technical Conference and Exhibition, San Antonio, TX., 24-27 September 2006.